

**Nutrient and Carbon Export from a Tidewater Glacier to the Coastal Ocean in the Canadian Arctic
Archipelago**

Patrick L. Williams^{1*}, David O. Burgess², Stephanie Waterman³, Megan Roberts⁴, Erin M. Bertrand⁴, and Maya P. Bhatia^{1*}

¹ Department of Earth and Atmospheric Sciences, University of Alberta, Edmonton, AB, Canada

² Geological Survey of Canada, Natural Resources Canada, Ottawa, ON

³ Department of Earth, Ocean and Atmospheric Sciences, University of British Columbia,
Vancouver, BC, Canada

⁴ Department of Biology, Dalhousie University, Halifax, NS, Canada

* Corresponding authors: Patrick Williams (plwillia@ualberta.ca) and Maya Bhatia
(mbhatia@ualberta.ca)

Key Points:

- Buoyant glacier meltwater plume entrains nutrient-rich deep water and delivers it to the ocean surface at a shallow tidewater glacier
- Glacial meltwater directly contributes labile carbon to the ocean near the glacier terminus
- Higher concentrations of Chlorophyll *a* are associated with areas of glacier-driven nutrient delivery

Abstract

As glaciers melt, a range of glacier processes modify and export freshwater and sediments to the ocean. This glacial runoff may influence biological productivity in coastal ecosystems by supplying essential nutrients and labile carbon. Previous studies of glacial meltwater export to the ocean have primarily been conducted on rivers draining land-terminating glaciers, or in fjords with large tidewater glaciers. These studies speculate about downstream effects (river studies) or upstream causes (fjord studies) of differing carbon and nutrient availability and biological productivity, but do not measure them. Here, we conduct the first ice-to-ocean study at a marine-terminating glacier in the Canadian Arctic Archipelago (CAA). We characterize the nutrient and carbon content of ice and meltwater collected on the glacier surface, at its margins, and in the near-shore coastal ocean, all within 1 to 25-km of the glacier terminus. Results demonstrate that while meltwater from a shallow tidewater glacier did not directly increase downstream carbon and nutrient concentrations, it can induce upwelling of deeper nutrient-rich marine water. Also, although carbon concentrations in meltwater were low, results show that this carbon is potentially more bioavailable than marine carbon. Glacially-mediated delivery of labile carbon and upwelling of nutrient-rich water occurs in summer, when surface waters are nutrient-limited. Collectively, these processes could benefit surface marine plankton, potentially stimulating production at the base of the food web. Shallow tidewater glaciers are commonly retreating in Arctic regions like the CAA and Svalbard, and understanding how increased meltwater output from these systems impacts marine ecosystems is critical.

Plain Language Summary

As glaciers melt, nutrients and carbon contained in runoff may impact recipient marine ecosystems. The last study to explore the relationship between tidewater glaciers and nutrient availability in the Canadian Arctic Archipelago (CAA) was in the 1970s. Here we measure nutrient and carbon concentrations in ice, glacial melt, and marine waters in front of a shallow tidewater glacier in the CAA. We find that nutrient and carbon concentrations in glacial melt are not high enough to augment downstream marine concentrations. However, the carbon in glacial melt appears more protein-like and may be more bioavailable than marine carbon. Additionally, with the release of submarine discharge at the terminal ice front, glacial meltwater entrains deeper nutrient-rich marine water and delivers nutrients to the surface as the meltwater plume rises. This upwelling is associated with the turbid meltwater plume and higher concentrations of chlorophyll. Upwelling of nutrients forced by a shallow tidewater glacier, common in the Canadian Arctic, could locally benefit surface marine plankton and stimulate production at the base of the food web.

1. Introduction

Polar ice caps and glaciers in the Canadian Arctic Archipelago (CAA), Greenland and Antarctica are melting faster than they were 30 years ago in response to climate change (Box et al., 2018; Shepherd et al., 2020). Compared to the polar ice sheets, the CAA is populated by smaller ice caps, icefields, and glaciers, and in the future, these ice masses may be particularly susceptible to warming air temperatures (Cook et al., 2019). Similar to Greenland and Antarctica, many ice caps and icefields in the CAA are drained by glaciers that terminate in the ocean (Cook et al., 2019). Recent studies show that glacial runoff into the coastal ocean can affect marine nutrient and carbon supply (Hawkings et al., 2015; Hood et al., 2009; Wadham et al., 2016), coastal circulation (Straneo & Cenedese, 2015), and biological productivity (Juul-Pedersen et al., 2015; Meire et al., 2017; Meire et al., 2015). Since most previous work investigating glacially-mediated nutrient delivery has been undertaken on large tidewater glaciers in Greenland, it is not clear whether the mechanisms by which large tidewater glaciers promote marine productivity apply to the smaller ice masses present in the CAA (Hopwood et al., 2018).

Traditional knowledge from northern communities document waters off glacier termini to be rich in wildlife (pers. comm. J. Qaapik, Grise Fiord Rangers). In 1938, “brown zones” in waters adjacent to glaciers around Disko Bay (Greenland) were identified as areas of upwelling that supported large populations of coastal birds (e.g. Kittiwake) which fed on zooplankton in a freshened meltwater plume (Hartley & Dunbar, 1938). The ability of glaciers to erode and deliver rock-derived nutrients like silicate (SiO_4^{4-}) and phosphate (PO_4^{3-}), important to downstream phytoplankton communities, was also recognized early in the 20th century (Vibe, 1939). In the most recent study of glacially-derived nutrients in marine waters in the CAA, Apollonio (1973) found elevated concentrations of nitrate (NO_3^-) and silicate within a glacierized fjord when compared to a non-glacierized fjord before the spring thaw. Apollonio noted that these nutrients were critical to arctic phytoplankton and augmented by glacial activity.

One main mechanism by which glacial melt can deliver nutrients and carbon to downstream marine environments is via direct delivery of chemical constituents in meltwater. In early summer, glacial runoff consists predominantly of surface snow melt which delivers a source of atmospherically-deposited nitrate to the marine environment (Wolff, 2013). As the melt season progresses, the proportion of ice melt in glacial runoff increases (Nienow et al.,

1998; Richards et al., 1996), which drains from the surface to the glacier bed via crevasses and moulins (Boon & Sharp, 2003; Das et al., 2008). At the bed, glacial meltwater can become chemically enriched in crustally-derived nutrients (e.g. silica, iron, and phosphorus) and carbon (Bhatia et al., 2013b; Hawkings et al., 2016; Hawkings et al., 2017; Hood et al., 2009) before discharging into the marine environment (Kanna et al., 2018). Numerous studies suggest that *in situ* microbial communities on the glacier surface or at the bed are capable of high rates of biogeochemical/physical weathering and cycling of organic carbon (Dubnick et al., 2017; Dubnick et al., 2020). *In situ* microbial nitrogen fixation at the glacier bed is a second important source of nitrate that may be delivered to marine waters (Boyd et al., 2011; Segawa et al., 2014; Telling et al., 2012; Wadham et al., 2016). These communities can further provide labile protein-like dissolved organic matter (DOM) to downstream environments (Bhatia et al., 2010; Hood et al., 2009; Musilova et al., 2017). Over the course of the melt season, basal flow evolves from a slow and distributed system, dominated by snow-melt and basal ice-melt, to a fast and channelized one, dominated by ice-melt originating from the surface (Flowers, 2015; Gray, 2005; Hubbard et al., 1995). This evolution leads to shorter retention and rock-water interaction times at the bed, and consequently lower entrained nutrient and carbon concentrations/fluxes during peak melt (Brown, 2002; Sharp, 2005).

A second mechanism by which glacial melt can facilitate nutrient addition to coastal waters is indirectly, via promoting the delivery of nutrients in deep nutrient-rich marine waters to the near-surface by entrainment, upwelling, and mixing. At the terminus of tidewater glaciers, runoff exits sub-glacially, sometimes hundreds of meters below the ocean surface (Straneo & Cenedese, 2015). As the buoyant meltwater plume rises, it can entrain deep marine water containing elevated levels of macronutrients (nitrate, phosphate, silicate) and transport it to the surface. This entrainment of nutrient-rich deep water has been tied to locally high rates of primary production observed in glacial fjords in Greenland and Svalbard (Halbach et al., 2019; Kanna et al., 2018; Meire et al., 2017). Additionally, estuarine circulation in fjords fed by glaciers can also drive upwelling and play an important role in nutrient delivery to the ocean surface in areas influenced by freshwater (Etherington et al., 2007). The strong tidal currents and shallow sill (moraine) entrances associated with glacial fjords and bays can further enhance vertical mixing, which in turn can enhance the delivery of deep-water nutrients to the surface (Etherington et al., 2007).

118 In the ocean, directly- or indirectly-sourced glacially-derived nutrients may fuel primary
119 autotrophic producers (phytoplankton) while labile carbon can feed microbial heterotrophs.
120 Phytoplankton communities require a host of macro- (e.g. nitrogen, phosphorus, silica) and
121 micro- (e.g. iron) nutrients to grow, but in the Arctic waters during the summer months, nitrogen
122 (N) is generally limiting following the spring bloom (Sorensen et al., 2017; Tremblay & Gagnon,
123 2009; Zhu et al., 2019). Since glacier meltwater delivery to the ocean occurs when NO_3^-
124 concentrations in surface waters are near zero, coastal phytoplankton communities could be
125 dependent on glacially-derived nutrients to sustain summer growth (Cape et al., 2018; Kanna et
126 al., 2018; Meire et al., 2017). In tandem, microbial heterotrophs may use glacially-derived
127 carbon, further stimulating higher trophic levels via the microbial loop (Azam & Malfatti, 2007).
128 Previous studies have found marine DOM to be recalcitrant, characterized by high humic-like
129 components, while glacial DOM tends to be more protein-like (bioavailable), suggesting that
130 glacially-derived carbon may better support downstream heterotrophic productivity (Bhatia et al.,
131 2013a; Hood et al., 2009; Musilova et al., 2017). The positive effects of glacial meltwater on the
132 availability of nutrients and carbon, and ultimately on productivity, are not necessarily restricted
133 to areas close to glacier termini, and they may extend further from shore to the continental shelf
134 (Cape et al., 2018; Painter et al., 2014).

135 Most previous work studying how glaciers impact marine nutrient and carbon availability
136 has been conducted either at land-terminating glaciers or in the ocean at large tidewater glaciers.
137 While some studies that span the ice-to-ocean continuum do exist (Halbach et al., 2019; Kanna et
138 al., 2018), there is a notable lack of research that considers the full ice-to-ocean system
139 (Hopwood et al., 2018; Meire et al., 2017). The absence of concurrent measurements on the ice
140 and in the ocean makes it challenging to determine whether enhanced nutrient concentrations
141 observed in coastal waters near tidewater glaciers (Kanna et al., 2018; Meire et al., 2017) are
142 controlled by direct delivery, deep water entrainment, or enhanced estuarine circulation.
143 Additionally, the regional focus on glacier systems in Greenland to date has led to a bias in the
144 modern literature towards large glaciers with deep submarine discharges draining into long fiords
145 at depths ≥ 140 meters below sea level (Cape et al., 2018; Kanna et al., 2018; Meire et al., 2017).
146 This bias may be problematic as according to these studies, the strength of meltwater-induced
147 upwelling, and thus the rate of indirect nutrient delivery, is largely dependent on the depth at
148 which submarine discharge enters the ocean and thus on the depth of the glacier grounding line

(Hopwood et al., 2018). Numerical models, based on these Greenland studies and parameterized using deep outlet glacier systems (Hopwood et al., 2018; Oliver et al., 2020), propose a productivity continuum between tidewater and land-terminating glaciers. These models predict that as submarine discharge from tidewater glaciers becomes shallower, less nutrient-rich deep water is delivered to the surface, and productivity enhancements decline as a result (Hopwood et al., 2018). Further, these models indicate that if the glacier grounding line shoaled above a given threshold depth (280 ± 200 m depth in the numerical model studied by Hopwood et al.), indirect nutrient delivery becomes decoupled from the glacier meltwater flux, suggesting that deep and shallow tidewater glaciers may impact indirect nutrient delivery to shallow waters in different ways.

Very few measurements have been made at intermediate-depth (Meire et al., 2017) and shallow-outlet (Halbach et al., 2019) tidewater glaciers. However, across the Arctic, intermediate-depth and shallow-outlet tidewater glaciers are common. For example, in the Queen Elizabeth Islands (northern CAA), the grounding line depth of tidewater glaciers averages ~ 230 m depth (Van Wychen et al., 2014) while on Baffin and Bylot Islands (southern CAA) grounding lines are estimated to be ~ 100 m depth on average (Van Wychen et al., 2015). Similarly, in the Svalbard archipelago, the average grounding line depth is estimated to be ~ 100 m depth (Błaszczuk et al., 2009). These glacier systems are significantly shallower than typical tidewater glaciers in Greenland, where the average grounding line depth is ~ 280 m depth (Morlighem et al., 2017). Further, there is *in situ* evidence that shallow-outlet tidewater glaciers have the potential to positively impact productivity: in a recent study of tidewater glaciers with grounding lines of ≤ 70 m depth in Kongsfjorden, Svalbard, Halbach et al., (2019) reported the presence of glacially-induced upwelling of nutrients in the fjord. Considering this result and the prevalence of shallow-to-intermediate depth outlet tidewater glaciers across the Arctic, further observations of shallow-terminating tidewater glaciers are necessary to gain a more complete understanding of the impacts of melting glaciers on coastal biogeochemistry.

With the goal of determining how a shallow tidewater glacier impacts nutrient and carbon availability in the proximate ocean, we conducted an ice-to-ocean study at Sverdrup Glacier, Devon Island in the CAA. In contrast to many previous study sites, submarine discharge exits Sverdrup Glacier relatively close to the surface. Here, we present *in situ* observations along a full ice-to-ocean transect with observations extending from the glacier surface and margins upstream

of the glacier terminus, through the turbid subglacial discharge plume in the coastal ocean, to more than 25 km out into open water (Jones Sound). Our study builds upon a very small number of studies that have incorporated both on-ice and marine data to date (Halbach et al., 2019; Kanna et al., 2018), and is the first in the CAA to document the biogeochemical influence of glacial melt routed through the marginal and subglacial environments from ice to ocean.

2. Materials and Methods

2.1. Site Description

2.1.1 Sverdrup Glacier

In 2019, spring (April 12 - May 12) and summer (July 22 - August 16) field sampling campaigns were undertaken on Sverdrup Glacier, a polythermal marine-terminating glacier located on the north coast of Devon Island, Nunavut Canada that drains ~805 km² (RGI Consortium, 2017) of the northwest sector of Devon ice cap. The 25-km long warm-based glacier overrides Precambrian metamorphic rocks of the Cumberland batholith, comprised primarily of granulitic high-K to shoshonitic monzogranite and granodiorite, and small amounts of low- and medium-K granitoid rocks (St-Onge et al., 2009; Whalen et al., 2010). Sverdrup Glacier's north-south oriented valley is bordered by steep walls with an average height of 300 m above the glacier surface (Vögtli, 1967). Surface mass balance and ice velocity measurements were first made on Sverdrup Glacier in the 1960s (Koerner, 1970; Koerner et al., 1961; World Glacier Monitoring, 2008), and six automatic weather stations (AWS) have been measuring air temperature and changes in height of the ice/snow surface within the Sverdrup glacier basin since 1999. The *in situ* measurements of ice velocity have shown that glacier flow rates typically increase early in the melt season, an event first measured in 1961 (Cress & Wyness, 1961). This seasonal acceleration points to a well-connected englacial/subglacial hydrological system driven by inputs of supraglacial and ice-marginal meltwater draining to the glacier bed upstream from the terminus (Wyatt & Sharp, 2017). Recent monitoring of Sverdrup glacier has shown larger annual melt volumes associated with changes in climate. Surface mass balance (SMB) remained only slightly negative up to the mid 1990's, then shifted to a period of increasingly negative

mass balance after 2005 when melt rates became ~4 times greater than the long-term average (Sharp et al., 2011).

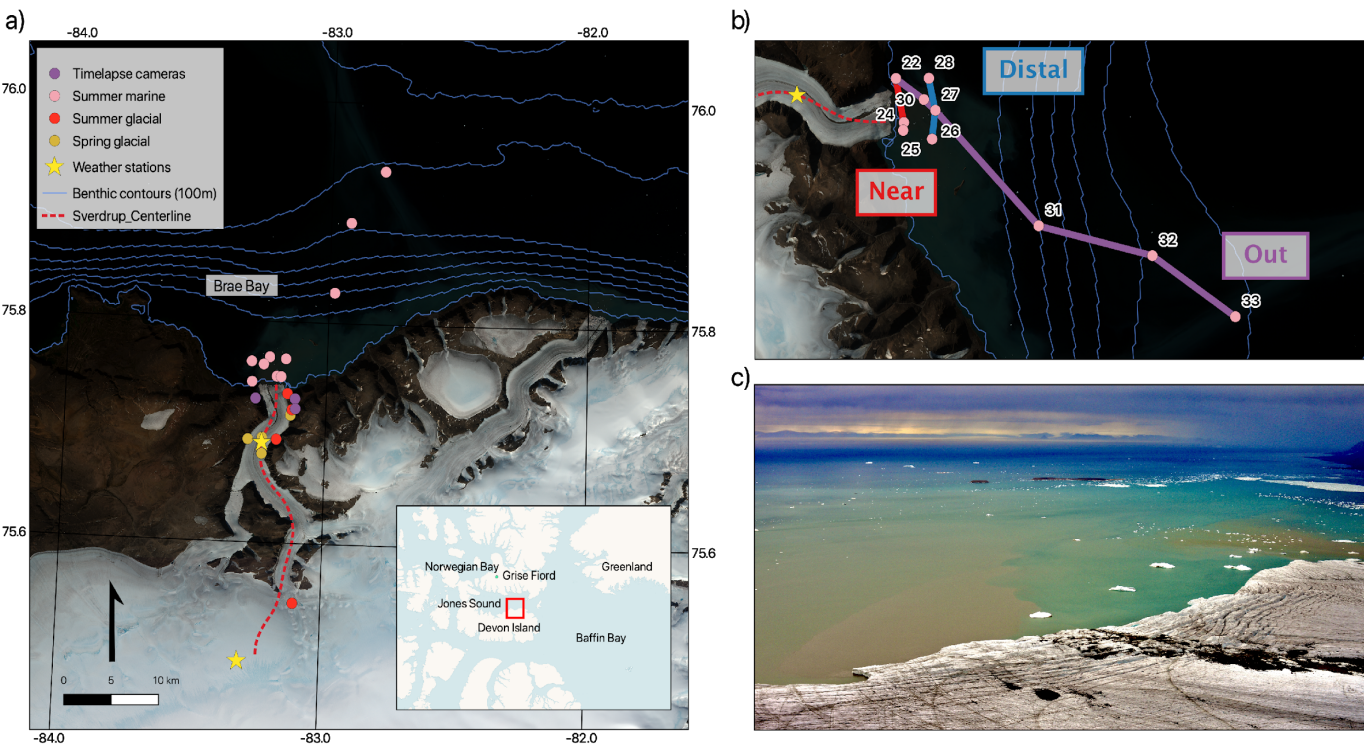


Figure 1. Map of study site. (a) Map of Sverdrup Glacier (Devon Island, Nunavut) showing 2019 spring on-ice (orange) and summer (red) sample sites, summer marine stations (pink), time lapse camera locations (purple), weather stations (yellow stars), bathymetry (blue lines), and the 2012 IceBridge centerline (red dotted line). (b) Enlarged view of Brae Bay showing the three “near” (red), “distal” (blue), and “out” (purple) transects used in this study. (c) View of the terminus of Sverdrup Glacier on July 23, 2019 taken from the western terminus time-lapse camera (orange circle outlined in pink) showing the turbid freshwater plumes at the glacier front. Image brightness and contrast have been heightened for better plume visualization (see Methods).

2.1.2 Marine Setting

Meltwater from Sverdrup Glacier discharges into a protected inlet, Brae Bay, Jones Sound (Figure 1). The 5.12 km calving front is grounded on the seafloor (Dowdeswell et al., 2004) with an annual calving flux of 0.006 Gt/y (Van Wychen et al., 2020). Based on a single airborne radar sounding transect from the 2012 NASA Ice Bridge field program, (Paden et al., 2019), ice within 1 km of the terminus is $\sim 20 \pm 10$ m thick (Sup. Figure 1). Unfortunately, the location of this centreline thickness measurement (Figure 1, red dotted line) does not coincide

with that of the outflows of either of the submarine plumes observed in 2019. While much of the surface meltwater runoff from Sverdrup Glacier is routed ice-marginally at higher elevations, historical field observations, as well as those made in 2019, reveal that the bulk of marginal meltwater enters the subglacial environment within 4 km of the glacier terminus (Keeler, 1964; Koerner et al., 1961). Due to the relatively low ice flow velocities on Sverdrup glacier (Cress & Wyness, 1961), fewer iceberg calving events have been observed here compared to other tidewater glaciers draining the ice cap (Cress & Wyness, 1961; Dowdeswell et al., 2004). This makes Sverdrup's terminus more readily accessible for oceanographic work than the termini of more active glaciers.

Once released into the marine environment, meltwater enters Jones Sound, a waterway between Devon Island and the southern end of Ellesmere Island. Water from the Arctic Ocean enters Jones Sound via Cardigan and Hellgate to the west and from Nares Strait to the east. Within Jones Sound, currents are cyclonic and the bulk of water exits the Sound into Baffin Bay and ultimately the North Atlantic (Barber & Huyer, 1977; Melling et al., 2008; Zhang et al., 2016). The bay in front of Sverdrup Glacier (Brae Bay) is hemmed by a series of submarine moraines extending ~9 km off-shore from Sverdrup's existing terminus; these moraines are located in shallow water, with some located less than 2 m below the surface (CHS Nautical Chart 7310, 2011).

2.2. Field instruments and sampling

2.2.1 On-Ice Instrumentation and Sampling

On-ice point measurements of surface mass balance were obtained from a network of 43 stakes drilled into the ice, and two automatic weather stations (AWS's) (Sup. Figure 1) in order to validate spatially continuous gridded model data across the Sverdrup glacier basin. The mass balance stake network spans the full elevational range from 100 to 1800 m a.s.l. including all glaciological zones within the Sverdrup glacier basin. The upper AWS, i.e. DICS, used in this study is situated at 1300m a.s.l., near the long-term equilibrium line altitude, while the lower SVD station at 400 m a.s.l. is located in the ablation zone where the glacier surface thins by ~ 1 m annually due to summer melting. Air temperature and change in ice/snow surface height data

from these AWSs (Figure 2) provide high temporal (hourly) resolution for tracking the evolution of the melt season; the latter are used to further assess bias in surface height modelling (see Section 3.1).

Time-lapse cameras were deployed at three different locations on the glacier in April 2019 (Figure 1) to capture the seasonal evolution of surface and marginal melt and to constrain characteristics of the freshwater plume that enters Jones Sound. These installations used Nikon D-3200 cameras fitted with Nikkor 28 mm lenses to capture high-quality JPEG images. Cameras were programmed to take an image every hour, provided there was enough light. The first photo was taken on April 28, 2019 and images were downloaded on August 9, 2019. 271 photos were taken by the time lapse cameras, but only images that were taken after the sea ice broke up and were minimally impacted by cloud / fog were used (13 images total).

Spring samples from different glacier “end-member” freshwater sources (basal ice, supraglacial snow / ice, and water stored at the base) were collected between April 23 and May 7, 2019. Bulk ice / snow / water samples were collected aseptically in trace metal clean ProPak® bags (Teledyne ISCO) using an ethanol-rinsed and flame-sterilized steel chisel and aluminum ice axe. Dissolved Organic Carbon (DOC) concentration and DOM fluorescence samples were collected in pre-combusted amber glass EPA vials with PTFE-lined septa. DOC samples were acidified with trace-metal grade concentrated HCl after collection to pH≈2. Samples were stored frozen and in the dark until analysed in the laboratory.

Summer 2019 freshwater melt samples from supraglacial and marginal runoff streams were collected between July 29 and August 15 and filtered in the field. Samples were collected in cleaned and sterilized 2 L Teflon bottles. Nutrient and oxygen isotope samples were filtered with sterile 60 mL plastic syringes, passed through a 0.22 µm polyethersulfone (PES) filter, and stored in HDPE scintillation vials. Oxygen isotope samples were stored in the dark at ambient temperature and nutrient samples were frozen within a few hours of collection. Samples for DOC, Total Dissolved Nitrogen (TDN), and DOM fluorescence were filtered with all-plastic polypropylene syringes (Norm-Jet), passed through a 0.22 µm PES filter, acidified to pH≈2 (DOC only) and stored in EPA vials as described above.

2.2.2 Marine Sampling

Ship-board work conducted from a polar sailboat (*S/Y Vagabond*) sampled the marine waters in front of Sverdrup Glacier from August 4-8, 2019 (Figure 1). Sensor-based hydrographic measurements, echo soundings, and bottle samples were taken at 12 marine stations, of which 10 spanned three individual transects (“near”, “distal”, and “out”) in front of the glacier terminus. Coordinates for all stations are provided in Sup. Table 1. Two lateral transects, one termed “near” (located ~0.8 km from the ice terminus, stations 22, 24, and 25) and the other “distal” (located ~2.5 km from the ice terminus, stations 26, 27, and 28), were sampled to gain insight into how glacial melt altered the near-shore marine environment in Brae Bay. The third transect (“out”, stations 22, 27, 30, 31, 32, and 33) followed the dispersion of a turbid plume from within 1 km of the ice terminus to >25 km out into Jones Sound in order to track the evolution in water column properties with increasing distance away from glacier terminus.

At each marine station, *in situ* measurements of electrical conductivity, temperature, pressure, dissolved oxygen, photosynthetically active radiation, chlorophyll *a* (Chl *a*), and turbidity were made using a RBRmaestro3 profiler (hereafter CTD). The CTD was hung from a Dynema rope and at each station was allowed to equilibrate just below the surface. The CTD was lowered by a winch at a rate of less than 1 m/s and recorded measurements at a frequency of 8 Hz. All data presented here were collected during the downcast.

Marine bottle sampling was also conducted at each station. Sample depths were chosen using data collected during the CTD downcast and visualized in real-time with the Ruskin iOS and Android app (RBR Ltd. 2017). At each station, multiple sample depths were selected: a near-surface depth, the depth of the deep chlorophyll maximum (if present), and one or two deeper sample depths (in the range of 50-400 m depth).

Marine water samples were collected using 10 L Teflon-lined, trace-metal-clean Go-Flo bottles (General Oceanic) that had been soaked in 0.1% acid detergent (Citranox), rinsed 3x with MilliQ, cleaned with isopropanol, soaked in 0.2 M HCl for 12 hours, and rinsed 3x with MilliQ (Cutter & Bruland, 2012). Nutrient and oxygen isotope samples were collected directly from the Go-Flo bottles with silicon tubing and filtered and stored as described above for the summer freshwater samples, with nutrient samples immediately frozen after filtration. DOC, TDN, and DOM fluorescence samples were also collected from the Go-Flo bottles into 2 L Teflon bottles, and filtered, preserved and stored like the summer freshwater samples described above. Chl *a*

samples were collected in 4 L polycarbonate bottles, and between 600-1600 mL was vacuum-filtered through a GF/F Whatman 47 mm filter in the dark, and then immediately frozen. All plasticware, glassware, and tubing was soaked overnight in a 10% HCL bath and washed 3x with MilliQ water. Glassware was then combusted at 560°C for ≥ 4 hours. All solvents used for cleaning and sample analysis were trace-metal grade or better. In the field, plasticware and glassware were rinsed 3x with sample water prior to collection.

2.3. Laboratory analyses

Prior to analysis, all frozen on-ice freshwater samples were thawed in a glass beaker in the dark at 4 °C. Frozen marine samples were thawed in the dark at 4 °C in the original collection bottles. Samples for nutrients (nitrate, nitrite, ammonia, phosphate, silicate), oxygen isotopes, DOC, TDN, and DOM fluorescence properties were filtered through a glass vacuum apparatus with 0.22 μm Teflon (PTFE) Omnipore filters into scintillation vials.

On-ice freshwater nutrient samples (nitrite, nitrate, phosphate, silicate, and ammonia) were analyzed on a Lachat QuikChem 8500 series 2 flow injection analyzer at the Biological Analytical Services Laboratory (University of Alberta), via photometric detection for simultaneous measurement of nutrient concentrations. Samples and reagents were continuously pumped through the system, loaded onto one or more injection valves, and mixed in the QuikChem manifold under laminar flow conditions. Limits of detection (LODs) for nitrite+nitrate, nitrite, ammonia, phosphate, and silica were: 0.15, 0.15, 0.21, 0.06, and 0.71 μM respectively.

Marine nutrient samples (nitrite, nitrate, phosphate, silicate, and ammonia) were analyzed on a Skalar SAN++ Continuous Flow Nutrient Analyzer at the Canada Excellence Research Chairs Ocean Laboratory (Dalhousie University). Reagents and samples, segmented with air bubbles, were pumped through a manifold for mixing and heating before entering the flow cell. Nitrite, nitrate, phosphate, and silicate concentrations were detected colorimetrically with optical background correction, while ammonia concentrations were determined with a fluorometer. LODs for nitrite, nitrate, ammonia, phosphate, silicate were: 0.3, 0.15, 0.01, 0.2, and 0.08 μM respectively.

DOC and TDN for both on-ice freshwater and marine samples were measured on a Shimadzu TOC-V (CPH) analyzer. DOC was quantified as non-purgeable organic carbon (NPOC) via high temperature combustion (680 °C) and TDN was measured with a total nitrogen module. A 6-point standard curve was used with $R^2 \geq 0.9986$ and $R^2 \geq 0.9994$ for DOC and TDN respectively. Standards were diluted from a 0.5 ppm stock solution for DOC (AccuSPEC, SCP Science) and from potassium nitrate for TDN (Sigma, KNO_3) analyses. Reference standards for deep seawater and low carbon water were obtained from the Consensus Reference Materials Project (Hansell Laboratory, University of Miami). MilliQ blanks and reference waters were analyzed routinely to monitor instrument drift, and remained within 5% of accepted values. The LOD was 2.5 μM for DOC and 3.33 μM for TDN. Procedural blanks using MilliQ water filtered through the plastic syringe and omnipore filters used in sample collection had DOC and TDN concentrations below the detection limit.

The fluorescent characteristics of DOM were analyzed with a Horiba Aqualog-3 spectrofluorometer equipped with a xenon lamp. Samples were brought to room temperature before analysis in a quartz glass cuvette with a 10 mm path length. Absorbance and excitation scans were measured in 5 nm intervals from 230-600 nm with an integration time of 10 s with 10 nm slits. Emission spectra were measured from 218-618 nm with an excitation offset of 18 nm. Ultrapure water in a dedicated cuvette (Mandel Scientific, SN-RM-H20) was used to validate the instrument. Excitation emission matrices (EEMs) were corrected with a MilliQ blank using the same settings.

Freshwater oxygen and deuterium isotopes were measured on a Picarro (L2140-i) at the University of Alberta while isotopes in marine samples were measured on a Picarro (L2130-i) at Dalhousie University. A volume of one μL of water was injected, vaporized, and introduced into the analyzer and measurements of $\delta^{18}\text{O}$ and δD were obtained using cavity ring down spectrometry. Certified water standards (USGS-46 and USGS-48) were used to normalize raw isotope ratios to the Vienna Standard Mean Ocean Water-Standard Light Antarctic Precipitation (VSMOW-SLAP) scale. For both instruments, analytical error was $<0.5\text{‰}$ for δD and $<0.15\text{‰}$ for $\delta^{18}\text{O}$ (one standard deviation) based on routine analysis of an internal deionized water standard (QCDI 6-2).

Chl *a* was measured using a Turner Designs AquaFluor Handheld Fluorometer following EPA Method 445 (Arar & Collins, 1997). Whatman 47 mm GF/F filters were extracted in 10 mL of 90% acetone for 18-24 hours. A 5 mL aliquot of the supernatant was transferred to a glass cuvette and the fluorescence was measured. Samples were then acidified to 0.003 N using 0.1 N HCl and fluorescence was measured again to account for interference from non-photosynthetic phaeopigments. The fluorometer was calibrated using a pure Chl *a* standard (C5753, Sigma). The LOD for Chl *a* analysis was 0.024 ug/L of seawater.

2.4. Data processing and analyses

2.4.1 Plume detection from time-lapse images

A k-means pixel classification was performed on a subset of the images from the time-lapse camera (13 images total) following Danielson and Sharp (2017) to detect the extent of the plume exiting Sverdrup's terminus. To minimize the effects of the sun's reflection, only images taken between 22:00 and 04:00 UTC were used. Land and sky were masked before pixel classification commenced. The k-means algorithm allowed for color-based plume detection at Sverdrup's terminus in a variety of light conditions. The process followed four steps: 1) data cleaning, filtering, and color-correction; 2) k-means classification; 3) pixel area to relative area conversion; and 4) comparison of plume area over time. The k-means pixel classification was conducted in R following the algorithms described in (MacKay, 2003). Ten clusters were used in the analysis. While the clustering analysis detected the plume, the calculated color was not consistent across images and was therefore selected manually for each image. Converting pixel areas to relative areas was also done in R using a monophotogrammetric technique from Krimmel and Rasmussen (1986).

2.4.2 CTD data processing

Raw CTD data were processed using the Matlab *rsktools* toolbox distributed by RBR Ltd. Measured conductivity, temperature, and water pressure were used to derive salinity, depth, and seawater density according to the 2010 thermodynamic equation of seawater (McDougal & Barker, 2011). Salinity, depth, dissolved oxygen, PAR, Chl *a*, and turbidity vertical profiles were

built by applying a low-pass filter to match sensor time constants using a three-sample running average, and channels were binned by pressure into 1-m intervals for further analysis. The euphotic zone depth, defined as the depth at which PAR=0.1% of the surface value (see Banse, 2004), was also calculated at each station using CTD measurements of PAR.

2.4.3 Optical properties of DOM

Parallel Factor Analysis (PARAFAC) is a statistical tool used to decompose trilinear data arrays to identify and quantify independent underlying signals or “components” (Bro, 1997). This technique can be applied to EEMs (excitation/emission matrices - a three-order array of sample name, excitation wavelength, and emission wavelength) to break down complex spectra into generalized DOM components (Stedmon & Markager, 2005). While these components cannot be ascribed to specific organic species, they can be compared to previously described DOM fractions. The drEEM toolbox in Matlab (Murphy et al., 2013) was used to model five individual fluorescent components. Corrections for instrument spectral bias and inner filter effects were applied and Raman scatter was normalized using daily scans. EEMs were smoothed and normalized to unit variance. PARAFAC models were validated using split-half analysis (Murphy et al., 2013), making sure that each split dataset contained a mix of fresh and marine samples. Modeled components were compared to previous glacial studies (Dubnick et al., 2017; Fellman et al., 2010a; Fellman et al., 2010b; Hood et al., 2009; Pautler et al., 2012; Walker et al., 2009) and other published models in the OpenFluor database (Murphy et al., 2014). To summarize optical DOM composition across samples, fluorescent intensity of each component was summed and normalized. Principal component analysis (PCA), analysis of variance (ANOVA), and permutational multivariate analysis of variance (PERMANOVA) were subsequently performed in *R* using the *vegan* package.

2.4.4 Apparent oxygen utilization calculations

Apparent oxygen utilization (AOU) is the difference between measured dissolved O₂ and the theoretical equilibrium saturation concentration in water with the same physical and chemical properties. Differences between measured and theoretical dissolved O₂ concentrations are

usually a result of biological activity: elevated primary production increases dissolved oxygen concentration, while respiration consumes oxygen and decreases dissolved oxygen concentration. Thus, AOU can be a measure of the sum of all biological activity a sample has undergone since its last contact with the surface (Garcia et al., 2013). AOU was calculated using measured temperature, dissolved oxygen, and salinity as per Benson and Krause (1984) with the *LakeMetabolizer* toolbox in R.

2.4.5 Statistical analyses

All further statistical analyses were conducted in R using the *akima*, *candisc*, *caret*, *cowplot*, *ecodist*, *ggbiplot*, *ggisoband*, *interp*, *klaR*, *MASS*, *MBA*, *NISTunits*, *oce*, *ocedata*, *openair*, *gdal*, *RVAideMemoire*, and *zoo* packages.

2.5 Glacier surface mass balance modeling

Finally, in order to better constrain the meltwater inputs to the marine system we modelled the surface mass balance of the Sverdrup Glacier basin for the time period spanning our on-ice and marine observations. To do this, we estimate total meltwater runoff for the Sverdrup glacier basin (as defined by the Randolph Glacier Inventory v6 RGI Consortium, 2017; Table S1) from the 1 km resolution RACMO2.3 regional climate model (Noël et al., 2018) over the 2019 melt season as:

$$MF = \sum_{k=1}^{days} \sum_{j=1}^{Nsb} Sb$$

where MF is the meltwater flux, $days$ is the number of days since Julian day (JD) 182 (July 1st), Nsb is the number of daily RACMO2.3 grid cells showing negative balance, and Sb is the value of each 1 x 1 km grid cell. Values of MF were converted from centimeters to kilometers to provide a measure in gigatons of total melt. We assume that all melt is routed to the tidewater terminus where it enters the ocean. As such, retention of meltwater within the remaining snowpack and / or firn is not accounted for in this study.

Independent validation of model performance over the Sverdrup glacier basin was performed by comparing spatially-coincident 1 km grid cells with *in situ* measurements of SMB as per Burgess (2018). Comparisons of cumulative SMB from RACMO2.3 with *in situ* measurements at each AWS (1 km resolution) provided daily validation of the intensity and duration of melt over the summer of 2019 as estimated from RACMO2.3 (Sup. Figure 2b and 2c). Results from these comparisons show that between measured and modeled SMB, RACMO2.3 results over-estimated summer melt by 20.4 mm w.e. at the SVD AWS and by 124 mm w.e. at the DICS AWS. It should be noted that the AWSs record single point measurements of ablation, while the RACMO2.3 data are averaged over 1km²; as such, some discrepancy between measured and modelled values is expected. While both AWSs are situated in fairly different settings, i.e. DICS is exposed to high winds on the ice cap proper and the SVD station is relatively sheltered from the wind by the surrounding mountains in the Sverdrup glacier valley, both sites are situated on very shallow slopes (<1°). AWS locations are also both characterised by relatively low relief (< 0.5 m) sastrugi (wave-like features in snow caused by wind erosion) during the winter months. A higher degree of spatial variability occurs at SVD station during the summer months where surface ponds, stream channels and cryoconites are more common than at the higher elevation DICS. Reduced albedo due to the presence of these features could account for the bias towards higher estimations of modelled melt than was measured at the SVD station. Uncertainty of the total melt discharge from the Sverdrup glacier basin in 2019 as modelled by RACMO2.3 was assessed through comparisons with melt/accumulation measured at each stake in the Sverdrup glacier basin over the period from 2008 to 2015. The standard deviation of the differences between RACMO2.3 and *in situ* measurements averaged for all stakes indicate an uncertainty of ±120 mm w.e., with better agreement (±90 mm w.e.) at higher elevations (≥1200m a.s.l.) than at lower elevations (±110 mm w.e. at ≤400 m a.s.l.). This standard deviation corresponds to an uncertainty of ±0.1 Gt (Sup. Figure 2a) in modelled estimates of total meltwater flux from the Sverdrup glacier basin.

3. Results

3.1. Meltwater export

AWS and RACMO2.3 SMB data provide context for the timing and volume of meltwater exported from Sverdrup in 2019. AWS data confirms that the spring season glacial samples were collected pre-melt (Julian Days 94-132, Figure 2 blue box) and that summer season samples were collected during peak melt (Julian Days 209-227, Figure 2 red box). The net SMB directly measured at both AWSs and at 43 ablation stakes from 2008-2015 are consistent with previous work comparing RACMO2.3 results to SMB in the CAA (Burgess, 2018) and are in agreement with past assessments of RACMO2.3 SMB, where errors between measured and modeled melt were generally good ($\pm 4\%$), except for terminus regions on Agassiz, Devon, and Penny ice caps in the CAA (Noël et al., 2018). RACMO2.3 data from 2018-2019 for the Sverdrup watershed shows 0.34 Gt of summer melt over a 55 day melt season (Figure 2c). The first sign of summer melt (JD 154) was followed by ~ 10 days of net accumulation, with daily melt volumes peaking when the plume was first observed (Figure 2b). Summer field sampling took place during the second highest period of daily surface melt, and as sampling took place toward the end of the melt season, cumulative surface melt was near its highest.

The time-lapse camera (TLC) imagery and field observations at Sverdrup's terminus provide an independent and complementary characterization of the seasonal timing and characteristics of the turbid meltwater plume released at the ice front in 2019. These images and observations showed two persistent patches of turbid water in front of the terminus, one smaller and one larger, which were interpreted as the signatures of freshwater subglacial plumes rising to the surface. The main plume appeared to be discharged on the western side of the glacier, while a smaller plume was evident on the eastern side. TLC images showed that the first signs of summer melt (Julian Day 154, Figure 2b blue box) and plume development (Julian Day 194 Figure 2b red box) occurred on June 3 and July 13 (2019), respectively. Sup. Figure 3 shows results of the k-means pixel classification and an example image from the data set. Detected plume area was correlated with modeled cumulative surface mass balance (i.e. plume area increased as Sverdrup glacier lost mass over the melt season) from the AWS ($r = -0.71$, $p = 0.015$, Sup. Figure 3c). This correlation gives confidence that detecting plume areas using this method is reasonable.

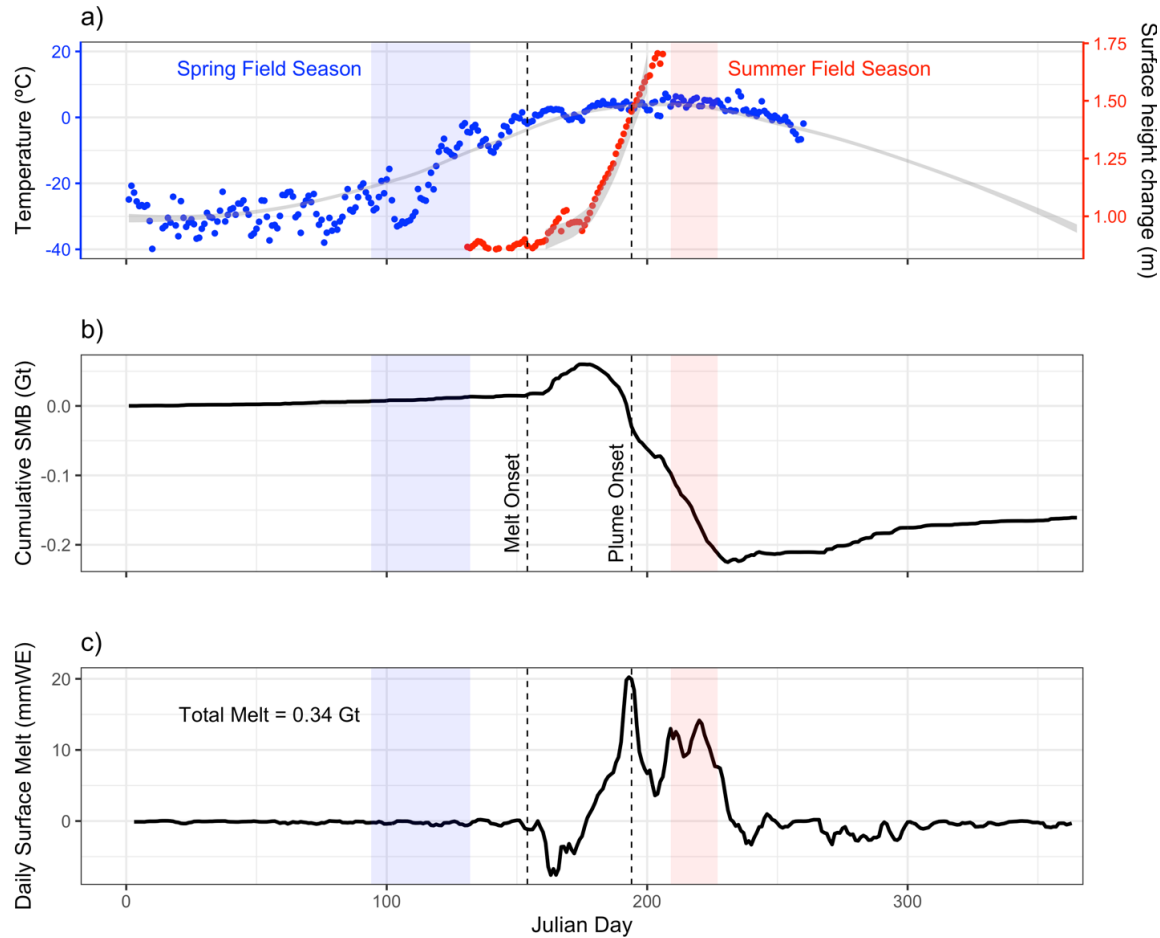


Figure 2. Sverdrup mass balance summary. (a) Temperature (blue) and surface height change (red) data from SVD AWS. Gray fill represents the 95% confidence interval of 2016-2018 temperature and surface height change AWS measurements. (b) Daily and (c) cumulative 2019 RACMO2.3 surface mass balance data for Sverdrup Glacier. Melt and plume onset dates determined using time lapse camera imagery (black dotted lines) and the duration of the spring and summer 2019 sampling periods (blue and red shading) are shown in all panels.

3.2. Meltwater nutrient delivery

Analyses of on-ice and marine bottle samples for nutrient concentrations gives insight into the role of glacial discharge in the direct delivery of chemical species to the marine environment. Table 1 shows a summary of macronutrient (NO_3^- , PO_4^{3-} , SiO_4^{4-} , NH_4^+) and TDN concentrations for glacial and marine samples. Marine samples are summarized for both the upper (≤ 40 -m depth) and deep (> 40 -m depth) water column in both relatively close proximity to the glacier front (≤ 4 km) and beyond 10 km from the glacier terminus. “Spring Glacial” samples

represent a variety of glacial environments (i.e., basal ice, supraglacial snow, supraglacial ice, and overwinter water) while “Summer Glacial” samples consist of marginal runoff and supra- and subglacial melt. Geochemically, spring glacial samples had higher macronutrient concentrations and fluxes compared to summer glacial samples, indicating that the export of macronutrients to the marine environment may have significant seasonal variability. In contrast, DON did not vary significantly with season. The lower concentrations of macronutrients in summer glacial samples likely reflect shorter retention and rock-water interaction times and the absence of snow in late-season melt (Nienow et al., 1998; Richards et al., 1996; Wolff, 2013). A higher degree of variability in spring relative to summer glacial samples likely reflects the diversity of sample types collected. Concentrations of macronutrients, except for ammonia, were all lower in glacial samples compared to deeper marine (>40-m depth) samples.

Table 1. Glacial freshwater and marine seawater values. Average and standard deviations for biogeochemical parameters (BLD = below limit of detection). Average values are given for marine depths as indicated. Samples ≤ 4 km from Sverdrup's terminus are within the moraines surrounding Brae Bay. The number of samples (n), is also given.

| Sample Type | n | NO ₃ ⁻ (μ M) | PO ₄ ³⁻ (μ M) | SiO ₄ ⁴⁻ (μ M) | NH ₄ ⁺ (μ M) | TDN (μ M) | $\delta^{18}\text{O}$ (‰) | DOC (μ M) | Chl a (μ g/L) |
|--|----|--|---|--|--|-------------------|------------------------------|-------------------|-----------------------|
| Spring Glacial | 10 | 2.2 \pm 0.3 | 0.3 \pm 0.0 | 5.0 \pm 0.2 | 1.4 \pm 0.1 | 1.8 \pm 0.2 | -27.8 \pm 0.4 | 16.4 \pm 1.1 | --- |
| Summer Glacial | 11 | 1.8 \pm 0.0 | 0.1 \pm 0.0 | *BDL | 1.2 \pm 0.0 | 1.8 \pm 0.0 | -26.7 \pm 0.2 | 10.9 \pm 0.4 | --- |
| Marine ($\leq 40\text{m}$, $\leq 4\text{km}$) | 18 | 2.0 \pm 1.8 | 0.5 \pm 0.2 | 5.6 \pm 3.4 | 1.5 \pm 1.8 | 5.6 \pm 2.8 | -3.0 \pm 1.0 | 63.9 \pm 44.2 | 2.2 \pm 2.4 |
| Marine ($\leq 40\text{m}$, $> 10\text{km}$) | 6 | 1.4 \pm 2.2 | 0.5 \pm 0.3 | 4.2 \pm 4.2 | 0.9 \pm 0.5 | 5.8 \pm 2.4 | -2.0 \pm 0.2 | 97.7 \pm 40.8 | 1.6 \pm 1.5 |
| Marine ($> 40\text{m}$, $\leq 4\text{km}$) | 3 | 6.6 \pm 0.2 | 0.9 \pm 0.1 | 12.9 \pm 0.3 | 0.3 \pm 0.5 | 12.2 \pm 1.2 | -1.7 \pm 0.0 | 126.9 \pm 20.7 | 0.1 \pm 0.1 |
| Marine ($> 40\text{m}$, $> 10\text{km}$) | 7 | 6.5 \pm 3.1 | 0.7 \pm 0.1 | 11.1 \pm 3.4 | 1.8 \pm 1.2 | 10.5 \pm 5.9 | -1.3 \pm 0.7 | 78.6 \pm 26.6 | 0.4 \pm 0.5 |

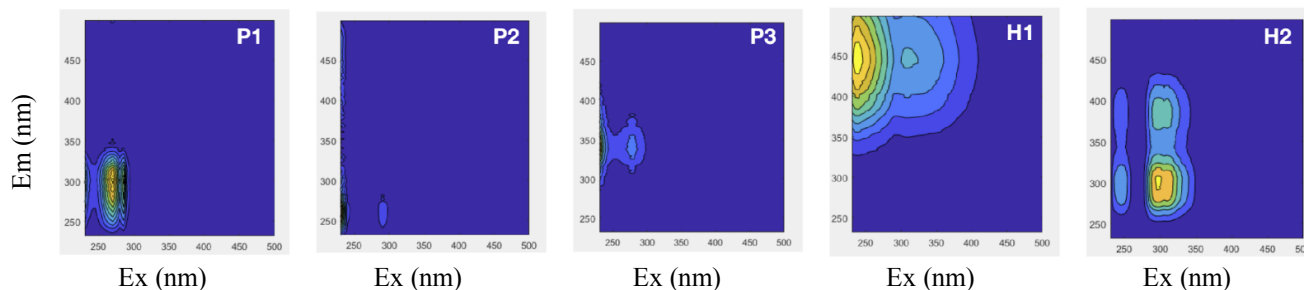
3.3. Meltwater carbon delivery

In addition to the potential for delivering nutrients, glacial discharge may also impact downstream carbon availability via the delivery of DOC in meltwater. Average and standard deviation DOC concentrations in spring and summer glacial samples as well as marine samples are shown in Table 1. Similar to macronutrient concentrations, there was more variability in DOC concentrations in spring glacial samples compared to summer samples, likely representing the larger variety of different sample types collected during the spring season. Further, also as with macronutrient concentrations, DOC concentrations in meltwater in both seasons were universally lower than marine concentrations. Given this, it appears that Sverdrup glacier does not export DOC in concentrations high enough to significantly augment DOC concentrations in the marine environment.

Though spring and summer glacial DOC concentrations were lower than those in marine samples, PARAFAC results show that the type of carbon present in spring and summer glacial water was significantly different than in marine waters; we thus explored the generalized DOM

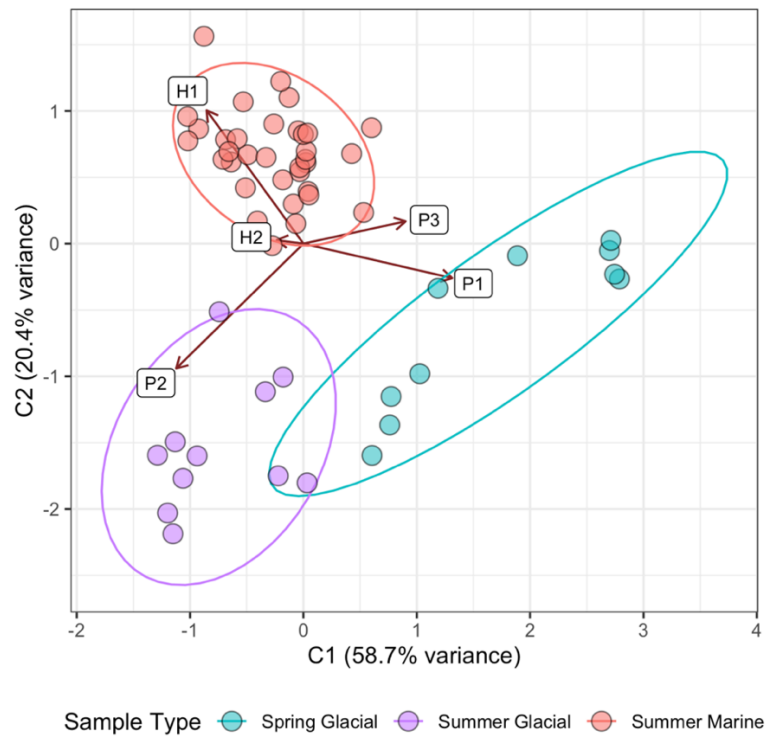
component composition of glacial and marine samples to gain insight into the possible influence of glacial input on DOM in the near-shore marine environment. A five-component PARAFAC model applied to all spring glacial, summer glacial, and marine samples explains 97.7% of the variance in the dataset. The loading patterns of the five modeled components can be matched to previously-described fluorescent DOM fingerprints in glacierized environments (Table 2). P1 (tyrosine) and P2 (tryptophan) match protein-like peaks identified in marine and terrestrial samples from around the world (Coble, 1996) and broadly indicate autochthonous production of DOM (Stedmon & Markager, 2005). P3 has been found in glacial ice and meltwaters from the McMurdo Dry Valleys (Antarctica) as well as on Axel Heiberg and Ellesmere Islands in the CAA (Dubnick et al., 2017; Pautler et al., 2012). Components H1 and H2 are similar to previously described humic-like peaks. H1 is similar to a humic-like component of terrestrial origin ubiquitous to a wide range of natural catchments during the warmer months of the year and generally absent in wastewater (Stedmon et al., 2007). H2 is similar to the classic M peak (Coble, 1996) and has been defined as a marine humic-like component. Respectively, spring and summer glacial samples contained >40% and ~18% more protein-like components than marine samples. In contrast, marine samples had >60% more humic-like DOM compared to summer glacial samples and >300% more humic-like DOM relative to spring glacial samples. Though bulk DOM concentrations in glacial melt were not high enough to significantly increase marine concentrations, proportionally, there was significantly more protein-like DOM in glacial melt vs. in marine waters, with more protein-like DOM in spring glacial melt compared to summer meltwater. There was also a higher fraction of the P1 component (associated with summer glacial melt) in the higher turbidity marine samples compared to marine samples outside the turbid meltwater plume. It thus appears that the freshened and turbid submarine meltwater plume delivers protein-like DOM to the surface of Brae Bay with a carbon signature similar to summer meltwater. This may be significant because secondary producers (marine heterotrophs) could benefit from this addition of bioavailable carbon.

Table 2. A summary of the 5 PARAFAC components. Components modeled using fresh and marine samples from Sverdrup Glacier and Brae Bay (n = 55). Described here are wavelengths (nm) of the component excitation (Ex) and emission (Em) spectral peaks, the potential carbon source (protein-like vs. humic-like), and examples of previous studies that have found similar peaks in similar environments.



| Component | Ex:Em (nm) | Potential Carbon Source | Literature Examples |
|-----------|-------------------|--|---|
| P1 | 270: 301 | Protein-like (Tyrosine) | Stedmon, 2005; Walker, 2009; Fellman, 2010 |
| P2 | 290: 265 | Protein-like (Tryptophan) | Coble, 1996; Walker, 2009; Fellman, 2010 |
| P3 | 280: 337 | Protein-like (autochthonous DOM via microbial degradation) | Coble, 2007; Pautler, 2012; Dubnick, 2010 |
| H1 | 235, 310:441 | Ubiquitous humic-like | Stedman, 2005; Stedmon, 2007; Dubnick, 2010 |
| H2 | 245, 295:300, 395 | Marine humic-like (microbial degradation) | Coble, 1996; Walker, 2009 |

To further assess seasonal and spatial differences in fluorescent DOM composition, a principal component analysis (PCA) was conducted using the relative abundance of the 5 modeled PARAFAC components (Figure 3). The first and second principal components described 58.7% and 20.4% of the variance in the normalized PARAFAC dataset, respectively. PCA results show a clear differentiation between glacial and marine samples and between the spring and summer glacial samples. A PERMANOVA test confirms that these clusters are significantly different ($p < 0.004$) while the ANOVA f-test ($f > 10^{20}$) confirms that this difference is due to between-group variability. P1 and P3 are associated with spring melt, P2 is associated with summer melt, and H1 and H2 are associated with summer marine samples. This analysis confirms the unique DOM signatures of the glacier meltwater relative to the marine waters, and further the seasonal evolution of meltwater DOM characteristics.



610

611 **Figure 3. Principal component analysis (PCA) of the five modeled PARAFAC components.**
612 Data is grouped by season (spring vs. summer) and water type (glacial vs. marine). A
613 PERMANOVA test ($p < 0.004$) confirms these clusters are significant while the ANOVA f-test (f
614 > 1020) confirms that this significance is due to between-group variability.

615

616 3.4. Glacial meltwater in the near-shore marine environment

617 The fate of glacial meltwater in the marine environment is mapped by measurements of
618 $\delta^{18}\text{O}$, salinity, oxygen, and turbidity in marine water sampled at various locations relative to the
619 glacier terminus. Specifically, marine profiles along the “near” (~0.8 km from the ice terminus),
620 “distal” (~2.5 km from the ice terminus), and “out” (from within 1-km to more than 25 km from
621 the ice terminus) transects suggests that glacial meltwater is largely confined to the upper 30-40
622 m of the water column, directly impacting waters ≤ 4 km from the glacier front. Marine water
623 column profiles show a spatial gradient in $\delta^{18}\text{O}$ (Figure 4a), salinity (Figure 4b), dissolved
624 oxygen (Figure 4c), and turbidity (Figure 4d), with fresher, more ^{18}O -depleted, oxygen-rich, and
625 turbid waters found closer to the ocean surface and the calving front. ^{18}O -depleted water is

characteristic of glacial meltwater due to Rayleigh fractionation (Tranter, 2011). The “out” transect (Figure 4) shows a clear spatial correlation between ^{18}O -depletion (glacial melt) and areas of low salinity, high dissolved oxygen, and high turbidity – all indicators of glacially-impacted waters. For all samples, water deeper than 10 m was less depleted in ^{18}O (average $\delta^{18}\text{O}$: -1.75‰) than water above 10 m (average $\delta^{18}\text{O}$: -3.45‰). Further, surface (≥ 10 m depth) samples of the “near” transect (Figure 5, top) were more depleted in ^{18}O (average $\delta^{18}\text{O}$: -4.24‰) than the “distal” transect (Figure 5 bottom, average $\delta^{18}\text{O}$: -3.42‰), which in turn were more depleted than surface samples collected >10 km from shore outside of the ring of moraines enclosing Brae Bay (average $\delta^{18}\text{O}$: -2.18‰). These values indicate a glacial meltwater signal in the marine environment which appears to be largely confined to upper 30-40 m of the water column and quickly diluted within 4 km of calving front. Rising submarine discharge plumes can be patchy (Andersen et al., 2010; Everett et al., 2018; Jackson et al., 2017), but using turbidity as an indicator, the plume can be detected as far out as station 27, ~ 3.7 km from Sverdrup’s terminus (Figure 4d). Turbidity thus corroborates the $\delta^{18}\text{O}$ picture of meltwater impacting waters primarily within 4 km from the glacier front. CTD sensor measurements of dissolved oxygen provide a more highly-resolved view of the potential meltwater plume and further show an extended glacial influence: a “plume-like” region of elevated dissolved oxygen concentration is observed within the top 20 m of the water column and extends to station 31, ~ 13 km from the terminus (Figure 4c). In the “near” and “distal” transects (Figure 5c) there is evidence of a subsurface plume with elevated dissolved oxygen concentrations centered at ~ 12 m depth in the “near” transect, which rises (centered ~ 10 m depth) and dilutes/disperses in the “distal” transect.

The mapped density structure indicates that the meltwater, which enters the marine environment at depth, rises to the surface within 4 km of Sverdrup’s terminus. The “plume-like” feature seen in $\delta^{18}\text{O}$, salinity, dissolved oxygen, and turbidity follow the $>1025 \text{ kg m}^{-3}$ isopycnal which slopes upwards from the terminus within the first 4 km of the “out” transect (Figure 4, Station 22-30, white lines). Upsloping isopycnals associated with the plume along this transect (i.e. those associated with densities $\leq 1026 \text{ kg m}^{-3}$) begin at depths ≥ 30 m depth (Figure 4); by linearly extrapolating these lines of equal density back to the terminus, it appears that the plume originates from depths between 30-40 m.

A two-component mixing model using summer marginal melt and Jones Sound deep water as end-members (Sup. Table 1) was constructed to quantify the fraction of glacially-

derived water in marine samples and to track its extent in the near-shore environment. The model uses $\delta^{18}\text{O}$ and salinity values of the most ^{18}O -depleted marginal runoff summer sample (Sup. Figure 4, “MR”) and $\delta^{18}\text{O}$ and salinity values of the most ^{18}O -enriched deep marine sample (Sup. Figure 4, “JS”) to calculate the fraction of glacial melt in all marine samples (Figure 6). Calculations of glacial meltwater fraction are based on similar work done by Östlund and Gert (1984) and Kanna et al. (2018). Surface waters (≥ 10 m) in the “near” transect have the highest meltwater fractions ($\sim 12\%$ glacial melt and $\sim 88\%$ marine water on average). However, even these fractions are low and the surface plume water contains significant amounts of marine water even in the freshest part of the sampled plume. The meltwater fraction declines with depth, where subsurface water (10-40 m below surface) averaged $\sim 6\%$ glacial melt, while deep waters (>40 m below surface) contained $<5\%$ glacial melt (Figure 6). Glacial melt fraction declines with distance from the glacier terminus; surface water within 4 km of shore was $\sim 12\%$ melt, while the average melt fraction >4 km from shore at the surface was $\sim 7\%$. Overall the model suggests the plume, as sampled, is diluted with marine water even at close proximity to the terminus; further it suggests glacial melt primarily impacts near-surface waters and is diluted/dispersed efficiently with distance from the terminus.

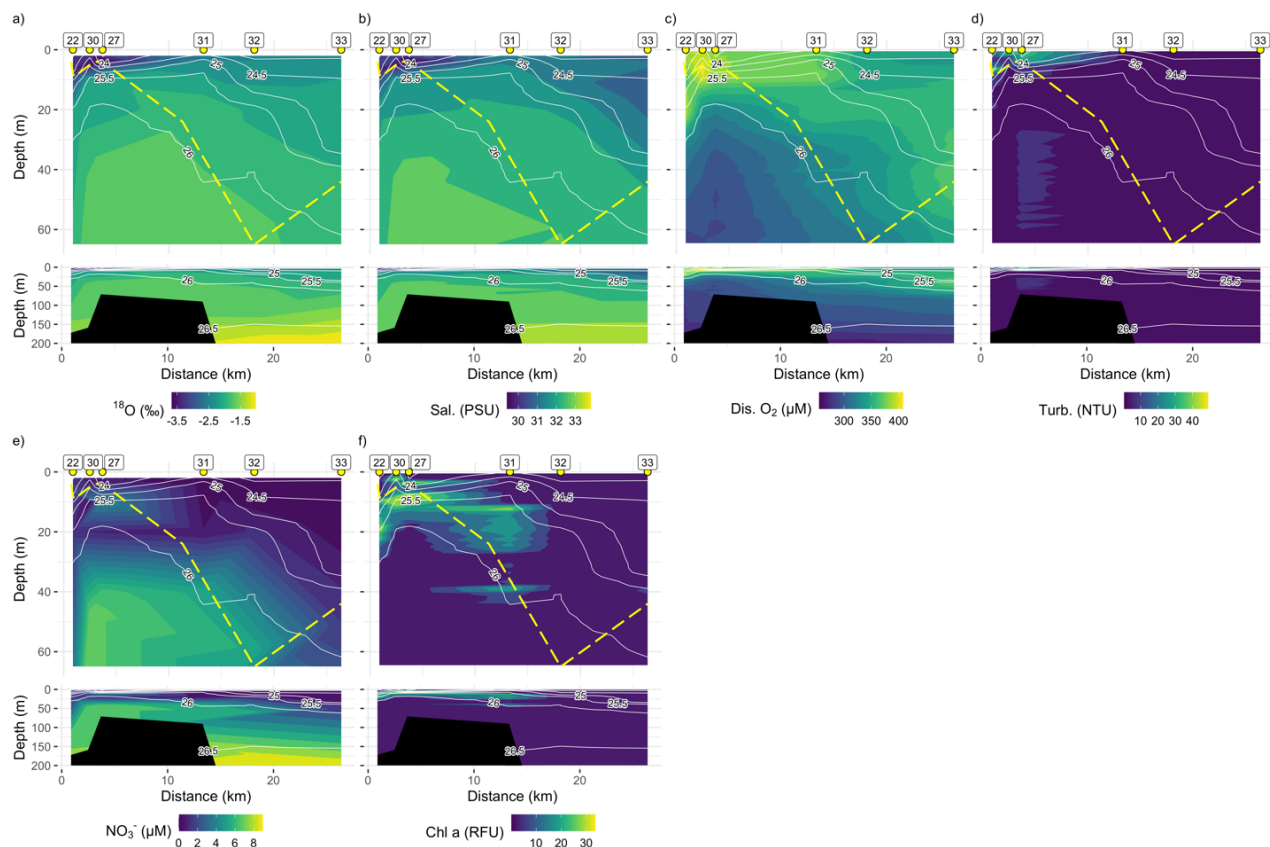


Figure 4. Plots of (a) $\delta^{18}\text{O}$, (b) salinity, (c) dissolved oxygen concentration, (d) turbidity, (e) nitrate concentration, and (f) Chl *a* concentration along the “out” transect in Brae Bay. Density anomaly (kg m^{-3}) contours are shown in white. The dotted yellow line represents euphotic depth, calculated at 0.1% of surface PAR. Only the NO_3^- concentration profile is shown, but PO_4^{3-} and SiO_4^{4-} concentrations follow similar patterns. Station numbers are indicated at the top of the plot and distance is defined as starting at the glacier calving front. Bathymetry data (black) is from echo soundings made at each station.

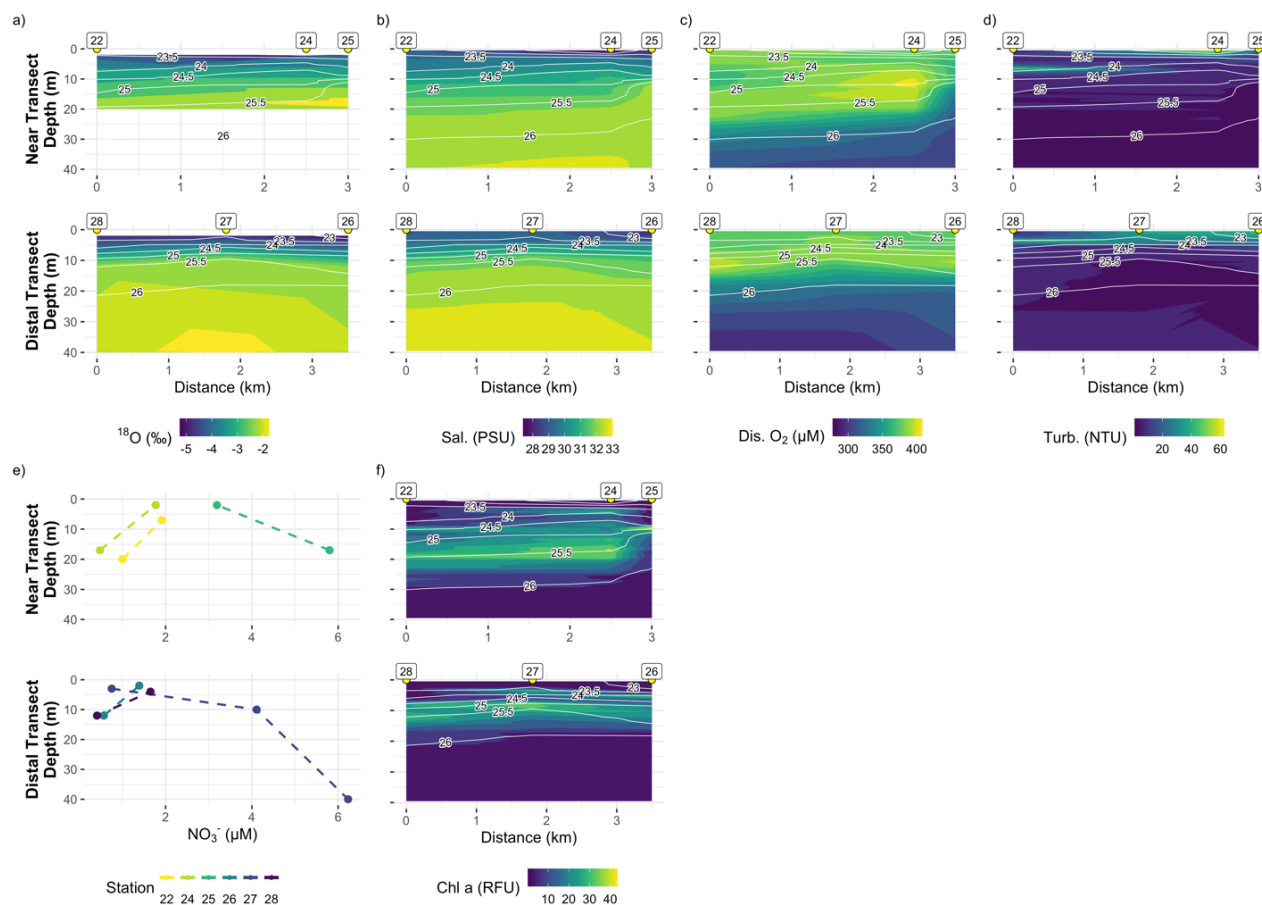


Figure 5. Plots of (a) $\delta^{18}\text{O}$, (b) salinity, (c) dissolved oxygen concentration, (d) turbidity, (e) nitrate concentration, and (f) Chl *a* concentration along the “near” (top) and “distal” (bottom) transects in Brae Bay. Density anomaly (kg m^{-3}) contours are shown in white. Only the NO_3^- concentration profile is shown, but PO_4^{3-} , and SiO_4^{4-} concentrations follow similar patterns. Station numbers are indicated at the top of the plot and distance is defined as starting at the first station along the lateral transect.

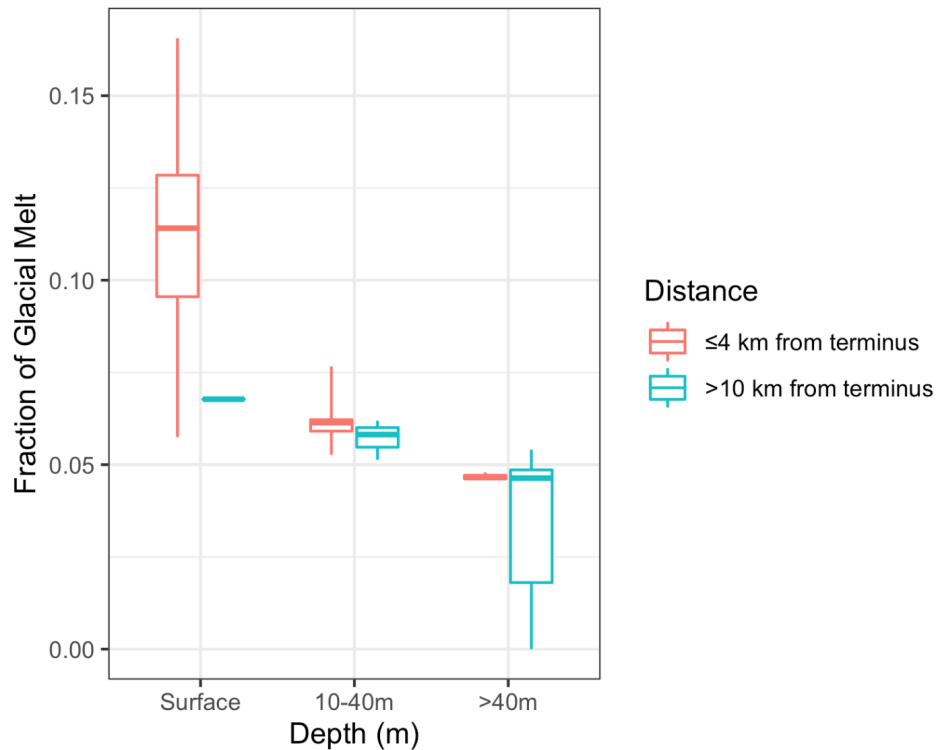


Figure 6. Boxplot of glacial melt fractions for all samples. The median and interquartile range for each water type are shown for marine surface water (0-10m depth), marine near-surface water (10-100m depth), and marine deep water (>100 m depth). Colors denote distance away from Sverdrup glacier's terminal ice edge.

3.5. Meltwater impacts on the marine environment

Finally, the impacts of glacial input on nutrient availability, light availability, and primary production are explored via marine water column measurements. Although nutrient concentrations in glacial vs. marine samples (Table 1) show that glacial meltwater does not significantly impact near-terminus marine water nutrient concentrations, marine measurements suggest that glacial input at Sverdrup glacier does drive the delivery of marine-sourced nutrients from deeper water to the near-surface. This delivery likely occurs via entrainment in the rising meltwater plume and/or the estuarine upwelling circulation forced by the glacier's freshwater input. The mapped density structure (Figure 4, white contours) indicates that isopycnals in the density range of 1025-1026 kg m⁻³ slope upwards towards the glacier terminus starting >26 km from the glacier. This structure provides an adiabatic pathway for marine waters at depths >60 m

in the open waters of Jones Sound to upwell to depths of 5-10 m in near-coastal waters in close proximity to the glacier terminus. Nutrient concentrations (nitrate, phosphate, and silicate) are generally lower at the surface and higher at depth (Table 1) as is typical in marine waters in the late summer (Randelhoff et al., 2020). Thus, the upwelling implied by the isopycnal structure likely plays a role in delivering marine waters with significant major nutrient concentrations to the near-surface. Measured nutrient concentrations (Figures 4e, 5e) are consistent with this scenario: nitrate concentrations are enhanced on the underside of the rising meltwater plume at concentration levels consistent with those of the 1025-1026 kg m⁻³ density classes. The entrainment observed at Sverdrup glacier is shallow compared to that observed at deep tidewater glaciers in Greenland (Kanna et al., 2018; Meire et al., 2017) but nevertheless appears important for enhancing nutrient concentrations: nutrient samples indicate that the nutricline (defined here as the depth where NO₃⁻ concentrations exceed 1-μM) at all the stations within Brae Bay (≤10 km of the glacier terminus) occurs at or above 30 m depth. Further, average NO₃⁻ concentrations in the upper 100 m of the “out” transect were higher at stations within Brae Bay (stations 22, 30, 27) than those farther out in Jones Sound (stations 30, 31, 32).

The conclusion that nutrients present in near-surface waters in close proximity to the glacier terminus are marine – as opposed to glacier-sourced – is further supported by the observed linear relationship between nutrient concentrations and AOU. The relationship between nutrient concentrations and AOU can be used to determine if marine nutrient concentrations are being impacted by direct addition of glacially-derived nutrients, as such a “disturbance” to a water mass is expected to cause a departure from a linear relationship. In this system, nitrate, phosphate, and silicate concentrations show linear relationships with AOU in both surface and subsurface water throughout Sverdrup Bay (Figure 7a,b,c), as expected for nutrients that are deep water-sourced. Expectedly, turbidity does not show this linear relationship, as turbid waters in this system are glacially-sourced (Figure 7d).

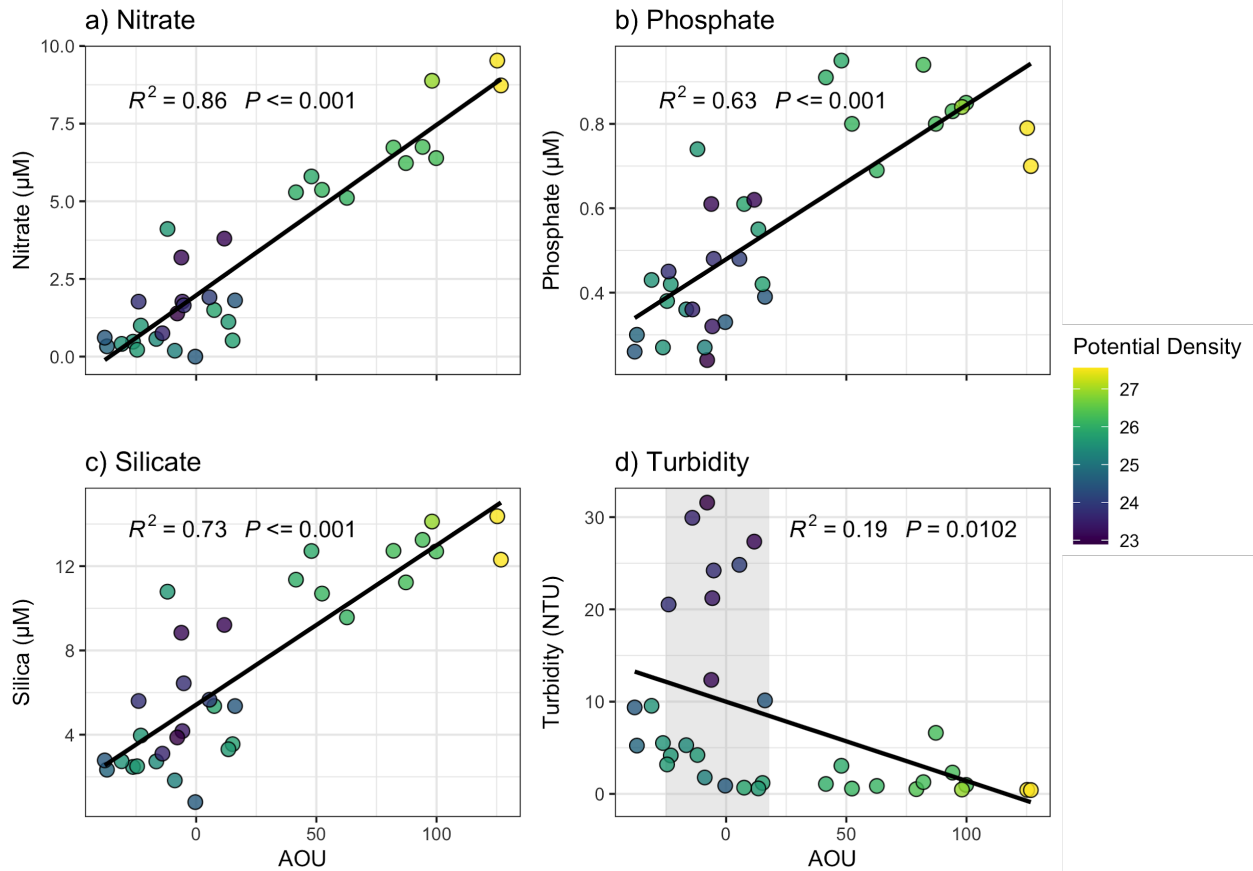


Figure 7. Apparent oxygen utilization (AOU) versus (a) NO_3^- , (b) PO_4^{3-} , (c) SiO_4^{4-} , and (d) turbidity of marine water in Sverdrup Bay. The colour scale shows the log of depth (m). The AOU range of plume water in (d) is shown in grey.

A second important impact of glacial input on the marine environment is its impact on light availability in near-surface coastal waters in close proximity to the glacier terminus. The export of sediment-laden glacial runoff from Sverdrup Glacier into Jones Sound leads to areas of high turbidity and low light availability in the upper ~10m of the water column close (<4 km) to the freshwater outlet at the glacier terminus (Figures 4d and 5d). As a consequence, the euphotic zone (Figure 4, above the yellow dotted line) in close proximity to the terminus is influenced significantly: at station 22 (that closest to the glacier terminus) the euphotic zone depth was 9 m and it decreased to less than 5 m at stations 30 and 27 as the buoyant turbid plume rose towards the surface with distance offshore. Consistent with other indicators of the meltwater plume, which suggest that the plume is quickly diluted within 4 km of the calving front (Section 3.4), euphotic zone depths increase to over 20 m beyond a distance of ~4 km from the ice front.

Glacially-induced nutrient entrainment and elevated turbidity limiting light availability in close proximity (within ~4km) of the terminus are likely to impact primary production in these waters, although the combined net influence is not straight-forward to predict. On a large scale, elevated near-surface Chl *a* concentrations were found at stations closest to the glacier front and declined with depth and distance away from the glacier terminus: on average, higher Chl *a* concentrations were present at all three “near” stations compared to stations on the “distal” transect (Figure 6f) and Chl *a* concentrations were higher at “near” and “distal” transect stations than at stations further from shore along the “out” transect (Figure 5f). On a smaller scale, relationships between Chl *a* concentration, turbidity, and nutrient concentrations were variable. Consistent with expectations, the least turbid and most nutrient-rich (~6 $\mu\text{M NO}_3^-$) “near” station (station 25) had the highest Chl *a* concentration (>40 RFU from CTD data). However, at many stations close the glacier front (e.g. stations 22, 24, 27, and 30) the highest Chl *a* concentrations (30-40 RFU from CTD data) were measured below regions of high turbidity despite the impacts of the turbid plume limiting light (Figures 4d,f and 5d,f). Peaks in Chl *a* concentration at stations 22, 24, 26, 28 and 30 coincide with lower nutrient concentrations, while stations 25 and 27 have elevated Chl *a* and nutrient concentrations (Figures 4e,f and 5e,f).

4. Discussion

4.1. Plume dynamics in the near-shore environment

At Sverdrup Glacier, a shallow, warm-based tidewater glacier in the CAA, time-lapse camera imagery (Section 3.1) and *in situ* marine observations (Section 3.4) confirm the existence of a freshwater plume in the near-shore marine environment tied to the glacier melt season evolution. Unsurprisingly, the correlation between plume area and cumulative mass balance (Sup. Figure 3) suggests that plume surface area is tied to the total volume of melt coming from the glacier. It can be inferred that as the melt season progresses, a larger area of marine waters in Brae Bay are impacted by glacial melt. *In situ* water samples in this study were collected when the cumulative meltwater flux was near its peak (late summer), and cumulative mass balance was at its most negative. It can be assumed that this sampling reflects a time of year when meltwater extent in Brae Bay was likely near its maximum.

Results suggest that meltwater exits Sverdrup's terminus 10's of meters below the sea surface. A single on-ice transect from 2009 showed the grounding line within 0.6 km of the terminus to be 20 ± 10 m below the surface (Larsen, 2010). Subsequent erosion caused by continued subglacial drainage likely results in the plume now exiting at even greater depth (Anderson et al., 2006; Catania et al., 2018; Kessler et al., 2008). Additionally, the ice elevations off the centerline are more than 10 m lower than the measured transect. These lower ice surfaces are around a tunnel where subglacial melt was observed in 2019 to exit into Brae Bay (Sup. Figure 1b). This depression suggests that discharge is exiting Sverdrup Glacier at a depth of >30 m on the eastern side of the terminus where the main plume was observed in 2019 (Figure 1c).

The injection of the subsurface meltwater plume has important implications for water column structure in the ocean near the glacier front (Section 3.4). Near-surface (≤ 30 m depth) isopycnals within 4 km of the terminus slope upwards away from the terminus, mapping the rise of the buoyant plume to the surface between stations 22 (<1 km distance) and 31 (~ 13 km distance). The plume's influence appears to extend down to the $1026 \text{ kg m}^{-3} \sigma_\theta$ isopycnal, and extrapolation of this isopycnal's slope to the terminus indicates that the plume is originating from below 40 m depth, consistent with Sverdrup's estimated grounding line depth at this point along the terminus (≥ 30 m depth). This location also corresponds to the location of the main plume discharge that was observed in 2019. Further offshore between 13 and 26 km from the terminus (at stations 31, 32 and 33), isopycnals slope upwards towards the shore, characteristic of fjord-estuarine circulation. Here, the upward-sloping isopycnals begin outside the ring of moraines that hem in Brae Bay, therefore it is unlikely that this upwelling is driven directly by submarine glacial discharge solely from Sverdrup Glacier. Rather, this distal upwelling could be driven by variations in bathymetry (data not collected) between 4-10 km from the terminus (Timmermans & Marshall, 2020) or wind-driven Ekman transport in Jones Sound (Dmitrenko et al., 2016; Woodgate et al., 2005). Regardless of the forcing, this upwelling of deeper waters originating from Jones Sound has important implications for nutrient transport (Section 4.5).

4.2. Nutrient and carbon export in glacial meltwater runoff to the surface ocean

It has been proposed that glacial meltwater can be a primary mechanism for the delivery of macronutrients to the ocean (Hawkings et al., 2016; Hawkings et al., 2017; Tranter et al.,

2002). Our results suggest however that at Sverdrup Glacier in summer the concentrations of macronutrients in glacial meltwater runoff were not high enough to significantly augment marine concentrations. Specifically, phosphate and silicate concentrations in glacial runoff were lower than in marine water samples (Table 1). While average summer glacial and upper (>40 m depth) marine water column nitrate concentrations were not significantly different (Table 1), the volume of freshwater exported from Sverdrup Glacier over the melt season (0.34 Gt; Section 3.1) is small compared to the reservoir of receiving seawater. RACMO2.3 model results suggest that on average 5.8×10^9 L of glacial melt are delivered to the ocean each day over the 55 day summer melt season (Figure 2). Given the average nitrate concentrations in summer meltwater of $1.8 \pm 0.0 \mu\text{M}$ (Table 1), this implies an average daily nitrate delivery rate of $(1.1 \pm 0.2) \times 10^4$ mol per day. Accounting for summer plume extent, this delivery rate is estimated to impact a minimum volume of $\sim 0.01 \text{ km}^3$ of ocean water. Thus, even under the assumption of no biological uptake of nitrate, the glacial melt delivery rate is an order of magnitude too small to account for the observed $0.6\text{--}1.9 \mu\text{M}$ nitrate found in the upper 10 m of the marine water column at stations affected by the plume (Figures 4 and 5). Collectively, these results suggest that glacial melt from Sverdrup Glacier does not appreciably augment existing macronutrient concentrations in the coastal ocean distal to the ice front.

This conclusion agrees with recent studies that found direct addition via glacial meltwaters to not be a primary mechanism for delivery of macronutrients to the ocean (Cape et al., 2018; Kanna et al., 2018; Meire et al., 2017). However, debate remains, and seasonality and hydrology appear to play important roles in carbon and nutrient availability (Beaton et al., 2017; Hawkings et al., 2017; Hopwood et al., 2020). In the case of NO_3^- , a large fraction of glacially-sourced NO_3^- is derived from atmospheric deposition on the surface snowpack (Wolff, 2013), and because snow is the first to melt in summer, most of this NO_3^- is exported early in the season (Wadham et al., 2016). For example, Wadham et al. (2016) found significant concentrations of NO_3^- ($>4 \mu\text{M}$) in runoff rivers draining Leverett Glacier in samples collected before June, but by late July, NO_3^- concentrations were comparable to average concentrations observed on Sverdrup Glacier ($\sim 2 \mu\text{M}$). The low NO_3^- concentrations in the summer glacier meltwater found in this study are likely influenced by the time of sampling, i.e. at the peak of melt, when ice melt rather than snow melt dominates glacial runoff. However, should the seasonal variation of NO_3^- concentrations in meltwater from Sverdrup Glacier be of a similar magnitude to that of Leverett

Glacier (~2-fold difference in nutrient concentrations between early and late melt), we note that seasonal variation is still insufficient for direct NO_3^- delivery rates to account for the observed NO_3^- enrichment in the surface waters in Brae Bay. Further, we note that spring vs. summer glacial water samples from Sverdrup Glacier do not show a large difference in NO_3^- concentrations (Table 1).

In contrast, other studies of glaciers in Greenland have found glacial meltwater to be a significant source of crustal elements, including silica (Hawkings et al., 2017; Meire et al., 2016; Tranter et al., 2002) and phosphate (Hawkings et al., 2016), during peak meltwater flow. In the context of these studies, our findings of low silicate and phosphate concentrations in summer meltwater at Sverdrup Glacier are anomalous. The elevated concentrations of crustal elements seen in the Greenland glacier studies are likely the result of bedrock geology, a prolonged melt season and/or longer subglacial hydrological flow-paths, the latter two of which can result in extensive water–rock interaction and enhanced physical and biologically-mediated weathering (Aciego et al., 2015; Ravier & Buoncristiani, 2018). The Canadian Shield underlies both eastern Devon Island and Greenland, so it is unlikely that fundamentally different bedrock geologies are the cause of the variation in these macronutrient concentrations between Sverdrup Glacier and the glaciers studied in Greenland. Instead, it is more likely that glacier hydrology and meltwater routing played a role in generating the low meltwater nutrient concentrations observed in this study (Brown, 2002). Similar to previous work (Hawkings et al., 2017; Meire et al., 2017), meltwater samples here were collected in late summer, when basal hydrology is characterized by fast efficient export and short rock-water interactions, limiting enrichment of crustal elements in the meltwater. Phosphate and silicate concentrations in frozen spring samples were significantly higher than in the summer (Table 1), and thus, these lower crustal nutrient concentrations in summer melt may be evidence of low contact times. Further, on Sverdrup Glacier, most glacial melt is routed marginally until just prior to the terminus. This marginal routing likely denotes significantly shorter rock-water interactions with the glacier bed, explaining the lower summer PO_4^{3-} , SiO_4^{4-} , and carbon concentrations observed (Bennett, 2011). Finally, Sverdrup Glacier's slow ice velocities may result in less basal erosion and a subsequent lack of crustal elements in meltwater. Indeed, Milner et al. (2017) proposed that as glaciers and ice caps shrink, the quantity of soluble reactive phosphorus exported in runoff decreases.

Similar to major nutrient concentrations, the concentration of glacial DOC was not high enough for glacier meltwater inputs to significantly augment marine concentrations (Table 1). However, as discussed in Section 3.3, glacier meltwater differed significantly from marine waters with respect to the types of carbon present, with potentially important implications for the bioavailability of DOM to support marine ecosystems. Specifically, meltwater runoff from Sverdrup Glacier had a higher proportion of protein-like DOM compared to the more humic-like marine DOM and based on PARAFAC and PCA analyses, protein-like DOM components (P1-P3) were most associated with glacial samples. Tyrosine (P1) and tryptophan (P2) components were identified by Yamashita et al. (2015) to be indicators of the bioavailability of DOM in marine waters; this suggests that glacial samples from Sverdrup Glacier have a higher proportion of bioavailable protein-like DOM compared to marine water samples. The P3 component is related to the production of DOM via biological degradation; thus, the association between spring glacial samples and P3 we find could be an indicator that protein-like DOM is a result of microbially-mediated processes occurring in the basal and marginal environments (Smith et al., 2018). These three protein-like components have been previously found in DOM collected from Devon Island (Dubnick et al., 2017) and northern Alaska (Walker et al., 2009), as well as in riverine, and to a lesser extent estuarine, waters draining the Juneau Ice Field (Fellman et al., 2010b). As found in numerous other glacier studies, protein-like DOM in supraglacial and basal samples (>90% protein-like) is likely the result of productive microbial communities living on and under the ice that are able to generate and recycle bioavailable DOM for downstream export and consumption (Bhatia et al., 2010; Hood et al., 2009; Smith et al., 2018). The elevated ammonium concentrations observed here may also indicate microbial degradation of glacial DOM (Kumar et al., 2016). A recent study by Dubnick et al. (2020) corroborates this, having found abundant and distinct microbial communities in surface and basal ice at Sverdrup Glacier. The humic-like component H1 has been found in both marine and terrestrial studies (Coble, 2007; De Souza Sierra et al., 1994; Stedmon et al., 2003) and has been previously observed in basal ice from numerous glaciers on Devon Island (Dubnick et al., 2017). The marine humic-like component H2 has also previously been found in basal ice from Devon Island (Dubnick et al., 2017) and Alaska marine DOM (Walker et al., 2009). Broadly, the protein-like glacier DOM found in meltwater runoff draining Sverdrup Glacier and the humic-like marine DOM found in the surrounding coastal ocean is consistent with previous findings, indicating that glaciers are

microbially-based ecosystems capable of supplying comparatively labile DOM to downstream environments (Dubnick et al., 2020; Hood et al., 2009). In the ocean, this labile DOM in glacial melt can promote secondary productivity, with bacteria and microzooplankton using it as a carbon source. These organisms then go on to feed higher trophic levels in the marine food web (Pomeroy, 1974). Recent work in the McMurdo Dry Valleys (Antarctica) has found that heterotrophic production relies on labile DOM freshly-derived from photosynthetic bacteria rather than legacy organic carbon (Smith et al., 2017). While delineating the source of protein-like DOM in the ocean or its relative importance to CAA heterotrophs is beyond the scope of this study, if marine microbes preferentially use labile glacially-derived protein-like carbon over humic-like marine carbon, as has been found in previous studies in Alaska and Colorado (Arimitsu et al., 2018; Fegelman et al., 2019; Fellman et al., 2015), tidewater glaciers like Sverdrup Glacier, which export labile DOM to the ocean, may play an important role in stimulating local secondary production in Arctic waters distal to the ice terminus during the summer months.

4.3. Impact of the submarine discharge plume on the surface ocean

While carbon and nutrient concentrations in glacial melt were not high enough to directly impact the marine environment, signatures of buoyant plume rising close to the terminus (within 4 km) and the upwelling of deeper marine waters consistent with an estuarine-like circulation farther out in Jones Sound (~13-23 km from the terminus) were both detected (Sections 3.4 and 3.5). In marine water unaffected by external nutrient sources, AOU will have a positive linear relationship with nutrient concentration because oxygen consumption and nutrient additions have a shared source: organic matter remineralization. This linear relationship is observed in Brae Bay (Figure 7), further confirming that glacial melt is likely not the important source of the enriched macronutrient concentrations observed in marine waters surrounding Sverdrup Glacier.

Previous studies of glacier-induced upwelling focus primarily on the delivery of nitrate from depth, as NO_3^- is generally the limiting nutrient in the North Atlantic and Arctic oceans in the summer (Randelhoff et al., 2020). Nutrient ratios in Brae Bay suggest that surface phytoplankton are nitrogen limited at this time of year (Sup. Figure 5), and though upwelling at Sverdrup Glacier is shallow, it occurs below the nutricline (≥ 30 m depth) and is therefore sufficient to deliver waters with elevated nutrient concentrations ($\sim 5 \mu\text{M}$) to the surface. Recent

studies of four tidewater glaciers (Kronebreen, Kongsvegen, Conwaybreen, and Kongsbreen) in Kongsfjorden (Svalbard) all with relatively shallow (≤ 70 m depth) grounding lines found similar upwelled NO_3^- concentrations ($4.2 \mu\text{M}$) (Halbach et al., 2019). In comparison, deep tidewater glaciers in Greenland have been shown to be capable of entraining marine water with nearly double the NO_3^- concentration ($\sim 10 \mu\text{M}$) that is observed here (Kanna et al., 2018; Meire et al., 2017). However, given that NO_3^- is limiting at this time of year following the spring bloom, the delivery of waters with even modest concentrations of NO_3^- to the euphotic zone may promote productivity. The analysis of glacial melt fraction (Section 3.4) indicated that the rising meltwater plume is $\sim 13\%$ glacial melt (87% marine water), and RACMO2.3 modeling (Section 3.1) predicted that over the melt season Sverdrup exports a total of 0.34 Gt of meltwater to Brae Bay. These estimates and measured NO_3^- concentrations thus imply that 2.0 Gt of deeper marine water and $>10^{15}$ mol of NO_3^- may be delivered to surface waters during the summer – compared to the <0.5 Gt of NO_3^- delivered in spring and summer glacial melt. If this delivery is typical of the over 300 tidewater glaciers in the CAA, this implies that tidewater glaciers in this region may be responsible for delivering >3 Gt of NO_3^- to the surface ocean annually. It should be noted that the differences in underlying geology of CAA glaciers likely makes this estimation highly uncertain. Further, while most tidewater glaciers in the CAA have shallow discharge plumes relative to glaciers in Greenland, Sverdrup Glacier is an example of a very shallow tidewater glacier, even for the CAA (Cook et al., 2019), and thus, this estimate may be an underestimation. Regardless, this value represents nearly 2x more nitrate than is exported by the Mackenzie River in a year (Holmes et al., 2011). Note, however, that riverine input represents a source of ‘new’ nitrogen to the marine environment while glacially-derived upwelling redistributes marine nitrogen. Both are important for supporting productivity, but only ‘new’ nitrogen can alter the total marine nitrogen budget.

4.4. Glacier effects on primary productivity in front of a shallow tidewater glacier

Past studies of glaciers in Greenland and Svalbard have observed elevated surface concentrations of Chl *a* associated with regions of glacially-driven upwelling of nutrient-rich marine waters (Halbach et al., 2019; Kanna et al., 2018; Meire et al., 2017). Here, peaks in Chl *a* concentrations are primarily found within (stations 22, 30) or at the edges (station 24 and 25) of

the turbid meltwater plume in Brae Bay (Figures 4 and 5). The presence of high Chl *a* concentrations in areas of low nutrient concentrations suggests the biological uptake of macronutrients. Higher Chl *a* concentrations at all three “near” stations compared to the stations on the “distal” and “out” transects suggest that the strongest biological response to the buoyant meltwater plume upwelling occurs within 1 km of the terminus, where entrained nutrient-rich marine water is delivered to the surface. We also observe elevated Chl *a* concentrations ~13 km from the terminus (station 31) in an area of upwelling of deeper marine waters outside of the moraines hemming Brae Bay. It is unlikely that this estuarine-like upwelling >10 km from Sverdrup’s terminus is wholly dependent on subglacial discharge exiting at ≥ 30 m depth from the terminus of Sverdrup Glacier, but freshwater delivery along the coast may play an important role in driving estuarine-like circulation. Regardless, the distal upwelling does appear to promote the delivery of nutrient-rich water to the surface farther out in Jones Sound, sustaining elevated Chl *a* concentrations compared to surface waters >20 km from Sverdrup.

The Chl *a* responses seen in the Sverdrup Glacier system differ from those reported in studies on larger Greenland glaciers in important respects; specifically, the response is less extreme and spatial extent more limited at Sverdrup Glacier. Maximum Chl *a* concentration at Sverdrup was ~7.5 $\mu\text{g/L}$ (extracted concentration), while concentrations in previous studies in Greenland can exceed 20 $\mu\text{g/L}$ (Meire et al., 2017). These relatively small Chl *a* enhancements appear consistent with shallower tidewater glacier systems. Recent work at shallow tidewater glaciers in Svalbard report maximum Chl *a* concentrations of ~2.8 $\mu\text{g/L}$ (Halbach et al., 2019) during late July and early August. The different Chl *a* concentrations observed between these studies does not directly follow differences in glacier grounding line and submarine discharge depths, as current models would predict (Hopwood et al., 2018; Oliver et al., 2020). That being said, model values are not directly comparable to single point in time Chl *a* measurements, so more work and samples are necessary to fully evaluate how measured Chl *a* compare to modeled productivity estimates. Meire et al. (2017), observed high (~20 $\mu\text{g/L}$) Chl *a* concentrations at glaciers with deeper grounding lines (≥ 140 m depth) than Sverdrup, lower turbidities (<15 NTU), and a deeper euphotic zone than Sverdrup Glacier. However, the proximity of the closest Chl *a* measurement in that study was almost 10 km away from the glacier terminus, making direct comparisons to this work difficult. At Bowdoin glacier (in Greenland), however, Kanna et al., (2018) collected samples within 1 km of the terminus, finding similar proportions of glacial

melt in the plume water at that site (14%) as found here (13%). There, the highest observed Chl *a* (~6.5 µg/L) are similar to the maximum concentrations observed in this study (~7.5 µg/L). This is surprising, considering that Sverdrup has an estimated grounding line of >30 m depth compared to >200 m depth at Bowdoin Glacier. However, Kanna et al. (2018) did find elevated Chl *a* concentrations nearly 20 km into Bowdoin Fiord, while we see an elevated Chl *a* response extending a maximum of ~13.3 km from Sverdrup's terminus. The confined walls of Bowdoin fjord, different meltwater fluxes, and deeper grounding line may induce a larger degree of circulation, promoting similar levels of productivity farther away from the glacier terminus in the case of Bowdoin Glacier relative to Sverdrup Glacier. In Kongsfjorden (Svalbard), at both Kronebreen and Kongsvegen glaciers (discharge ~70 m depth) and Conwaybreen and Kongsbreen glaciers (discharge <10 m depth), Chl *a* concentrations were universally low (Halbach et al., 2019). In these cases, the marine waters around the deeper (~70 m depth) glaciers had lower Chl *a* concentrations (0.2–1.9 µg/L) likely due to higher turbidity, with differences in particle size and type (carbonate vs. silicates) between the sites playing an important role in light limitation. Thus, the “productivity continuum” between land terminating and tidewater glaciers, as defined by grounding line depth, does not appear to entirely hold for shallow tidewater glacier systems. Indeed, productivity at Sverdrup Glacier may be similar to or higher than productivity at other glaciers with deeper grounding lines (Halbach et al., 2019; Kanna et al., 2018). However, more study is clearly necessary to understand the full range of controls on entrainment, upwelling, nutrient delivery, and productivity at shallow tidewater glaciers.

5. Conclusion

Historically, tidewater glaciers have been identified as areas of heightened productivity (Lydersen et al., 2014; Vibe, 1939). Recently, glacially-induced upwelling of nutrient-rich deep water has been proposed as a mechanism that can support primary productivity at the termini of tidewater glaciers in Alaska, Greenland, Svalbard, and Antarctica (Arimitsu et al., 2016; Lydersen et al., 2014; Meire et al., 2017). No study has been conducted on this topic in the CAA in almost 50 years (Apollonio, 1973). Here we find that carbon and nutrient concentrations in glacial melt are too low to enrich surface marine concentrations in the coastal ocean. However,

similar to other studies, glacially-derived organic carbon exported within submarine discharge appears to be more bioavailable than marine carbon in the receiving seawater. We also observe that as the submarine discharge plume rises at the terminal ice cliff, it impacts the hydrography of the surrounding water column, inducing upwelling of intermediate (>30 m depth) marine water with elevated nutrient concentrations. The heightened Chl *a* concentrations observed at the interface between turbid freshened water and upwelled marine water close to the glacier terminus suggests that tidewater glaciers with shallow submarine outlets can promote primary productivity during nutrient-limited times of year.

Based on nutrient concentrations and Chl *a* response, Sverdrup Glacier falls between deep tidewater and land terminating glaciers, while it lies near the shallow end of the spectrum of grounding line depths (Hopwood et al., 2018). Compared to many glaciers examined in previous studies, Sverdrup Glacier is less dynamic, with a smaller meltwater flux and a shallower depth of submarine discharge. However, within 4 km of its terminus, the marine waters distal to Sverdrup Glacier may be as productive as tidewater glaciers in Svalbard and Greenland with deeper grounding lines (Halbach et al., 2019; Kanna et al., 2018). The differences between deep and shallow tidewater glaciers in the magnitude and variability of observed nutrient and Chl *a* concentrations speak to the importance of determining the impacts of runoff on a variety of proglacial aquatic environments. Further, simultaneous measurements of carbon and macronutrients in both on-ice and marine environments allowed us to detect glacially-induced entrainment of deep water and estuarine upwelling in Brae Bay, while confirming that glacial concentrations were too low to augment downstream nutrient and carbon pools. With continued retreat of large tidewater glaciers in Arctic seas, future work on how shallow tidewater glaciers affect downstream marine ecosystems will only become more relevant to the region as a whole.

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