

1 An article submitted to Journal of Geophysical Research Oceans

2

3

4 Contribution of biological effects to carbonate-system variations and the
5 air–water CO₂ flux in inner and outer bays in Japan

6

7

8 Tatsuki Tokoro^{*,1,2}, Shin-ichiro Nakaoka¹, Shintaro Takao¹, Tomohiro Kuwae², Atsushi
9 Kubo³, Toru Endo⁴, Yukihiro Nojiri^{1,5}

10

11

12 ¹Center for Global Environmental Research, National Institute for Environmental
13 Studies, Tsukuba, Japan

14

15 ²Coastal and Estuarine Environment Research Group, Port and Airport Research
16 Institute, Yokosuka, Japan

17

18 ³Department of Geosciences, Shizuoka University, Shizuoka, Japan

19

20 ⁴Graduate School of Engineering, Osaka City University, Osaka, Japan

21

22 ⁵Graduate School of Science and Technology, Hirosaki University, Hirosaki, Japan

23

24 Key points:

- 25 ● Inner bays in Japan are reportedly annual atmospheric CO₂ sinks but
26 comprehensive measurements are few.
- 27 ● We quantified biological effects on carbonate parameters and CO₂ flux using an
28 empirical relationship with riverine and oceanic endmembers.
- 29 ● The biological effect was regulated mainly by nutrient concentrations related to
30 wastewater treatment.

31

32 *Corresponding author: Tatsuki Tokoro (tokoro.tatsuki@nies.go.jp)

33

34

35

36 **Abstract**

37

38 We evaluated the contribution of biological effects (photosynthesis, respiration, and
39 decomposition) to the carbonate parameters and air–water CO₂ fluxes in Tokyo Bay, Ise
40 Bay and Osaka Bay in Japan. The carbonate parameters were measured mainly by cargo
41 ships travelling between Japan and other countries. We used the measurement data from
42 three inner bays and surrounding outer bays in Japan along with reference data from
43 previous studies for complementary analysis. We found that 1) the inner bays in this
44 study were strong annual atmospheric CO₂ sinks, 2) the annual biological effect on the
45 air–water CO₂ fluxes was about 5–25% of the measured CO₂ fluxes and it affected the
46 seasonal variation of the CO₂ flux, and 3) the biological effect was largest in Tokyo Bay,
47 and almost the same in Ise and Osaka Bays. The intensity of the biological effect
48 corresponded mainly with nutrient concentrations, which seemed to be controlled by the
49 wastewater treatment in urbanized areas around the bays. The CO₂ flux was also
50 affected by the seawater residence time, salinity, and stratification. Our results suggest
51 that labile carbon/nutrient ratio of wastewater should be a major consideration for
52 evaluating the biological effect on the carbon cycle in urbanized inner bays, which will
53 continue to expand globally.

54

55 **Plain Language Summary**

56

57 We analyzed the biological effect (photosynthesis, respiration, and decomposition) on
58 air–water CO₂ exchange in Tokyo Bay, Ise Bay and Osaka Bay in Japan using data from
59 cargo-ship measurements and previously published reports. We concluded that 1) bay
60 water strongly absorbs atmospheric CO₂, 2) biological effects accounted for 5–25% of
61 the evaluated CO₂ absorption and had significant effects on its seasonal variation, and 3)
62 the biological effects seemed to be mediated mainly by the carbon/nutrient ratio in
63 wastewater. This study should improve our understanding of the carbon flow in
64 urbanized coastal areas, which are expanding globally.

65

66 1. Introduction

67

68 The ocean is one of the largest carbon reservoirs on earth, and the
69 quantification of the exchange of atmospheric carbon dioxide (CO₂) with the ocean is
70 necessary for predicting future climate change. The air–water CO₂ flux in the major
71 oceans has been studied since the late 1970s and the regional and seasonal variations
72 have been estimated (e.g., Takahashi et al., 2009; Wanninkhof et al., 2019). Meanwhile,
73 the quantification of the flux in coastal areas is still challenging because of the large
74 temporal and spatial variations. Recent studies have shown that near-shore areas are
75 sources of atmospheric CO₂ on average because of the input and of organic carbon and
76 the mineralization (Borges et al., 2005; Cai, 2011; Chen and Borges, 2009; Chen et al.,
77 2013), whereas some other studies showed a local annual CO₂ sink in areas with
78 submerged autotrophic ecosystems (Kayanne et al., 1995; Tokoro et al., 2014). Marginal
79 seas (continental shelves) have been reported as atmospheric CO₂ sinks but there is still
80 uncertainty surrounding the actual estimates has been debated (Borges et al., 2005; Cai,
81 2011; Chen and Borges, 2009; Chen et al., 2013; Laruelle et al., 2014).

82 The role of inner and outer bays in atmospheric CO₂ exchange has not been
83 analyzed comprehensively despite their importance in anthropogenic activities (Chen
84 and Borges, 2009; Chen et al., 2013). Although the near-shore area is generally
85 considered to be a CO₂ source region, some studies have reported that urbanized inner
86 bays in Japan are annual atmospheric CO₂ sinks (Endo et al., 2017; Fujii et al., 2013;
87 Kubo et al., 2017). In these bays, CO₂ undersaturation might result from wastewater
88 treatment (Kubo et al., 2017; Kuwae et al., 2016). The treatment process removes
89 labile carbon, yielding water with relatively less carbon than nutrients, which promotes
90 primary production in the bay water. In addition, the organic matter remaining in the
91 treated water is refractory (Kubo et al., 2015) and thus further mineralization and
92 increase in the CO₂ concentration in the bay water is suppressed. However, the effects of
93 biological processes such as photosynthesis, respiration, and decomposition on the
94 carbonate parameters and the air–water CO₂ flux have not been precisely quantified.

95 Here, we evaluated the biological effects of inner-bay water on temporal and
96 spatial changes in the carbonate parameters and the air–water CO₂ flux, including the
97 surrounding outer bays, in Japan. We discuss the biological effects on the flux in terms
98 of the regulating factors and the extendibility of our observations to other areas and for
99 global estimation.

100

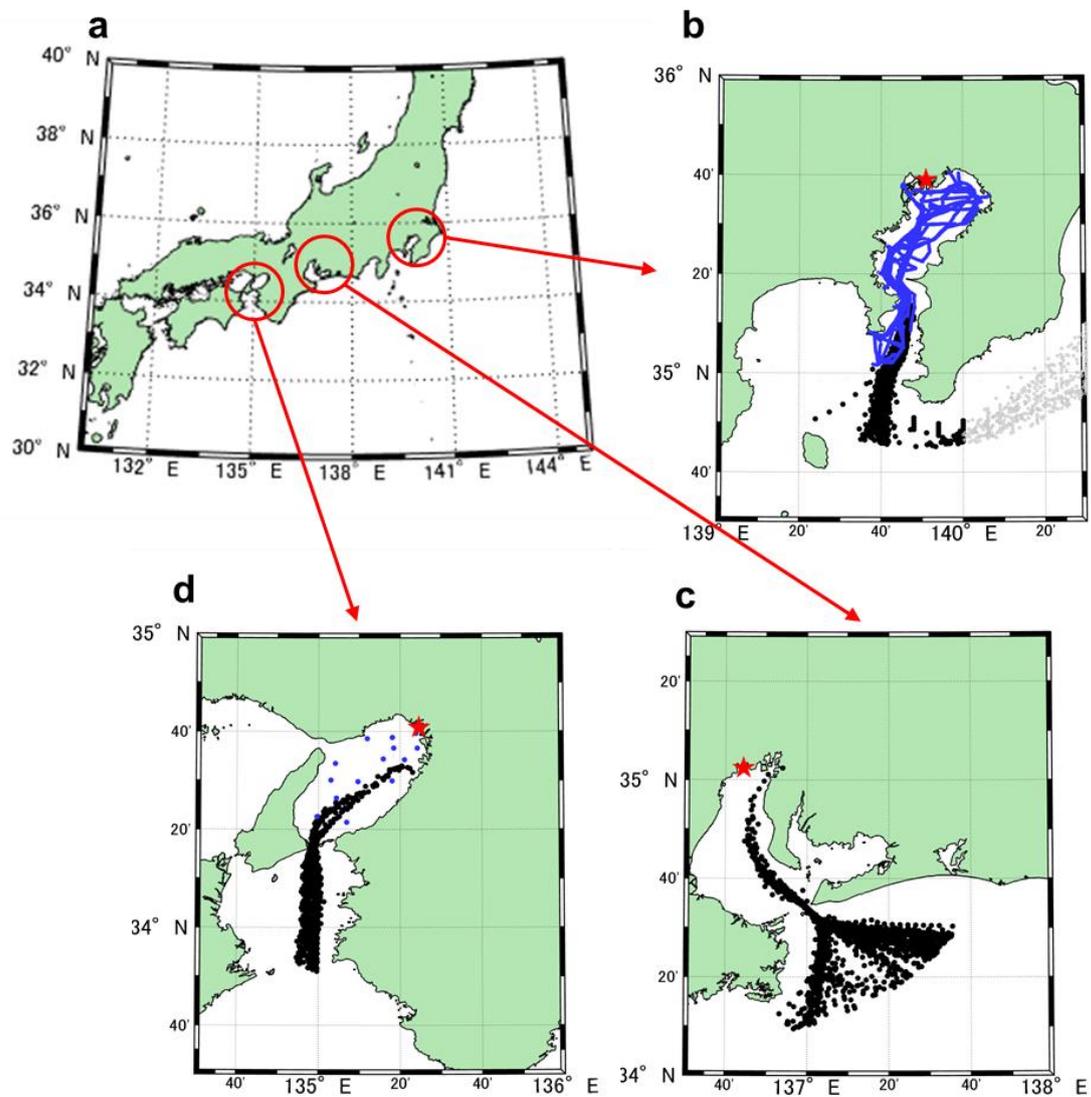
101 2. Methodology

102

103 2.1 Study site

104

105 This study took place in Japan in Tokyo Bay, Ise Bay, and Osaka Bay, and
106 surrounding areas (Figure 1). These bays have similar topographic conditions such as a
107 southward entrance to the bay open to the Pacific Ocean and a surrounding, highly
108 urbanized coastal area. The populations of the urban areas surrounding Tokyo, Ise and
109 Osaka Bays were 37, 9 and 19 million, respectively, in 2014 (Global Metro Monitor;
110 <https://www.brookings.edu/research/global-metro-monitor/>). In this study, the term “bay”
111 refers to the inner and outer areas of each bay, which were analyzed together for
112 expedience.



113

114 **Figure 1.** (a) Map of Japan and the general locations of the bays and surrounding seas included in
115 this study. (b) Tokyo Bay, (c) Ise Bay, (d) Osaka Bay. The black filled circles indicate the locations
116 of the National Institute for Environmental Studies (NIES) data used in this study. The grey circles in
117 (b) are filtered NIES data from east of 140°E, which were defined as data external to Tokyo Bay. The
118 blue circles indicate additional data collected by Tokyo University of Marine Science and
119 Technology (TUMSAT) in Tokyo Bay and by Osaka City University (OCU) in Osaka Bay. The red
120 stars mark the river mouths of the main river located in the inner part of each bay and used as the
121 zero point for the distance parameter *dist* (Tokyo Bay, 35.65°N, 139.85°E; Ise Bay, 35.04°N,
122 136.74°E; Osaka Bay, 34.68°N, 135.41°E).

123

124 2.2 Data processing

125

126 The data used for this study are measurements of water temperature, salinity
127 and fugacity of CO₂ (fCO₂) in the main urbanized inner bays and in the surrounding
128 outer bays, as observed by the National Institute for Environmental Studies (NIES),
129 Tokyo University of Marine Science and Technology (TUMSAT) and Osaka City
130 University (OCU). The other carbonate parameters—total alkalinity (TA) and
131 dissolved inorganic carbon (DIC)—were estimated using an empirical relationship
132 between TA and salinity, and the equilibrium calculation. The biological effect on
133 carbonate parameters and the air–water CO₂ flux was calculated from the difference
134 between the above estimated DIC and the value from the conservative mixing line
135 between the oceanic and riverine endmembers. We quantified the air–water CO₂ flux
136 and the magnitude of the biological effect to evaluate the contribution of the biological
137 effect in the bays to the exchange of atmospheric CO₂.

138 The water temperature, salinity, and fCO₂ in water and air were obtained from
139 the NIES database (<https://soop.jp>). The data are also available from the Surface Ocean
140 CO₂ Atlas (SOCAT; <http://www.socat.info>), which has been a public database since
141 2011 and represents an international collaboration among research institutes. The NIES
142 observations implemented as the Voluntary Observing Ship (VOS) programs by cargo
143 ships were originally for understanding the global carbon cycle but their data also
144 include inner bay measurements from 2005 to 2016 in Tokyo Bay and Ise Bay and
145 from 2011 to 2016 in Osaka Bay, where the cargo ships taking measurements have
146 anchor stations. We first extracted the data for 30–40°N and 130–145°E observed by
147 the cargo ships as the original coastal data. The details on fCO₂ measurements are
148 reported by Nakaoka et al. (2013).

149 For complementary analysis, we included some previous data as more

150 landward information than the NIES data. For Tokyo Bay, we added the data collected
 151 by TUMSAT from 2007 to 2010 (Kubo et al., 2017) to the NIES data after converting
 152 pCO₂ to fCO₂ using the empirical relationship incorporating temperature (Körtzinger,
 153 1999). Likewise, for complementary analysis, we added more landward measurement
 154 data from Osaka Bay collected by OCU in spring and autumn of 2014 (Endo et al.,
 155 2017). The additional raw data for Osaka Bay included the water temperature, salinity,
 156 and DIC. We therefore estimated fCO₂ using the equilibrium calculation (Zeebe and
 157 Wolf-Gladrow, 2001; we used their “recommended” coefficients for the calculation)
 158 and the TA from the empirical relationship with salinity (Taguchi et al., 2009).

159 The data were filtered by the distance from the inner part of each bay. The
 160 distance parameter *dist* (km) was calculated as follows:

161

$$162 \quad dist = 6370 \times \sqrt{(lat - dist0_lat)^2 + (lon - dist0_lon)^2 \times \cos\left[\left(\frac{lat + dist0_lat}{2}\right)^2\right]} \quad (1)$$

163

164 where *lat* and *lon* are the latitude and longitude of the measurement point in radians,
 165 respectively. *dist0_lat* and *dist0_lon* are the latitude and longitude of the point
 166 representing the mouth of the river with the highest flow among the rivers in the inner
 167 part of each bay (Ara River in Tokyo Bay, Kiso River in Ise Bay, and Yodo River in
 168 Osaka Bay; Figure 1). The data for which *dist* > 100 km were excluded from analysis as
 169 being out-of-range of the inner and outer bays, as determined by changepoint analysis
 170 using the salinity and DIC (see Text S1). In addition, the data from Tokyo Bay with
 171 longitude >140°E were excluded because even though these data were within the range
 172 of *dist* ≤ 100 km, they were from locations on the opposite side of a peninsula from the
 173 inner bay (Figure 1). In total, we analyzed 18,118 data points from Tokyo Bay (16,924
 174 from TUMSAT), 1926 from Ise Bay and 1067 from Osaka Bay (28 from OCU).

175 Because the NIES data for the inner bays were distributed uni-dimensionally
 176 along the course of the cargo ships, the spatial information for the data in this study
 177 was standardized by the distance parameter *dist* mentioned above. For evaluating
 178 seasonality, we calculated the parameter *monthlydata* for each data point using the
 179 temporal information as follows:

180

$$181 \quad monthlydata = month + \frac{(day-1)}{365.25} \times 12 \quad (2)$$

182

183 where *month* and *day* are the month (1–12) and day (1–28 or 30 or 31) of the

184 measurement, respectively. Additionally, given the temporal and spatial heterogeneity
185 of the measurement data, we used natural neighbor interpolation (Sibson, 1981) to
186 interpolate a grid with values at intervals of 1 km (*dist*) and 0.1 (*monthlydata*). In order
187 for the interpolation to reflect the seasonal cycle from the minimum and maximum
188 values for *monthlydata* (on 1 January and 31 December, respectively), the interpolation
189 was performed with the data from the latter half of the year (July to December) added
190 before the start of original data and that from the first half (January to June) appended
191 to the end.

192 In order to minimize the effect of the trend in fCO₂ due to anthropogenic CO₂
193 input to the ocean, we corrected for the increase rate in each bay and in the additional
194 data, except for the OCU data because those measurements were only taken for a single
195 year. The linear increase rate for each bay was estimated from the annual average fCO₂.
196 The increase in fCO₂ in water was corrected to that in 2010 if the increase was
197 significant. The linear rates of fCO₂ increase in water were 5.83 and 3.24 μatm yr⁻¹ in
198 Tokyo and Ise Bay, respectively, and were significant ($p < 0.001$), whereas that in Osaka
199 Bay was not significant ($p = 0.64$). Also, the additional TUMSAT data for Tokyo Bay
200 showed no significant increase ($p = 0.85$). Therefore, the rate of increase was corrected
201 to the base year of 2010 only for NIES data for Tokyo Bay and Ise Bay. For fCO₂ in air,
202 we assumed an increase of 1.5 μatm yr⁻¹ and corrected to this from the base year value.

203

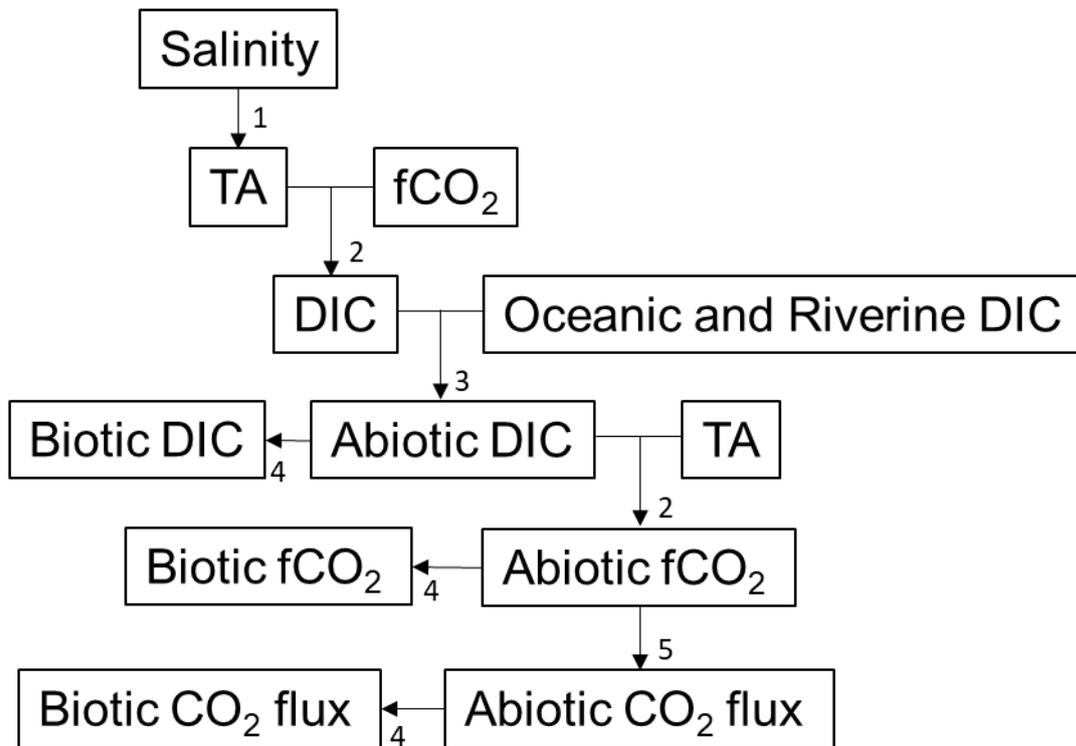
204

205 2.3 Calculation of the biological effect

206

207 Because fCO₂ in water is affected by temperature and salinity, the biological
208 effect cannot be quantified using only carbonate-system parameters. Although some
209 previous studies normalized fCO₂ using an empirical relationship between temperature
210 and fCO₂ (e.g., Takahashi et al., 2009), such a technique cannot be applied to coastal
211 carbonate analysis because the salinity varies widely and affects TA and DIC
212 non-linearly. Instead, we evaluated the biological effect using DIC, which is a
213 conservative parameter, and direct information about the biological effect from
214 processes such as photosynthesis, respiration, and decomposition (Figure 2). The DIC
215 was calculated using the carbonate equilibrium calculation (Zeebe and Wolf-Gladrow,
216 2001) with TA estimated from the empirical relationship with salinity in each bay from
217 a previous study (Taguchi et al., 2009).

Calculation protocol



218

219

220 **Figure 2.** Procedure for calculating abiotic and biotic CO₂ flux. 1: Calculation of total alkalinity
 221 (TA) using the empirical relationship of Taguchi et al. (2009). 2: Carbonate equilibrium calculation
 222 (Zeebe & Wolf-Gladrow, 2001). 3: Calculation of abiotic dissolved inorganic carbon (DIC) using the
 223 interpolation of the oceanic and riverine endmembers. 4: Calculation of biotic parameters as the
 224 difference between the net parameter and the abiotic parameter. 5: Calculation using equation (4).
 225 fCO₂, CO₂ fugacity.

226

227 For quantifying the biological effect, we defined DIC_b as the increase or
 228 decrease in DIC through biological activities. The DIC_b was calculated as the difference
 229 between DIC estimated from the equilibrium calculation and DIC_{ab}, which was the
 230 interpolated value between the oceanic and riverine endmembers in the coordinate
 231 system of salinity and DIC (Tokoro et al., 2014). DIC_{ab} represents the DIC
 232 concentration resulting from the mixing of the endmembers without any additional
 233 change in the bays such as through biological activity.

234 The endmembers and the biological effect were determined as follows. 1) the
 235 salinity and DIC of the oceanic endmember were estimated as the value outside of the

236 respective outer bays. In this study, those values were estimated as the monthly average
 237 of the salinity and DIC between *dist* of 90 km and 100 km. 2) We assumed that
 238 biological activity reached a minimum ($DIC_b \approx 0$) in the winter period (here, the three
 239 months with the lowest average water temperatures), according to the positive
 240 correlation between production by seagrass and algae and water temperature (Davison,
 241 1991; Lee et al., 2007; Tait & Schiel, 2013). Then the least-squares line of the salinity
 242 and DIC through the oceanic endmember average value during the above winter period
 243 can be used to approximate the DIC_{ab} in the winter period, and the riverine endmember
 244 (DIC_r , $\mu\text{mol kg}^{-1}$). That is, the former parameter is the approximation using the
 245 regression line of DIC vs salinity, and the latter parameter is the regression line value
 246 when the salinity is zero. These parameters can be estimated as follows:
 247

$$248 \quad DIC_r = \frac{\sum[(DIC_w - DIC_{ow})(Sal_w - Sal_{ow})]}{\sum[(Sal_w - Sal_{ow})^2]} \times (Sal_r - Sal_{ow}) + DIC_{ow} \quad (3)$$

249 Here Sal_w and DIC_w are the salinity and DIC in each bay during the three months with
 250 the lowest average water temperature. Sal_{ow} and DIC_{ow} are the mean values of the
 251 salinity and DIC, respectively, of the oceanic endmember during these three months.
 252 Sal_r is the salinity of the riverine endmember and assumed to be zero in this study. 3)
 253 DIC_{ab} in each month was calculated again as the linearly interpolated value between the
 254 endmembers. To evaluate the uncertainty of the estimated riverine DIC, we defined the
 255 range of DIC as $\pm 200 \mu\text{mol kg}^{-1}$ (see Text S2) and calculated the precision of the range
 256 for the following procedures. 4) DIC_b was calculated as the difference between the
 257 estimated DIC and DIC_{ab} . 5) fCO_2 was calculated from the equilibrium calculation
 258 using DIC_{ab} as the fCO_2 without any biological effects (fCO_{2ab}) and fCO_{2b} (the
 259 difference between fCO_2 and fCO_{2ab}) as the fCO_2 change due to the biological effects in
 260 each bay.
 261

262 The air–water CO_2 flux (F , $\mu\text{mol m}^{-2} \text{s}^{-1}$ or $\text{mol m}^{-2} \text{yr}^{-1}$) was calculated as
 263 follows:

$$264 \quad F = k \cdot S(fCO_{2water} - fCO_{2air}) \quad (4)$$

265 where k is the gas transfer velocity (m s^{-1}) and calculated as described in the next
 266 paragraph. S is the solubility of CO_2 in water ($\text{mol m}^{-3} \text{atm}^{-1}$) and estimated using an
 267 empirical equation using water temperature and salinity (Weiss, 1974). fCO_{2water} and
 268 fCO_{2air} are the fugacity of CO_2 in water and air (μatm), respectively. A positive value for
 269
 270

271 the flux indicates a CO₂ efflux to the atmosphere, and vice versa. The air–water CO₂
 272 flux due to abiotic factors (F_{ab}) and the contribution of biological effect to the flux (F_b)
 273 were also calculated using $f\text{CO}_{2\text{water}}$ calculated from $f\text{CO}_{2ab}$ and $f\text{CO}_{2b}$, instead of the
 274 $f\text{CO}_{2\text{water}}$, respectively. Because the temporal and spatial information for $f\text{CO}_{2\text{air}}$ from
 275 TUMSAT did not completely correspond to that of $f\text{CO}_{2\text{water}}$ measurements, the data
 276 were approximated by the data with the same *dist–monthlydata* grid information, which
 277 was calculated by natural neighbor interpolation using the original TUMSAT $f\text{CO}_{2\text{air}}$
 278 data. Because the OCU data did not include $f\text{CO}_{2\text{air}}$ measurements, we used the average
 279 value of the NIES data for Osaka Bay (392.6 μatm) for the calculation.

280 The gas transfer velocity k (here, the units are cm hr^{-1}) was calculated as
 281 follows (Wanninkhof, 2014):

282

$$283 \quad k = 0.251 \times U_{10}^2 \times (Sc/660)^{-0.5} \quad (5)$$

284

285 where U_{10} (m s^{-1}) is the wind speed at the height of 10 m from the water surface. Sc is
 286 the Schmidt number, defined as the ratio of the CO₂ molecular diffusion coefficient to
 287 the dynamic viscosity of seawater; Sc of seawater can be calculated from an empirical
 288 equation using the water temperature (Jähne et al., 1987). The wind data were taken
 289 from the database of the NEDO Offshore Wind System (NeoWins;
 290 http://app10.infoc.nedo.go.jp/Nedo_Webgis/top.html), which is the open database of the
 291 wind simulator provided by the New Energy and Industrial Technology Development
 292 Organization (NEDO) for evaluating the efficiency of offshore wind power generation
 293 in Japan. The wind data were supplied as the wind rose data, monthly averaged data,
 294 and averaged wind profile data, with 500-m resolution. We took the monthly averaged
 295 data at 10-km intervals beginning at *dist* = 0 along the water course (Table S1). Because
 296 the monthly averaged data were supplied as the data for 60–140 m above the water
 297 surface, while the wind profile was at 10–200-m height, we calculated the monthly
 298 averaged data at 10-m height according to the power-law of wind profile as follows:

299

$$300 \quad U_{10} = U_{60} \times (10/60)^n$$

$$301 \quad n = \frac{\log(U_{p60}/U_{p10})}{\log(60/10)} \quad (6)$$

302

303 where U_{60} (m s^{-1}) is the monthly averaged data at 60-m height. U_{p10} and U_{p60} are the

304 wind-profile data at 10-m and 60-m height, respectively (Pagon, 1935). The wind data
305 were applied to the carbonate measurement data for each of the 12 months and 10 km of
306 *dist.*

307

308 2.4 Data interpolation and error range

309

310 Because the measurement data were not homogeneous temporally or spatially,
311 the averages and deviations described in this study were calculated from the interpolated
312 data, which is a homogeneous grid of the distance (every 1 km *dist*) and date (every
313 0.1-monthly data) data ($n = 100 \times 120 = 12,000$). The error values for the carbonate
314 parameters are shown in Table S2.

315 The procedure used for error estimation is described in the Supporting
316 Information (Text S3). The error propagation was complicated because of the
317 non-linearity of the carbonate equilibrium calculation, so the error was separated into
318 the random error mainly due to the TA estimation and the riverine error due to
319 uncertainty in estimating the riverine DIC.

320

321 3. Results

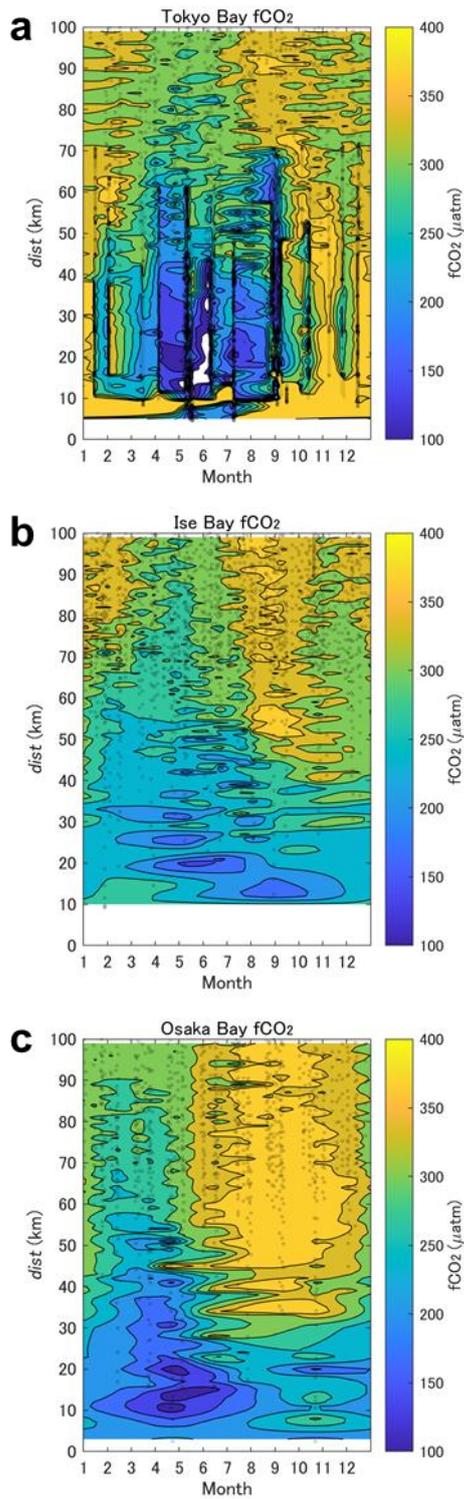
322

323 3.1 Temperature, salinity, and carbonate parameters

324

325 The water temperature and salinity (mean \pm SD) in each bay were $19.15 \pm$
326 4.59 °C and 31.42 ± 3.39 (Tokyo Bay), 18.78 ± 5.36 °C and 31.97 ± 2.23 (Ise Bay), and
327 18.66 ± 5.12 °C and 31.88 ± 1.43 (Osaka Bay), respectively (Figure S1). $f\text{CO}_2$ was
328 305.56 ± 87.80 μatm (Tokyo Bay), 294.88 ± 46.38 μatm (Ise Bay), and 300.56 ± 70.30
329 μatm (Osaka Bay) (Figure 3). The estimated TA and DIC were 2161 ± 125 and $1872 \pm$
330 129 $\mu\text{mol kg}^{-1}$ (Tokyo Bay), 2166 ± 115 and 1883 ± 120 $\mu\text{mol kg}^{-1}$ (Ise Bay), and 2179
331 ± 64 and 1893 ± 85 $\mu\text{mol kg}^{-1}$ (Osaka Bay) (Figure S2).

332



333

334

335

336

337

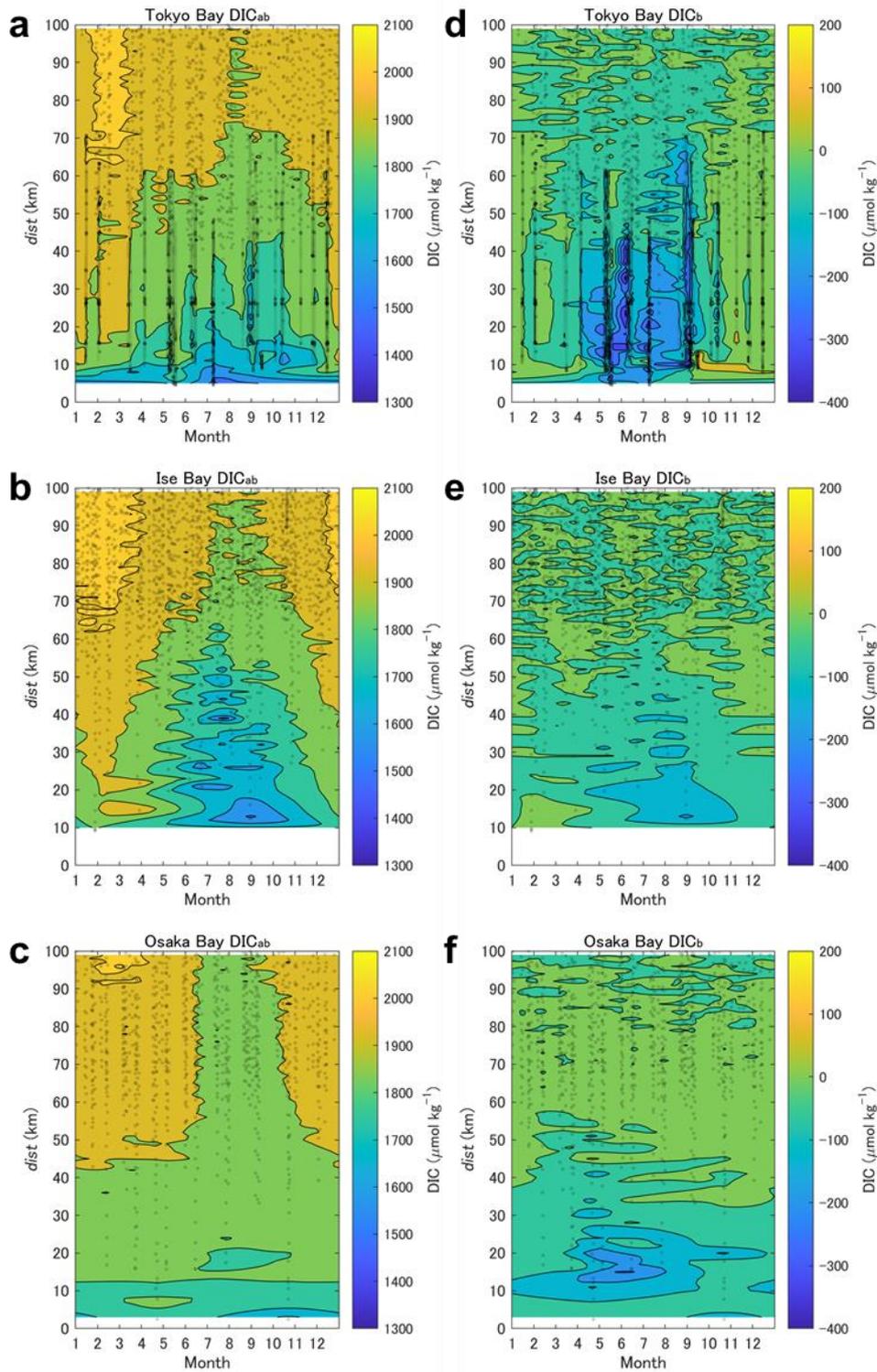
Figure 3. Temporal and spatial distributions of the fugacity of CO₂ (fCO₂) in (a) Tokyo, (b) Ise, and (c) Osaka Bays. The color indicates the interpolated 0.1-month × 1-km grid value. The gray dots show the distribution of direct measurements. The white area in (a) indicates extreme values that were excluded from the analysis for clarity. The parameter *dist* represents the distance from a zero

338 point in the mouth of the main river feeding the inner bay (equation [1] in the main text).

339

340 The salinity and DIC of the oceanic endmembers were calculated for each
341 month as the average values between 90 and 100 km from the river mouth reference
342 point in each bay: 34.34 ± 0.31 and $1968 \pm 35 \mu\text{mol kg}^{-1}$ in Tokyo Bay, 34.19 ± 0.39
343 and $1978 \pm 45 \mu\text{mol kg}^{-1}$ in Ise Bay, and 33.43 ± 0.66 and $1958 \pm 45 \mu\text{mol kg}^{-1}$ in
344 Osaka Bay, respectively. These values were higher during winter and lower during
345 summer and were consistent with the results from an empirical equation derived for the
346 Kuroshio stream area (Ishii et al., 2011) (Figure S3). The DIC of the riverine
347 endmember was estimated to be $1162 \pm 200 \mu\text{mol kg}^{-1}$ in Tokyo Bay, $675 \pm 200 \mu\text{mol}$
348 kg^{-1} in Ise Bay, and $852 \pm 200 \mu\text{mol kg}^{-1}$ in Osaka Bay.

349 The DIC_{ab} and DIC_{b} were 1899 ± 87 and $-27 \pm 64 \mu\text{mol kg}^{-1}$ in Tokyo Bay,
350 1896 ± 97 and $-13 \pm 28 \mu\text{mol kg}^{-1}$ in Ise Bay, and 1906 ± 55 and $-13 \pm 39 \mu\text{mol kg}^{-1}$ in
351 Osaka Bay, respectively (Figure 4). The estimated DIC_{b} showed a significant decrease
352 within 70 km from land in summer in all bays. Meanwhile, there was an increase in
353 Tokyo Bay within about 10 km of land from October to January. The seasonal trend of
354 photosynthetic activity (decrease in DIC_{b}) was almost the same among the bays. The
355 magnitude and spatial distribution of the photosynthetic activity was the greatest in
356 Tokyo Bay, where respiration and/or organic-matter decomposition (increase in DIC_{b})
357 were evident mainly in autumn to winter. In the other two bays, the magnitude and
358 spatial distribution of DIC_{b} were almost the same, and smaller than in Tokyo Bay.



359

360

361

362

363

Figure 4. Temporal and spatial distributions of abiotic dissolved inorganic carbon (DIC_{ab}) ((a): Tokyo Bay, (b): Ise Bay, (c): Osaka Bay) and biotic DIC (DIC_b) ((d): Tokyo Bay, (e): Ise Bay, (f): Osaka Bay). The colors and dots are as defined in Figure 3. The parameter *dist* represents the distance from a zero point in the mouth of the main river feeding the inner bay (equation [1]) in the

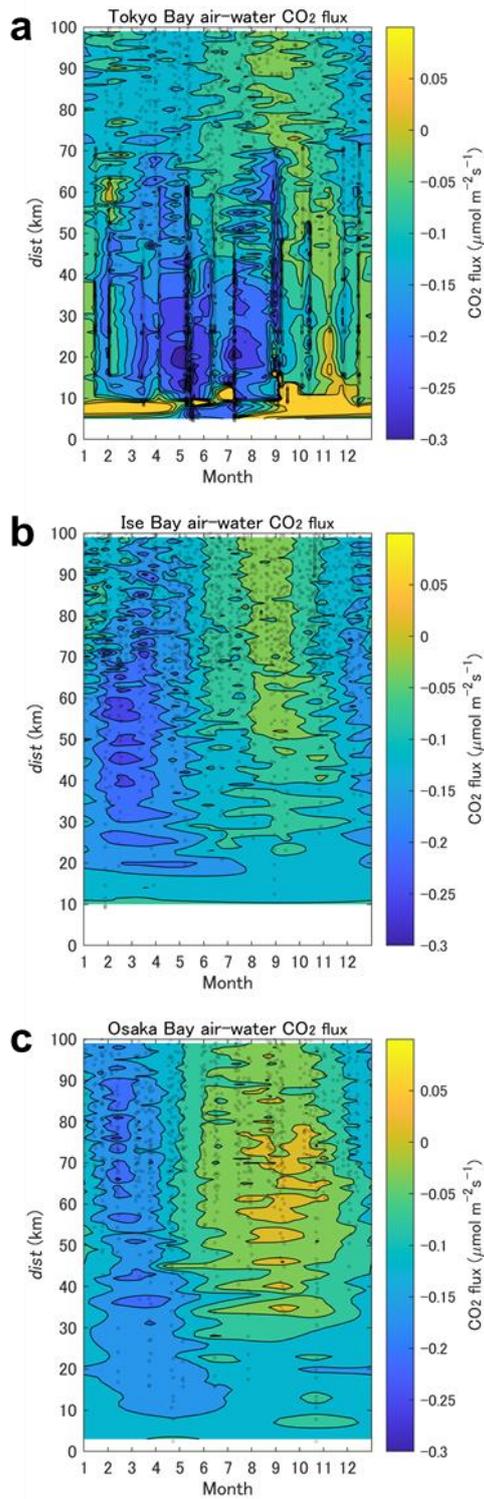
364 main text).

365

366 3.2 Air–water CO₂ fluxes

367

368 The air–water CO₂ fluxes in the three bays indicated that these areas were
369 annual atmospheric CO₂ sinks (-2.87 ± 2.39 , -3.20 ± 1.52 and -2.44 ± 1.71 mol m⁻² yr⁻
370 ¹ in Tokyo, Ise and Osaka Bays, respectively) (Figure 5). The temporal and spatial
371 distribution of the air–water CO₂ flux in Ise and Osaka Bays were similar and showed a
372 seasonal pattern whereas the distribution in Tokyo Bay seemed to be more consistent
373 with that of DIC_b than a seasonal pattern (Figure 4). The peak of CO₂ absorption was
374 during February to April, and CO₂ absorption was lowest in August and September in
375 Ise and Osaka Bays, and in October and November in Tokyo Bay.



376

377 **Figure 5.** Temporal and spatial distribution of air–water CO₂ flux in (a) Tokyo, (b) Ise, and (c)
 378 Osaka Bays. The colors and dots are as defined in Figure 3. The parameter *dist* represents the
 379 distance from a zero point in the mouth of the main river feeding the inner bay (equation [1] in the
 380 main text).

381

382 The fluxes without (F_{ab}) and from biological activity (F_b) were -2.14 ± 1.75
383 and $-0.73 \pm 2.68 \text{ mol m}^{-2} \text{ yr}^{-1}$ (Tokyo Bay), -2.82 ± 1.73 and $-0.38 \pm 1.09 \text{ mol m}^{-2} \text{ yr}^{-1}$
384 (Ise Bay), and -2.31 ± 1.51 and $-0.13 \pm 1.09 \text{ mol m}^{-2} \text{ yr}^{-1}$ (Osaka Bay), respectively
385 (Figure S4). The standard deviations of the biotic fluxes (F_b) indicated that the
386 biological effect in Ise and Osaka Bays was weaker than in Tokyo Bay. The temporal
387 and spatial distribution of F_{ab} showed a seasonal pattern whereas the distribution of F_b
388 seemed to be related to the distribution of DIC_b .

389

390

391 **4. Discussion**

392

393 The validity of our estimates of biological effects on DIC (DIC_b) and CO_2 flux
394 (F_b) depended on the precision of the riverine DIC endmember because the abiotic DIC
395 (DIC_{ab}) and CO_2 flux (F_{ab}) were determined from the riverine endmember, along with
396 the oceanic endmember whose error was relatively small. Although we could not find
397 reasonable reference data for riverine DIC in the bays, the reported riverine TA data by
398 Taguchi et al. (2009) which are 1006, 518 and $759 \mu\text{mol kg}^{-1}$ in Tokyo, Ise, and Osaka
399 Bays, respectively, support the validity of the DIC estimation because water with a
400 higher TA can contain more carbonate and bicarbonate ions as DIC under the same
401 fCO_2 conditions. The range of estimated riverine fCO_2 in each bay also supports the
402 validity of our estimates of riverine DIC and its precision ($\pm 200 \mu\text{mol kg}^{-1}$) (see Text
403 S2).

404

405 Our results support the hypothesized mechanisms related to wastewater
406 treatment (Kubo et al., 2017; Kuwae et al., 2016). Typical wastewater treatment method
407 removes carbon in the form of sludge and CO_2 gas more efficiently than nutrients such
408 as nitrogen and phosphorus (Sedlak, 1991). Through these effluents, the balance of
409 primary production and respiration in inner bay is offset towards an excess of primary
410 production and the resultant suppression of fCO_2 increase. Second, the effluent contains
411 relatively refractory carbon, because labile organic matter has already been removed by
412 treatment (Kubo et al. 2015). Therefore, respiration and mineralization rates of effluent
413 are low, and subsequent fCO_2 increases are suppressed.

413

414 The negative average annual biotic DIC in all three of the bays indicates that
415 the ecosystems were net autotrophic. The magnitude of the biotic DIC seemed to be
416 consistent with the nutrient concentrations reported in the reference material of the
Ministry of the Environment, Japan

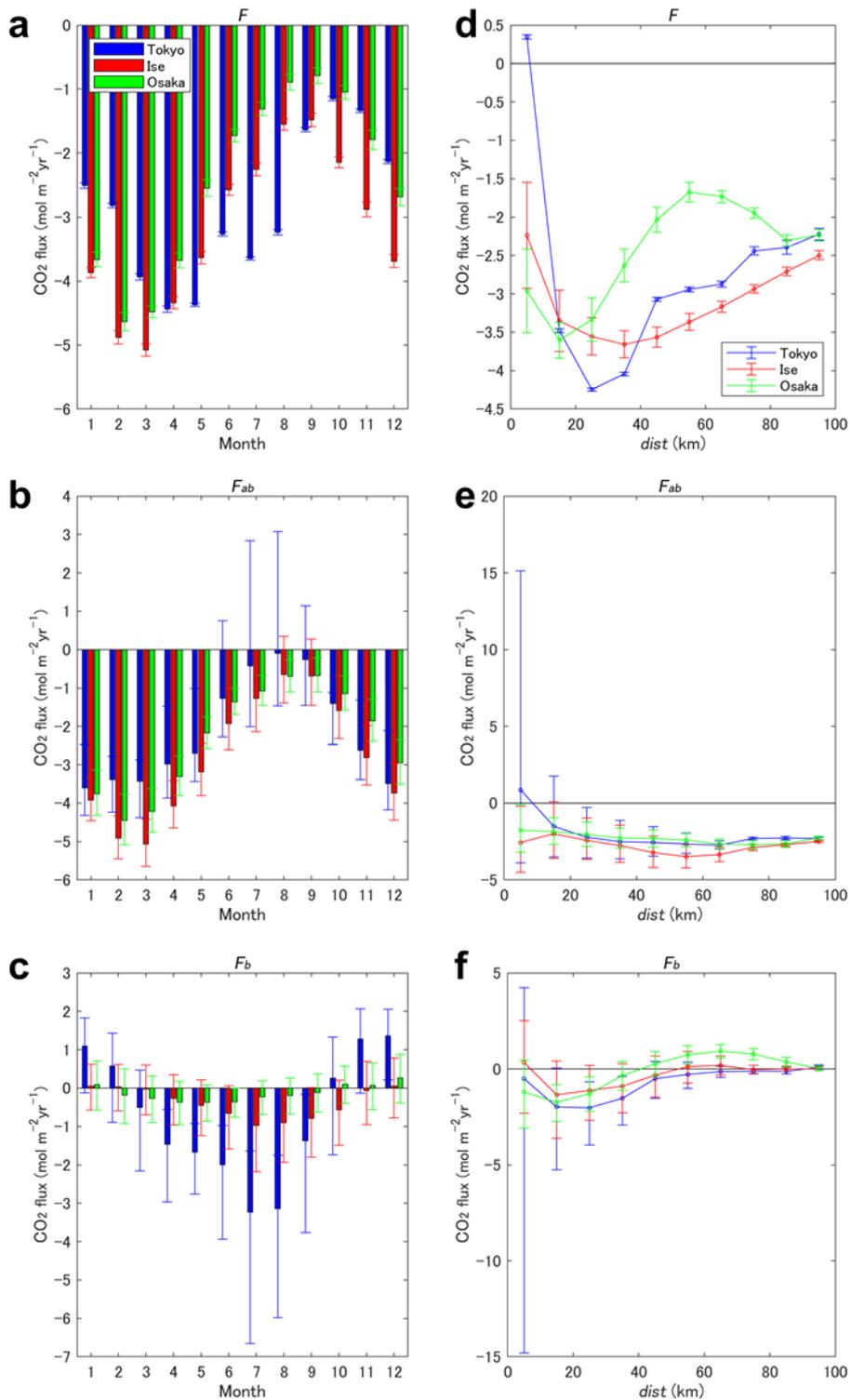
417 (<https://www.env.go.jp/council/09water/y0917-07/ref02.pdf>). In this report, the highest
418 concentrations of total nitrogen and phosphorous were in Tokyo Bay, and were almost
419 the same in Osaka and Ise Bays (average total nitrate, 0.8, 0.3, 0.4 mg L⁻¹; average total
420 phosphate, 0.06, 0.03, 0.04 mg L⁻¹ in Tokyo, Ise, and Osaka Bays, respectively). The
421 nutrient concentrations in the three bays were probably determined by the volume of
422 treated wastewater discharged into the bays, which is related to the size of the
423 population of the surrounding urbanized area. For example, the inflow of the treated
424 water is reported as the main component of the total freshwater inflow to Tokyo Bay
425 and thus the effect of treated water should be noticeable on the nutrient concentration in
426 the bays (Kubo et al., 2015). Although the net primary production in the bays depends
427 on several parameters such as the seawater residence time and vertical stratification, its
428 general magnitude would be similar to that of the biotic DIC because it is similarly
429 influenced by hydrographic conditions.

430 It is possible that the net primary production in Osaka Bay was underestimated
431 because Osaka Bay connects with the Seto Inland Sea at its landward end and this
432 topography results in strong tidal currents in the bay (Odamaki, 2002). Stronger tides
433 would enhance the water exchange between the bay and the surrounding areas and
434 weaken the biological effects on DIC and the air–water CO₂ flux. In addition, the area at
435 *dist* of 50–100 km coincides with narrow straits (Kitan Strait and Kii channel) where
436 several large rivers flow into the bay. Thus, the calculation of the endmember effect in
437 Osaka Bay might be biased compared with those for the other two bays.

438 The CO₂ fluxes found in this study indicate that the inner bays and surrounding
439 outer bays in Japan are one of the largest atmospheric CO₂ sinks among the global
440 coastal areas reviewed in previous studies (Borges et al., 2005; Chen and Borges, 2008).
441 The overall average CO₂ flux for the bays (–2.84 mol m⁻² yr⁻¹) is more negative than
442 the average in these previous studies for estuaries (7.74–10.26 mol m⁻² yr⁻¹) and
443 marginal seas (continental shelves) (–1.64 to –1.06 mol m⁻² yr⁻¹). The absorption in the
444 bays was mainly based on the abiotic flux (75%, 88%, and 95% of the net CO₂ flux (*F*)
445 in Tokyo, Ise, and Osaka Bays, respectively). The oceanic endmember from the
446 Kuroshio stream area is the most plausible explanation for the CO₂ absorption (overall
447 average CO₂ absorption of –2.32 mol m⁻² yr⁻¹). That area has been reported as the one
448 of the largest CO₂ sinks in the world because of the cooling effect of cold Oyashio water
449 on warm Kuroshio water (Takahashi et al., 2002, 2009). We suggest that the CO₂
450 absorption in the bays was enhanced by additional cooling due to the terrestrial effect
451 during winter (Figure S1).

452 The biotic CO₂ flux was the largest in Tokyo Bay, followed in order by Ise Bay

453 and Osaka Bay. The magnitude was basically the same as that of the biotic DIC and
454 nutrient concentrations in each bay. The higher CO₂ absorption in Ise Bay compared to
455 Osaka Bay despite almost identical biotic DIC can be explained by the lower salinity
456 and TA in the near-shore area in Ise Bay (Figures S1 and S2). Because lower-TA water
457 has less buffering effect on fCO₂, the decrease in fCO₂ in Ise Bay was greater than that
458 in Osaka Bay even when the decrease in DIC was the same. Although the annual
459 average of the biological effect was limited (25%, 12%, and 5% of the net CO₂ flux in
460 Tokyo, Ise, and Osaka Bays, respectively), it affected the temporal and spatial
461 distribution of the air–water CO₂ flux. Temporally, the pattern of F_b was the opposite of
462 F_{ab} , with an influx (F_b) or efflux (F_{ab}) in summer and vice versa in winter (Figure 6).
463 The biological effect was the strongest in Tokyo Bay and less notable in the other bays,
464 as with the biotic DIC. As a result, the seasonal variation of the CO₂ flux in Tokyo Bay
465 was different from that in the other two bays despite having almost the same variation in
466 the abiotic CO₂ flux (Figure 6). For example, the peak CO₂ influx in winter in the bays
467 continued into summer in Tokyo Bay, but the winter influx in Tokyo Bay was about
468 two-thirds that in the other bays.



469
 470 **Figure 6.** Temporal (left) and spatial (right) variations in the air–water CO₂ flux (F) ((a), (d)), abiotic
 471 flux (F_{ab}) ((b), (e)) and biotic flux (F_b) ((c), (f)). Note that the unit is different from Figure 5. The
 472 error bars were estimated by using equation (S4). The parameter *dist* represents the distance from a
 473 zero point in the mouth of the main river feeding the inner bay (equation [1] in the main text).

474

475 The difference in the spatial distribution of F_b among the bays was less
476 noticeable than that in the temporal distribution because of the offset of the influx in
477 summer and efflux in winter in Tokyo Bay (Figure 6). The magnitude of the influx peak
478 at *dist* of about 20 km was almost the same among the bays whereas the efflux at *dist* >
479 50 km was the largest in Osaka Bay. This might be caused by the decomposition of the
480 organic matter produced by the photosynthesis at about 20 km *dist*.

481 Meanwhile, there was limited CO₂ released to atmosphere in the near-shore
482 area (*dist* ≤ 10 km) although fCO₂ of the riverine endmember was estimated at more
483 than 2000–5000 μatm in the case of the intermediate riverine DIC (Figure 5). As for
484 Tokyo Bay, there was a tendency toward an efflux in the near-shore area both in F_{ab} and
485 F_b (Figure 6), corresponding to the decrease in the DIC_{ab} in summer and the increase in
486 DIC_b in winter (Figures 4 and 5). Because the distribution of the decrease in DIC_{ab} was
487 consistent with the salinity distribution (Figures 4 and S1), an increase in riverine flow
488 from precipitation might cause a CO₂ release as an abiotic factor. On the other hand, the
489 increase in DIC_b was consistent with the increase in pCO₂ observed in a previous study
490 (Kubo et al., 2017) in which the increase was due to the weakening of stratification in
491 Tokyo Bay due to the cooling of surface water, a decrease in precipitation, and the
492 weakening of the seasonal southward wind. These factors probably contributed to
493 bringing high-fCO₂ water and resuspended organic sediments from the bottom to the
494 surface, resulting in the CO₂ efflux to the atmosphere. Although we could not perform a
495 similar analysis for the other two bays because of a lack of measurement data, we would
496 expect a similar, considerable efflux tendency because the other bays share the same
497 hydrographic and climate conditions.

498 The comparison between Tokyo Bay and other two bays suggests that the
499 enhanced urbanization results in increased biotic CO₂ absorption. Because the
500 development of coastal areas will likely continue for several decades, the biotic
501 absorption of CO₂ in the bays is expected to be a mitigating factor for future climate
502 change. We assumed that the area within 100 km of the global coastline (6.2×10^7 km²)
503 could absorb atmospheric CO₂ additionally at the same rate as the biotic flux in Tokyo
504 Bay ($0.73 \text{ mol m}^{-2} \text{ yr}^{-1}$), and roughly estimated the potential for additional biotic
505 absorption to be 0.054 Pg yr^{-1} . This is on the same order of magnitude as the estimated
506 estuarine CO₂ efflux (e.g., Chen et al., 2013). The estimation of worldwide abiotic CO₂
507 flux is difficult because the oceanic and riverine carbonate parameters differ at each
508 location. In addition, the carbon export from urbanized areas is observed to increase
509 along with the development (Barnes and Raymond, 2009; Lopes et al., 2020; Wang et

510 al., 2017). This increase in carbon export could mitigate the increase in biotic CO₂
511 absorption.

512 For accurate estimates of the global CO₂ absorption resulting from coastal
513 urbanization, it is necessary to obtain more carbonate measurements from a variety of
514 areas. For example, data from measurements at the lagoons in Ivory Coast and
515 Guanabara Bay in Brazil suggested that strong CO₂ absorption and release are mixing in
516 the tropical urbanized bay where the rate of wastewater treatment is low (Cotovicz et al.,
517 2015; Koné et al., 2009). Both of these studies indicated that eutrophication and
518 stratification were the factors regulating the undersaturation of CO₂. Meanwhile, studies
519 in the Chesapeake Bay reported that both the absorption and release of CO₂ were
520 observed by numerical simulation model and abundant pH measurement, respectively
521 (Herrmann et al., 2020; St-Laurent et al., 2020). Because the main bottleneck for further
522 study is the difficulty of obtaining comprehensive measurements in coastal areas, the
523 development of novel methods like the above Chesapeake Bay studies will facilitate
524 filling gaps in the temporal and spatial distributions of available data. The review study
525 of such measurements is expected to understand the quantitative relationship between
526 the coastal urbanization and the carbon cycle.

527

528

529 **5. Conclusions**

530

531 In this study we clarified the temporal and spatial variations of carbonate
532 parameters in three bays in Japan from riverine water to the outer bays within a range of
533 100 km. Our results are the first to fill the gap between studies of nearshore areas and
534 the marginal seas. We found a notable atmospheric CO₂ influx due to biological activity
535 in the areas about 20 km from land in summer, accounting for 25%, 12%, and 5% of the
536 measured net CO₂ flux in Tokyo, Ise, and Osaka Bays, respectively. In addition, the
537 biological effect in the highly developed Tokyo Bay significantly affected the seasonal
538 variation of air–water CO₂ flux. The potential for atmospheric CO₂ absorption was
539 mainly regulated by the water in the surrounding marginal seas. The biological effect
540 seemed to be associated with the nutrient concentrations, which are related to the
541 volumes of treated wastewater entering the bays. In addition, the degree to which the
542 water area is enclosed likely influences the biological effect. Conditions that are more
543 closed or open would enhance or mitigate the biological effect on the air–water CO₂
544 flux, respectively.

545 This study should contribute to future investigations into the carbon cycle in

546 urbanized coastal areas, which will likely continue to expand for the next several
547 decades. However, more detailed investigations in inner and outer bays are required for
548 more precise evaluation of their contribution to the global carbon cycle. To expand the
549 results of this study to the global scale will require further measurements in bays in a
550 variety of regions.

551

552

553 **Acknowledgements**

554 We deeply appreciate the generous cooperation of Toyofuji Shipping Co. and
555 Kagoshima Senpaku Co. with the NIES VOS program. We thank the captains and crews
556 of M/S *Pyxis*, M/S *New Century 2*, M/S *Trans Future 5*, and TR/V *Shinyoumaru*. We
557 appreciate the help of K. Watanabe and H. Moki at the Port and Airport Research
558 Institute in their valuable comments for this study. This research was financially
559 supported by Global Environmental Research Coordination System, Ministry of the
560 Environment, Japan (grant number E1751).

561

562 **References**

563

564 Aufdenkampe, A. K., Mayorga, E., Raymond, P. A., Melack, J. M., Doney, S. C.,
565 Simone, R. A. et al. (2011). Riverine coupling of biogeochemical cycles between land,
566 oceans, and atmosphere. *Frontiers in Ecology and the Environment*, 9(1), 53–60.
567 <https://doi:10.1890/100014>

568

569 Barnes, R., & Raymond, P. A. (2009). The contribution of agricultural and urban
570 activities to inorganic carbon fluxes within temperate watersheds. *Chemical Geology*,
571 266 (3), 318–327. <https://doi:10.1016/j.chemgeo.2009.06.018>

572

573 Borges, A. V., Delille, D., & Frankignoulle, M. (2005). Budgeting sinks and sources of
574 CO₂ in the coastal ocean: Diversity of ecosystems counts. *Geophysical Research Letters*,
575 32, L14601. <https://doi:10.1029/2005GL023053>.

576

577 Cai, W.-J. (2011). Estuarine and coastal ocean carbon paradox: CO₂ sinks or sites of
578 terrestrial carbon incineration? *Annual Review of Marine Science*, 3, 123–145.
579 <https://doi:10.1146/annurev-marine-120709-142723>

580

581 Chen, C.-T. A., & Borges, A. V. (2009). Reconciling opposing views on carbon cycling
582 in the coastal ocean: Continental shelves as sinks and near-shore ecosystems as source
583 of atmospheric CO₂. *Deep Sea Research., Part II*, 56, 578–590.
584 <https://doi:10.1016/j.dsr2.2009.01.001>

585

586 Chen, C.-T. A., Huang, T.-H., Chen, Y.-C., Bai, Y., He, X., & Kang, Y. (2013). Air-sea
587 exchanges of CO₂ in the world's coastal seas. *Biogeosciences*, 10, 6509–6544.
588 <https://doi:10.5194/bg-10-6509-2013>

589

590 Cotovicz Jr, L. C., Knoppers, B. A., Brandini, N., Costa Santos, S. J., & Abril, G. (2015).
591 A strong CO₂ sink enhanced by eutrophication in a tropical coastal embayment
592 (Guanabara Bay, Rio de Janeiro, Brazil). *Biogeosciences*, 12, 6125–6146,
593 <https://doi.org/10.5194/bg-12-6125-2015>

594

595 Davison, J. R. (1991). Environmental effects on algal photosynthesis: *Temperature*.
596 *Journal of Phycology*, 27, 2–8.

597

598 Endo, T., Shimano, J., Sakai, D., & Fujiwara, R. (2017). Relationship between vertical
599 distribution of pCO₂ in sea and air-sea CO₂ exchange in inner part of Osaka Bay.
600 *Journal of Japan Society of Civil Engineering., B2 coastal engineering*, 73 (2), I_1231–
601 1236.
602

603 Fujii, T., Fujiwara, T., & Nakayama, K. (2013). Fluxes of carbon dioxide in the eastern
604 regions of Osaka Bay. *Journal of Japan Society of Civil Engineering., B2 coastal*
605 *engineering*, 69 (2), I_1111–1115.
606

607 Herrmann, M., Najjar, R. G., Da, F., Friedman, J. R., Friedrichs M. A. M., Goldberger, S.
608 et al. (2020). Challenges in quantifying air-water carbon dioxide flux using estuarine
609 water quality data: case study for Chesapeake Bay. *Journal of Geophysical Research*
610 *Oceans*, 125 (7), e2019JC015610. <https://doi.org/10.1029/2019JC015610>
611

612 Ishii, M., Kosugi, N., Sasano, D., Saito, S., Midorikawa, T., & Inoue, H. Y. (2011).
613 Ocean acidification off the south coast of Japan: A result from time series observation of
614 CO₂ parameters from 1994 to 2008. *Journal of Geophysical Research*, 116, C06022.
615 <https://doi:10.1029/2010JC006831>
616

617 Jähne, B., Heinz, G., & Dietrich, W. (1987). Measurement of the diffusion coefficients
618 of sparingly soluble gases in water with a modified Barrer method. *Journal of*
619 *Geophysical Research*, 92, 10767–10776. <https://doi:10.1029/JC092iC10p10767>
620

621 Kayanne, H., Suzuki, A., & Saito, H. (1995). Diurnal changes in the spatial pressure of
622 carbon dioxide in coral reef water. *Science*, 269, 214–216. [https://doi:](https://doi:10.1126/science.269.5221.214)
623 [10.1126/science.269.5221.214](https://doi:10.1126/science.269.5221.214)
624

625 Koné, Y. J. M., Abril, G., Kouadio, K. N., Delille, B., & Borges, A. V. (2009). Seasonal
626 variability of carbon dioxide in the rivers and lagoons of Ivory Coast (West Africa).
627 *Estuaries and Coasts*, 32, 246–260. [https://doi 10.1007/S12237-008-9121-0](https://doi:10.1007/S12237-008-9121-0)
628

629 Körtzinger, A. (1999). Determination of carbon dioxide partial pressure (pCO₂). In K.
630 Grasshoff, K. Kremling, M. Ehrhardt (Eds.), *Methods of Seawater Analysis* (pp. 149–
631 158), Weinheim: Wiley-VCH.
632

633 Kubo, A., Kawai, M. Y., & Kanda, J. (2015). Seasonal variations in concentration and

634 decomposition of dissolved organic carbon in Tokyo Bay, *Biogeosciences*, 12, 269–279.
635 <https://doi.org/10.5194/bg-12-269-2015>
636

637 Kubo, A., Maeda, Y., & Kanda, J. (2017). A significant net sink for CO₂ in Tokyo Bay.
638 *Scientific Reports*, 7:44355. <https://doi: 10.1038/srep44355>
639

640 Kuwae, T., Kanda, J., Kubo, A., Nakajima, F., Ogawa, H., Sohma, A., & Suzumura, M.
641 (2016). Blue carbon in human-dominated estuarine and shallow coastal systems. *Ambio*,
642 <https://doi:10.1007/s13280-015-0725-x>
643

644 Laruelle, G. G., Lauerwald, R., Pfeil, B., Regnier, P. (2014). Regionalized global budget
645 of the CO₂ exchange at the air-water interface in continental shelf seas. *Global*
646 *Biogeochemical Cycles*, 28, 1199–1214. <https://doi: 10.1002/2014GB004832>
647

648 Lee, K-S, Park, S. R., & Kim, Y. K. (2007). Effects of irradiance, temperature, and
649 nutrients on growth dynamics of seagrasses: A review. *Journal of Experimental Marine*
650 *Biology and Ecology*, 350 (1–2), 144–175. <https://doi.org/10.1016/j.jembe.2007.06.016>
651

652 Lopes, M. N., Decarli, C. J., Pinheiro-Silva, L., Lima, T. C., Leite, N. K. & Petrucio, M.
653 M. (2020). Urbanization increases carbon concentration and pCO₂ in subtropical
654 streams. *Environmental Science and Pollution Research*, 27, 18371–18381.
655 <https://doi.org/10.1007/s11356-020-08175-8>
656

657 Nakaoka, S., Telszewski, M., Nojiri, Y., Yasunaka, S., Miyazaki, C., Mukai, H., & Usui,
658 N. (2013). Estimating temporal and spatial variation of ocean surface pCO₂ in the North
659 Pacific using a self-organizing map neural network technique. *Biogeosciences*, 10,
660 6093–6106. <https://doi.org/10.5194/bg-10-6093-2013>
661

662 Odamaki, M. (2002). Improved co-tidal charts around Osaka Bay, Seto Inland Sea.
663 -Influence of Coriolis force on the tidal distribution-. *Report of Hydrographic*
664 *Researches*, 38, 85–99.
665

666 Pagon, W. W. (1935). Wind velocity in relation to height above ground. *Engineer's*
667 *News Record*, 114 (21), 742-751
668

669 Sedlak, R. I. (1991). *Phosphorus and nitrogen removal from municipal wastewater:*

670 *principals and practice*. Boca Raton: CRC Press.

671

672 Sibson, R. (1981). A brief description of natural neighbor interpolation. In V. Barnett
673 (Eds.), *Interpreting Multivariate Data* (Chapter 2 pp.21–36), Chichester: John Wiley.

674

675 St-Laurent, P., Friedrichs, M. A. M., Najjar, R. G., Shadwick, E. H., Tian, H., & Yao, Y.
676 (2020). Relative impacts of global changes and regional watershed changes on the
677 inorganic carbon balance of the Chesapeake Bay. *Biogeosciences*, 17, 3779–3796.
678 <https://doi.org/10.5194/bg-17-3779-2020>

679

680 Taguchi, F., Fujiwara, T., Yamada, Y., Fujita, K., & Sugiyama, M. (2009). Alkalinity in
681 coastal seas around Japan. *Bulletin on Coastal Oceanography*, 47 (1), 71–75.

682

683 Tait, L. W., & Schiel, D. R. (2013). Impacts of temperature on primary production and
684 respiration in naturally structured macroalgal assemblages, *PLoS One*, 8 (9), e74413.
685 <https://doi:10.1371/journal.pone.0074413>

686

687 Takahashi, T., Sutherland, S. C., Sweeney, C., Poisson, A., Metzl, N., Tilbrook, B., et al.
688 (2002). Global sea-air CO₂ flux based on climatological surface ocean pCO₂, and
689 seasonal biological and temperature effects. *Deep Sea Research., Part II*, 49, 1601–1622.
690 [https://doi:10.1016/S0967-0645\(02\)00003-6](https://doi:10.1016/S0967-0645(02)00003-6)

691

700 Takahashi, T., Sutherland, S. C., Wanninkhof, R., Sweeney, C., Feely, R. A., Chipman,
701 D. W., et al. (2009). Climatological mean and decadal change in surface ocean pCO₂,
702 and net sea-air CO₂ flux over the global oceans. *Deep Sea Research., Part II*, 56, 554–
703 577. <https://doi:10.1016/j.dsr2.2008.12.009>

704

705 Tokoro, T., Hosokawa, S., Miyoshi, E., Tada, K., Watanabe, K., Montani, S., et al.
706 (2014). Net uptake of atmospheric CO₂ by coastal submerged aquatic vegetation.
707 *Global Change Biology*, <https://doi:10.1111/gcb.12543>

708

709 Wang, X., He, Y., Yuan, X., Chen, H., Peng, C., Zhu, Q., et al. (2017). pCO₂ and CO₂
710 fluxes of the metropolitan river network in relation to the urbanization of Chongqing,
711 China. *Journal of Geophysical Research: Biogeosciences*, 122, 470–486, [https://doi:](https://doi:10.1002/2016JGR003494)
712 [10.1002/2016JGR003494](https://doi:10.1002/2016JGR003494)

713

714 Wanninkhof, R. (2014). Relationship between wind speed and gas exchange over the
715 ocean revised. *Limnology and Oceanology: Methods*, 12, 351–362. [https://doi:](https://doi.org/10.4319/lom.2014.12.351)
716 [10.4319/lom.2014.12.351](https://doi.org/10.4319/lom.2014.12.351)
717

718 Wanninkhof, R., Pickers, P. A., Omar, A. M., Sutton, A., Murata, A., Olsen, A., et al.
719 (2019). A surface ocean CO₂ reference network, SOCONET and associated marine
720 boundary layer CO₂ measurements. *Frontiers in Marine Science*, 12.
721 <https://doi.org/10.3389/fmars.2019.00400>
722

723 Weiss, R. F. (1974). Carbon dioxide in water and seawater: The solubility of a non-ideal
724 gas. *Marine Chemistry*, 2, 203–215. [https://doi.org/10.1016/0304-4203\(74\)90015-2](https://doi.org/10.1016/0304-4203(74)90015-2)
725

726 Zeebe, R. E., & Wolf-Gladrow, D. (2001). CO₂ in seawater: Equilibrium, Kinetics,
727 Isotopes. In *Elsevier Oceanography Series* (Vol. 65). Amsterdam: Elsevier.