

The Physics of the Carbon Cycle: About the Origin of CO₂ in the Atmosphere

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IPCC claims that Global Warming is caused more or less exclusively by anthropogenic emissions of CO₂. Therefore, only a total stop of these emissions could avoid a disaster. In IPCC's reasoning, two concepts are central: The "constant airborne fraction", according to which about 50 % of all anthropogenic emissions remain in the atmosphere, whatever the emissions and whatever the concentration, and the "fixed carbon budget", which is the maximum amount humans may emit, when Global Warming should stay below a given limit, independent of the temporal distribution of the emissions. In this article, it is shown that three prerequisites must be fulfilled for these two central concepts to be viable: All natural sources of CO₂ must have remained constant, the short-term partners of the atmosphere must store the same amount of CO₂ as the atmosphere itself, and the atmosphere together with its short-term partners must be a closed system. And it is shown with high confidence that, according to the rules of physics, all three are not fulfilled. If this were confirmed, it would have a serious impact on the entire climate debate. A careful review seems to be needed urgently.

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Key Words: Global warming, carbon cycle, carbon budget, airborne fraction, CO₂ concentration, anthropogenic emissions.

Key Points:

- According to the rules of physics a "constant airborne fraction" and a "fixed carbon budget" cannot exist in an open system.
- Increased natural emissions are the main driver of the enhanced atmospheric CO₂-concentration.
- Climate is either driven by naturally released CO₂, or other factors dominate. Reducing anthropogenic emissions is no more necessary.

Plain Language Summary

Global Warming is generally seen as one of our greatest challenges. Most scientists regard CO₂ as the main cause of warming, and our emissions as the single source of its growth in concentration. Consequently, these emissions must be terminated completely to avoid disaster. This view is widely accepted, notwithstanding some heated scientific debates, especially about the actual strength of CO₂'s impact on climate ("climate sensitivity"). But however that may be, there seems to be another weak point in the above argumentation, which is discussed hardly ever: Can an increase in CO₂ emissions by 5 % really increase the concentration by 50 % in the open system "atmosphere"? That is thoroughly scrutinized in this article. With the result: It is very unlikely! If confirmed, there are only two possibilities left: Either climate is determined by CO₂, then it is determined by naturally emitted CO₂, or other factors dominate, then CO₂ only plays a subordinate role, whatever its climate sensitivity. In both cases, nature is stronger than man regarding CO₂ too, and we do not need to reduce emissions, we can continue to benefit from cheap fossil energy and from improved agricultural production, driven by CO₂. A careful clarification is needed urgently.

1. Introduction

The current global warming is usually assessed to be the result of the anthropogenic emissions of CO₂. However, it is hard to understand why all the enormous decades-long research effort has not led to a sufficiently accurate and undisputed clarification of the correct value of the "climate sensitivity of CO₂" (that is the warming in case of doubling the concentration). Even in its latest report, AR6, IPCC gives the band width of uncertainty with 2.5 to 4 °C, almost a factor of 2 (IPCC, [2021](#))! Others claim higher or lower numbers, even way off. An end of the dispute is not in sight. This impossibility to adequately narrow the uncertainty could perhaps be due to scientists focusing too much on the interactions of CO₂

48 and climate, without sufficiently validating the starting point of their considerations, the assumption that
49 the increase in CO₂ concentration is manmade. It might even be impossible to achieve unambiguous
50 results when the base is erroneous. This article tries to scrutinize the reliability of the base.

51 **2. The element carbon and its cycle**

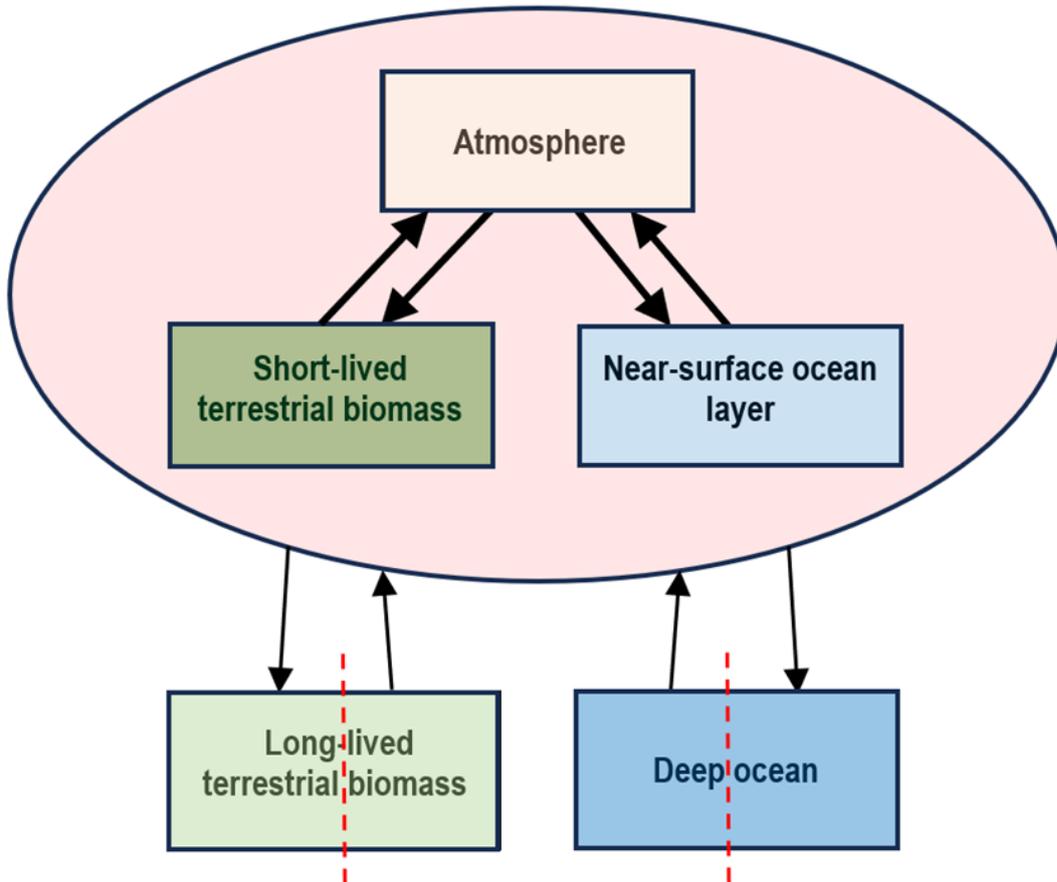
52 In its organic form, carbon is the basis of all life: *Without carbon, no life!* And in its chemical form CO₂, it
53 again is the basis of all life: *Without CO₂, no photosynthesis*, no plants, no animals, no humans.
54 Concentration doesn't even have to be zero, whenever it falls below about 150 ppm, photosynthesis would
55 stop and all life, as we know it, would be terminated.

56 Initially, CO₂ was the main component of the atmosphere. But meanwhile, most of it has been transferred
57 to rocks, reducing it to a trace gas with only 0.04 % concentration. However, this was not a
58 straightforward process, rather large amounts of rocks with all their carbon content have been subdued
59 into the earth mantle by plate tectonic processes, and a part of that carbon has been reemitted into the
60 atmosphere as CO₂ by volcanoes, and then these processes repeated. This leads to the concept of a
61 “*carbon cycle*”.

62 Usually, scientists distinguish between the “*geological*” (or “slow”, or “long-term”) and the “*biological*”
63 (or “fast”, or “short-term”) carbon cycle (e. g. Harrison, [2024](#)). The “*geological*” cycle includes processes
64 such as sedimentation, weathering of rocks, plate tectonics, etc., running on time scales of millions of
65 years or longer. Therefore, this cycle is irrelevant regarding manmade climate changes. In contrast, the
66 “*biological*” cycle comprises all exchange processes between atmosphere and biosphere, respectively
67 ocean. These processes are generally performed on short time scales, from days to several thousand years.
68 Therefore, this cycle does play a role in climate discussions.

69 However, for a better understanding, it seems appropriate to subdivide the “*biological*” cycle even further,
70 depending on the speed of the processes (Roth, [2019](#), [2020](#), [2022](#), [2023](#)). In this article, the term “*small*
71 *cycle*” is used for all fast-running processes (high exchange rates, time scales up to several decades), and
72 the term “*large cycle*” is used for all slower-running processes (lower exchange rates, longer time scales).
73 The “*small cycle*” comprises all exchange processes between the atmosphere and the *near-surface ocean*
74 *layer*, respectively the *short-lived terrestrial biomass*, which all run at high exchange rates. The near-
75 surface ocean layer is roughly about 50 to 100 m thick, it is well mixed by wind and waves, it includes
76 organic material in different forms, it is sunlit (photosynthesis!), and it exchanges carbon with the
77 atmosphere on the one side and with the deep ocean on the other side. To the short-lived terrestrial
78 biomass belong annual plants, leaves, needles, and the like. It takes out CO₂ from the atmosphere by
79 photosynthesis and gives it back by respiration and rotting.

80 The “*large cycle*” then comprises all slower exchange processes of the atmosphere, respectively the
81 “*small cycle*”, with the *deep ocean* and with *long-lived terrestrial biomass* (long-lived woods, humus,
82 peat, etc., including permafrost). “*Small cycle*” and “*large cycle*” together form the “*biological cycle*”
83 (Fig. 1).



84
 85 *Fig. 1: Carbon Cycle, schematic. The "small cycle" occurs inside the ellipse, the "large cycle" is all*
 86 *together. Arrows symbolize CO₂-exchanges, dashed red lines indicate decoupling of inflow and backflow*
 87 *in that reservoir.*

88 3. What happens when CO₂ is emitted?

89 The events following an emission of CO₂ into the atmosphere can be described to proceed in 3 steps (Fig.
 90 1):

- 91 • First, distribution of that CO₂ within the atmosphere: This is performed very effectively by wind
 92 and weather, and the same concentration is reached everywhere within a few months (not
 93 precisely the same concentration, local and short-term variations exist, but can be neglected for
 94 the discussions here).
- 95 • Second, further distribution within the "small cycle": This is performed by high exchange rates
 96 (about a quarter of the atmospheric inventory per year! IPCC, [2007, 2013, 2018, 2021](#)), and an
 97 equilibrium is reached within a few years (same partial pressure all over the "small cycle").
- 98 • Third, transfer of carbon out of the "small cycle" into the deep ocean, respectively into long-lived
 99 terrestrial biomass: This is performed by considerably slower processes, and due to the large
 100 inventories (IPCC, [2007, 2013, 2018, 2021](#)), equilibrium will be reached only after about
 101 thousand years (equilibrium within the "large cycle").

102 These three steps will be discussed in more detail further below.

103 4. “Carbon Budget”, “Airborne Fraction”, and “Climate Sensitivity”

104 IPCC claims that anthropogenic CO₂ emissions are the only cause of global climate change (IPCC, [2007](#),
105 [2013](#), [2018](#), [2021](#)). To avoid a catastrophe, we must completely stop all our emissions. IPCC assumes that
106 there is a “maximum amount of cumulative net global anthropogenic CO₂ emissions that would result in
107 limiting global warming to a given level”, IPCC calls this the “*carbon budget*” (IPCC, [2021](#)). This
108 “carbon budget” we have almost used up, so time is pressing for countermeasures. According to its
109 definition, the “carbon budget” is independent from the temporal distribution of our emissions.

110 IPCC deduces the existence of this fixed “carbon budget” from *two assumptions*:

- 111 1. The rise in atmospheric CO₂-concentration is (exclusively!) the consequence of the anthropogenic
112 emissions, and that is because approximately 50 % of them remain in the atmosphere
113 (quantitatively, not necessarily the individual molecules emitted), independent of the height of
114 these emissions and independent of the concentration already reached; IPCC speaks of a constant
115 “*airborne fraction*” (IPCC, [2007](#), [2013](#), [2018](#), [2021](#)).
- 116 2. The concentration increases the temperature by a fixed amount for each doubling of the
117 concentration. This amount is labelled the “*climate sensitivity of CO₂*” (IPCC, [2007](#), [2013](#), [2018](#),
118 [2021](#)).

119 Focus of the considerations here is a careful check whether a “fixed carbon budget” and a “constant
120 airborne fraction” can exist in the real world. It will be shown that both concepts contradict physical rules
121 with high confidence. If this is confirmed, anthropogenic emissions *cannot be* the main cause of the rising
122 CO₂-concentration, and therefore also not of the rising temperature, whatever the “climate sensitivity” of
123 CO₂ is. Consequentially, this “climate sensitivity” is only of secondary importance here. It should only be
124 repeated that its exact value is scientifically highly controversial; and for those interested, a few helpful
125 publications are listed (Beemt, [2019](#); Curry, [2023](#); Koutsoyiannis, et al. [2023](#); Lewis, [2023](#); McKittrick &
126 Christiy, [2020](#); Scafetta, [2022](#), [2024](#); Spencer, [2024](#); Spencer & Cristy, [2024](#); Vahrenholt & Lüning, [2020](#)).

127 5. Prerequisites

128 Physics and logic require three prerequisites to be fulfilled, when the fixed “carbon budget” and the
129 constant “airborne fraction” should exist:

- 130 1. All other sources of CO₂ *must have remained constant*. Otherwise, they would contribute to the
131 growing concentration, perhaps they could even dominate it.
- 132 2. The short-term partners of the atmosphere must store the *same amount* of CO₂ as the atmosphere
133 itself. Otherwise, it would not be possible that always 50 % of the anthropogenic emissions
134 remain in the atmosphere.
- 135 3. The atmosphere together with its short-term partners (the “short cycle”) *must be a closed system*.
136 Otherwise, CO₂ would be taken out of this system, reducing the concentration, the more, the
137 higher the concentration.

138 It will be shown that all three prerequisites are not fulfilled with a high level of confidence.

139 6. Prerequisite 1: Constant natural sources

140 Inside the atmosphere, CO₂ reacts as an inert gas: There is no CO₂ produced and none vanishes. All CO₂-
141 molecules inside have been emitted from an external source, and all will leave again into a sink. This
142 outflow starts when the first molecules accumulate, it is *the stronger the higher the concentration is*, and
143 it lasts theoretically until the last molecule.

144 For clarification: This outflow is the *gross* outflow. It must be strictly distinguished from the *net* outflow,
145 which is the difference to the simultaneous gross inflow. The net flow between two reservoirs always
146 goes from higher to lower concentration, but this net flow is always the *superposition of two*
147 *countercurrent gross flows*, back and forth. In equilibrium, the net flow always is zero, but the two
148 countercurrent flows depend on the concentration in the respective emitting reservoir (exceptions:
149 emissions from human activities or volcanos depend on external effects!).

150 The dependency of the gross outflow on concentration follows inevitably from the main processes
151 involved, dissolution in water and photosynthesis: These are diffusion-processes, and diffusion depends
152 on concentration! This has an immediate consequence: Whenever the emissions into the atmosphere
153 remain constant, the concentration in it *adjusts itself to a fixed value*: that value, where *outflow equals*
154 *inflow*! Hereafter, the concentration remains constant, despite of ongoing emissions! This alone
155 *contradicts the existence of a fixed “carbon budget” and of a constant “airborne fraction” inevitably!*
156 That's rather all that needs to be said.

157 But two further consequences of the (gross) outflow rising with the concentration should be mentioned:

- 158 1. When the total emissions into the atmosphere rise by x %, then the concentration in it can rise *at*
159 *most by x % too* (in equilibrium, before even less)!
- 160 2. All sources contribute to the concentration according to their relative strength. *No one can*
161 *contribute disproportionately!*

162 These two consequences always exist whenever the (gross) outflow from a reservoir increases with
163 concentration, regardless of the exact shape of this dependency. But because the outflow from the
164 atmosphere runs mainly via diffusion (dissolution in ocean water and photosynthesis in plants!), this
165 shape *must be proportionality!* Well, not necessarily exact from zero concentration up to 100 %, but
166 within the range of interest here, from about 300 to about 400 or 500 ppm, proportionality applies at least
167 in good approximation! (Attention: this is the gross outflow, the net outflow is proportional to the
168 difference in concentration!).

169 Another important feature is that this gross outflow from the atmosphere is completely independent of
170 what happens subsequently to the molecules that have left the atmosphere (for example, whether they are
171 circulated back into the atmosphere or not), and it is also completely independent of how much molecules
172 are emitted into the atmosphere simultaneously (and from which source they are emitted)! Therefore, if
173 we know how much the concentration has changed (we can measure it!), we can calculate, how much *the*
174 *(gross) outflow must have changed* as a consequence. And if we know the gross outflow, we can calculate
175 *which inflow must have taken place* to let the concentration develop as it did, independent of the cause of
176 this inflow (Roth, [2019](#), [2020](#), [2022](#), [2023](#))!

177 Short supplement: As already said, the gross flowrate from the atmosphere into the ocean does not depend
178 on what happens to the molecules afterwards. This also holds for possible chemical transformations of
179 those CO₂ molecules in the ocean water. But the (gross) flowrate back into the atmosphere does depend
180 on those transformations. This is, because the transformation-products, carbonate and bicarbonate, do not
181 contribute to the CO₂ partial-pressure, which drives the flow back into the atmosphere. This is further
182 enhanced due to the solution equilibrium between these transformation-products and CO₂ depending on
183 concentration.

184 To summarize: Because the concentration in the atmosphere has risen by 50 %, the total gross outflow
185 from the atmosphere *must have risen by 50 % too*, at least approximately! That seems to be what physics
186 requires. And since the concentration has risen, the total inflow *must have risen even more!* That seems to
187 be, what the mass balance requires. In other words: The total inflow into the atmosphere *must have risen*
188 *by about 50 %!* The 5 % anthropogenic emissions are *far too small* in any case.

189 Interim result: This confirms the result already found: To explain the observations, other sources of CO₂
190 must have been enhanced substantially, prerequisite 1 *cannot be fulfilled!*

191 *Which sources* have been enhanced, is of secondary importance for the discussions here, but clearly it is
192 very interesting. One contribution inevitably comes as an *answer* to the higher concentration in the
193 atmosphere, whatever the cause: Because of the rather small size of the immediate reaction partners of the
194 atmosphere, the near-surface ocean layer and the short-lived terrestrial biomass (Fig. 1), the concentration
195 in them always *rises markedly* when they take up more CO₂. Consequently, they must deliver *more* CO₂
196 back to the atmosphere! Therefore, when the emissions into the atmosphere increase, the emissions back
197 into the atmosphere from these two reservoirs increase too, with only a short time lag. (Note: This is the
198 answer the *atmosphere* gets from its direct reaction partners, for the answer, the “*small cycle*” gets from
199 its partners, see section [8](#)).

200 This feedback to increased emissions into the atmosphere exists without doubt, but there must be other
201 enhancements of emissions too: At least the *higher temperature*, whatever the cause, must have increased
202 the emissions from ocean and biomass! Reasoning: Higher temperatures emit more CO₂ from the ocean
203 (temperature dependent solubility of gases in liquids!), and they also enhance the exchange rates between
204 atmosphere and biomass, the latter boosted even more by the growing of biomass (“global greening”, see
205 e. g. Chen et al., [2024](#); Scinexx, [2016](#); Zhu et al., [2016](#)). One can discuss the size of this temperature-
206 driven enhancement, but not its existence. Higher temperatures always increase emissions!

207 One more remark: Regarding the natural fluxes, IPCC always emphasizes that they have remained
208 constant. But in Fig. 5.12 of (IPCC, [2021](#)), numbers are given that clearly contradict this: The emissions
209 from land into atmosphere have risen from 111.1 to 136.7 PgC/y, and that from ocean into atmosphere
210 have risen from 54.6 to 77.6 PgC/y. The increase of together 48.6 PgC/y outweighs the anthropogenic
211 emissions of 11 PgC/y by more than a factor four! IPCC only does not discuss its own numbers.

212 Summarizing the fulfillment of prerequisite 1: Anthropogenic emissions are *much too small*, and
213 increased temperatures *must have enhanced* emissions too! Further enhancement might come e. g. from
214 relocations of ocean currents with different carbon content, or from volcanoes (on land or submarine), and
215 there are some more possibilities. The rules of physics *require* substantially enhanced emissions, but they
216 also *provide possibilities* for that to happen. There is no need to assume any unknown physical effect.

217 **7. Prerequisite 2: Equal storage capacity**

218 If the “small cycle” were a closed system with equal storage capacities in the atmosphere itself and in the
219 rest of the cycle, all CO₂ taken up would distribute itself with *half of it* remaining in the atmosphere. That
220 is exactly, what IPCC assumes regarding the anthropogenic emissions (IPCC, [2007](#), [2013](#), [2018](#), [2021](#))!
221 But that only works in a closed system with equally large storage capacities!

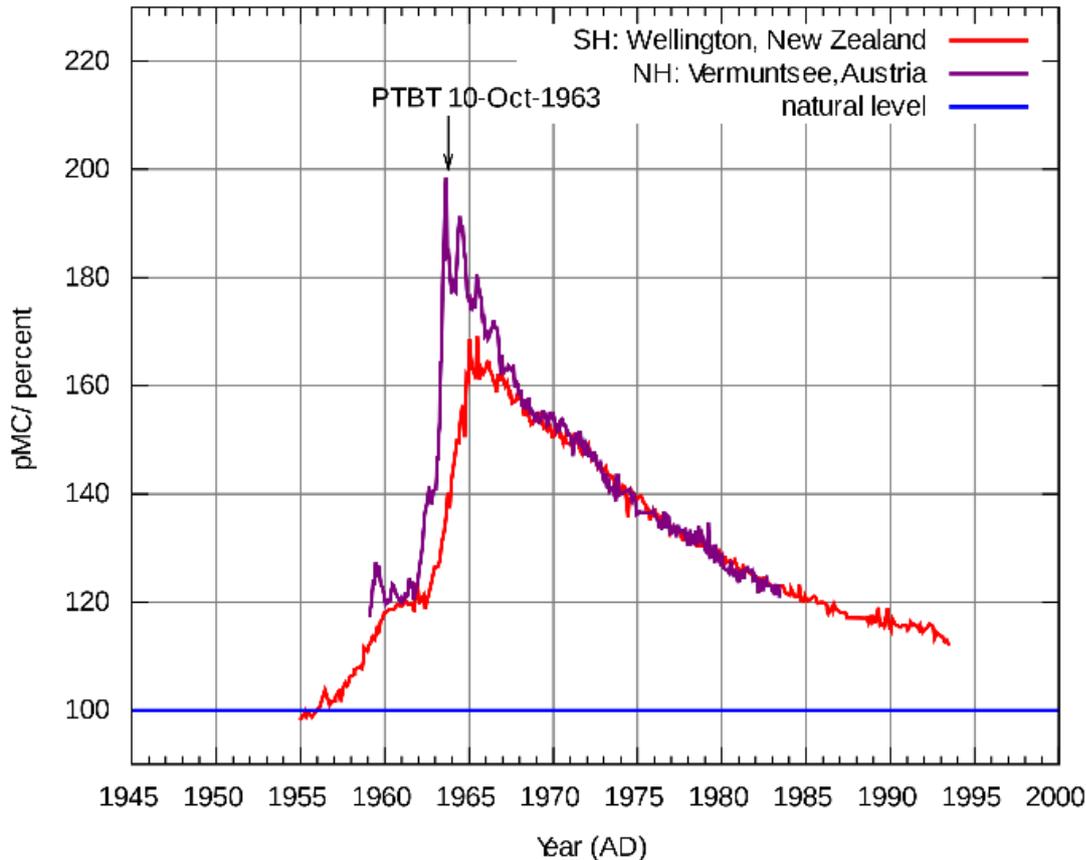
222 Fig. 5.12 in (IPCC, [2021](#)) gives numbers for carbon-inventories. Their interpretation is complex, because
223 most of the CO₂ dissolved in ocean is converted into carbonate and bicarbonate, which do not contribute
224 to the partial pressure. And regarding biomass, IPCC does not differentiate between short-lived and long-
225 lived biomass. But the numbers given strongly suggest that there is considerably less CO₂ in the
226 atmosphere than in the rest of the “small cycle”. If correct, *less than half* of the anthropogenic releases
227 remain in the atmosphere!

228 However, this “equal capacity” is only valid anyway if no other sources are enhanced. Otherwise, the
229 atmosphere must be larger to retain 50 %. Obviously, that is even *less likely*. Hence, the fulfillment of
230 prerequisite No. 2 is *seriously in doubt*.

231 But perhaps other observations can help: There is a radioactive variant of CO₂, ¹⁴CO₂ (8 neutrons in the
232 C-nucleus, T_{1/2} about 6000 years), with a very low natural atmospheric concentration. But following the
233 atomic bomb tests, its concentration almost doubled (Fig. 2). And after the test stop agreement in 1963,
234 this concentration decreased rapidly, essentially *down to its previous value* before the bomb tests. In this
235 case, definitely *less than 50 %* of the ¹⁴CO₂ released anthropogenically remained in the atmosphere!

236 And in this regard “normal” CO₂ *cannot behave differently!* It must distribute itself in the “small cycle”
237 according to the same pattern as ¹⁴CO₂, basically independent of the individual molecular weight. Even
238 with “normal” CO₂, only *significantly less than 50 %* can remain in the atmosphere! This confirms that
239 the observed increase of the concentration by 2.5 ppm/y most probably *is not* the result of half of the
240 anthropogenic emissions remaining in the atmosphere (as IPCC assumes), but rather the result of *much*
241 *stronger sources* combined with *substantially increased outflow* of CO₂ from the “small cycle” into the
242 deep ocean, respectively into long-lived biomass.

243 In summary, even if there is no real proof, there is strong evidence that prerequisite 2 *is not fulfilled!*



244 Fig. 2: $^{14}\text{CO}_2$ concentration in the atmosphere: Pulse and decay after the atomic bomb tests
 245 (Hakanomono, 2015).

246 8. Prerequisite 3: Closed system

247 In a closed system, all CO_2 emitted into it remains in it. Constant emissions enhance the concentration
 248 indefinitely. But in an open system, this is completely different: Because of the gross outflow rising with
 249 concentration, constant emissions enhance the concentration *only until outflow equals inflow*. Afterwards,
 250 the concentration *remains constant*, despite of ongoing emissions. Because the “small cycle” is an open
 251 system, prerequisite 3 *cannot be fulfilled*. Therefore, the fixed “carbon budget” and the constant “airborne
 252 fraction” *cannot exist!*

253 *Reasoning:* The only parameter really measured is CO_2 -concentration. And to interpret these
 254 measurements, a special feature of the carbon-exchange with the deep ocean is of particular importance:
 255 Forward flow and return flow *can diverge substantially!* The downward flow into the deep ocean
 256 increases with the atmospheric concentration (more accurate: with the concentration in the near surface
 257 ocean layer), because the two effective processes, the biological pump (sinking of dead organisms with
 258 calcareous shells) and the physical pump (sinking of entire water packages with their whole contents), run
 259 proportional to concentration, at least approximately. But due to the sheer size of the deep ocean, and due
 260 to the slow currents in it, the return flow back from the deep ocean remains basically *unchanged* for about
 261 1000 years! The answer of the deep ocean to an increasing atmospheric concentration simply takes that
 262 long. Therefore, when the concentration in the atmosphere changes, the exchange with the deep ocean is
 263 *imbalanced* for a significantly long period!

264 (IPCC, 2021) gives the downflow into the deep ocean with 275 PgC/y. Previously, it was about 100
 265 PgC/y (IPCC, 2007, 2013, 2018). Nothing shows clearer than this surprising suddenly jump that we are
 266 *far away from knowing everything for sure* in the carbon cycle! But whatever the real value, it doubtless
 267 proves that the “small cycle” *is an open system!*

268 Similar it is regarding the long-lived terrestrial biomass: Here, too, the storage time is large and therefore,
 269 the exchange between atmosphere and long-lived biomass is *imbalanced* for a longer period. IPCC only

270 gives combined numbers for the short-lived and the long-lived terrestrial biomass (IPCC, [2021](#)), but a
271 substantial part of the answer of the terrestrial biomass to rising atmospheric concentration is
272 considerably delayed unambiguously, contributing to the openness of the system.

273 For clarification: It is the *size* of the direct reaction partners, that determines the time delay of an answer
274 to increased emissions of CO₂: The atmosphere as such has small partners, resulting in a *quick* answer,
275 within maximal a few years, the partners of the “small cycle” in total are much larger, resulting in a delay
276 of about *1000 years*. During this time, the “small cycle” definitely *is an open system!*

277 **9. Dependence on temporal distribution**

278 As already said, the definitions of the fixed “carbon budget” and the constant “airborne fraction” require
279 *independence of the temporal distribution* of the anthropogenic emissions in both cases. For example,
280 emissions evenly distributed over 100 years must have the *same consequences* as an abrupt emission of
281 the same total amount in a single pulse. And the consequences must be the same, independent of the time
282 of this pulse, be it e. g. in year 1 or 100, or in any other year.

283 But if, for example, 50 % of that amount are emitted in year 1 and the other 50 % are emitted in year 100,
284 and if that should have the *same consequences* as the larger single pulse in year 100, *no CO₂* must be
285 taken out of the atmosphere for 100 years despite of substantially increased concentration! That
286 *contradicts* physics, which requires outflow to increase with concentration, and it also *contradicts* the
287 fixed “airborne fraction” of 50 %, which, if correct, would mean that 50 % of any amount emitted are
288 withdrawn within a few years! The two concepts, the “fixed carbon budget”, and the “constant airborne
289 fraction”, *contradict each other!*

290 Appraisal: Both assumptions, the fixed “carbon budget” and the fixed “airborne fraction”, are key
291 components in IPCC’s attribution of the rise in CO₂-concentration solely to human emissions (IPCC,
292 [2007](#), [2013](#), [2018](#), [2021](#)). Both seem to *contradict physics*, and they *cannot coexist*, because they are
293 mutually exclusive! Probably, both are incorrect. Most likely, the rise in CO₂-concentration is a *mood of*
294 *nature with only a small human contribution!*

295 **10. Driving force of 120 ppm**

296 Today, the concentration in the atmosphere is about 120 ppm higher than it was 150 years ago. And today
297 the net outflow from the atmosphere into the ocean and into the terrestrial biomass is about 2.5 ppm/y.
298 Sometimes, this is interpreted as the 120 ppm being the *driving force* to emit the 2.5 ppm/y from the
299 atmosphere. On that basis, it is calculated: If we freeze our emissions at today’s value, the concentration
300 increases only up to a new equilibrium at about 500 ppm, and if we reduce our emissions to 50 %, the
301 concentration remains constant immediately, and if we terminate our emissions, the concentration falls
302 rapidly, exponentially with a time constant of about 55 years down to the old equilibrium 150 years ago
303 (e. g. Vahrenholt & Lüning, [2020](#); Halperin, [2015](#); Spencer, [2019](#)).

304 But that seems to be *wrong for two reasons*:

- 305 • First, the driving force for the actual net outflow is the *actual* difference in concentration between
306 the atmosphere and its sinks, not the *mathematically calculated* difference between today’s
307 concentration and that 150 years ago. The atmosphere does not even have a memory for any past
308 concentration, it only knows today’s boundaries.
- 309 • And second, an imbalance with a driving force of 120 ppm appears to be *totally impossible* in a
310 system with an exchange rate of about a quarter of the inventory per year and the gross outflow
311 depending on concentration, at least for slow transients (and the real transients always have been
312 below 1 % of the inventory per year!). Such a high imbalance would be eliminated completely
313 within only a few years!

314 Today’s net flow of 2.5 ppm/y from the atmosphere into the near surface ocean layer and into the
315 terrestrial biomass is tantamount to the statement that these two reservoirs emit 2.5 ppm/y less than they
316 get. What they get, we know, is driven proportionally by the total concentration in the atmosphere, and
317 regarding their emissions, we do not know the driving force (temperatures, ocean currents, volcanoes,
318 etc.), but we can calculate the (gross) flux by obeying the mass balance. These simple physical

319 relationships show that a difference of 120 ppm does not exist in nature, and never can be the driving
320 force for the 2.5 ppm/y.

321 **11. What happens when we terminate our emissions?**

322 Presumably not very much. On the basis of the considerations here, the development of the concentration
323 has been determined by changes of natural emissions, and this will also hold for the future!

324 Anthropogenic emissions are much too low to be a real player. If natural emissions will continue to rise,
325 the concentration will rise too, and if nature reduces its emissions, the concentration will fall too,
326 whatever we do. The influence humankind can exert is so small that it hardly matters.

327 **12. Discussion**

328 The main result of the considerations here is that both, the fixed “carbon budget” and the constant
329 “airborne fraction”, *do not exist*, and as a consequence, the emissions from natural sources *must have*
330 *been increased substantially* to rise the concentration as it is observed. Against this, two objections are
331 raised in particular: The estimations made here for the size of the natural flows are *too unreliable*, and a
332 *sink cannot be a source*.

333 The first one is difficult to comprehend, because the estimations are made strictly on the basis of physical
334 rules, and they show that the natural flows have grown by a multiple of the anthropogenic emissions.
335 Therefore, the preponderance of natural emissions would remain valid, even if an important failure should
336 be found somewhere. This “remain valid” is supported by the fact that the two cycles, the one between
337 atmosphere and ocean, and the other between atmosphere and biomass, are totally independent of each
338 other (except for the atmosphere being part of both), and that they operate by essentially different
339 physical processes. Each of the two cycles is strong enough on its own to maintain the central statement
340 “nature dominates”, even if the argumentation presented here should break down in the other cycle. The
341 preponderance of nature would be less, but it would still apply, rejecting IPCC’s view “only the
342 anthropogenic emissions”. Therefore, there must be at least two independent failures to uphold IPCC’s
343 view. Of course, that is possible in principle, but it is even more unlikely. The central statement “nature is
344 stronger than humankind” seems to be robust.

345 The second objection seems to be self-evident on the first view: A sink cannot be a source! The numbers
346 are clear: At least for the last 60 or so years ocean and terrestrial biomass always *have been* a net sink.
347 They always have taken up CO₂ net, in an amount equal to about half of the anthropogenic emissions.
348 Therefore, it is often argued that the anthropogenic emissions *must be the only source* of the increased
349 concentration because ocean and terrestrial biomass always have been a *net sink!*

350 But that is a premature judgment. It simply ignores the fact that the (gross) outflow always *increases* with
351 rising concentration, whatever the cause. Let’s start with external emissions, that are emissions from the
352 outside of the “small cycle” into the atmosphere (e. g. anthropogenic emissions by burning fossil fuel or
353 by cement production, or emissions from volcanoes, etc.): In all such cases the concentration in the
354 atmosphere rises, and as a consequence the (gross) outflow from the atmosphere rises too, but it stays a
355 little bit behind the (gross) inflow. Consequently, ocean and terrestrial biomass act as a *net sink*, taking up
356 part of the enhanced emissions. That applies for all external emissions.

357 Not much different in case of internal emissions, that are emissions out of ocean or terrestrial biomass by
358 relocations between them and the atmosphere (caused for example by enhanced temperatures or by
359 changed ocean currents, but also anthropogenic emissions by land use changes fall into this category):
360 Here, too, the concentration in the atmosphere rises, and consequently the (gross) outflow from it rises
361 too, and again this outflow stays a little bit behind the inflow. But in this case, ocean and terrestrial
362 biomass lead the way, and they emit more than they take up. They now act as a *net source!* That applies
363 for all internal emissions.

364 Another important point is that in both cases the quantity as net sink, respectively as net source, is the
365 *momentary difference* between total input and total output to/from ocean and terrestrial biomass.

366 This is the separate consideration. But what is when both forms of enhanced emissions, external and
367 internal, occur *simultaneously*? Then both increase the concentration in the atmosphere, and by that both

368 increase the (gross) outflow from it. This doubly increased outflow now even can be larger than the
369 emissions from ocean and terrestrial biomass alone without crediting the anthropogenic emissions. In that
370 case, ocean and terrestrial biomass *switch to be a net sink* (with regard to the total emissions), in spite of
371 *still being a net source* for their own (under exclusion of the anthropogenic emissions)! This switch
372 always occurs, when the anthropogenic emissions are *strong enough* to enhance the concentration
373 enough, so that the (gross) outflow now overtakes the emissions from ocean and terrestrial biomass alone.
374 Then, ocean and terrestrial biomass act as *a sink and as a source at the same time!* That is no
375 contradiction, they are a sink with regard to all sources combined, and they are a source simultaneously
376 with regard to what they do alone, independent of the external emissions!

377 That fits well with the findings in (Ollila, [2016](#)) that ocean and biomass had been a *net source* of CO₂ for
378 200 years, with only very small anthropogenic emissions during that time, but they switched to be a *net*
379 *sink* around 1956, when the anthropogenic emissions became strong enough. For clarification: They
380 became strong enough to enable that switch, however, that does not tell *which one* of the two emissions
381 have contributed more to the observed rise in concentration. But this is an easy question: Since the
382 emissions from natural sources have increased about *tenfold* compared to the anthropogenic emissions
383 (about 50 % versus about 5 %), the answer is clear-cut (and there is even plenty of room left for
384 inaccuracies, see above)!

385 **13. Consequences and final remark**

386 Once again: The main result of the considerations here is that both, the fixed “carbon budget” and the
387 constant “airborne fraction”, *do not exist*, and that consequently the emissions from natural sources *must*
388 *have been enhanced substantially* to rise the concentration as it is observed! If confirmed, there are only
389 *two possibilities left*: Either the climate is determined by CO₂, then it is *natural* CO₂ that determines, or
390 other factors dominate, then CO₂ plays only a *subordinate role* at most! In both cases, there is *no need to*
391 *reduce anthropogenic emissions*, at least not for climate protections sake. We could benefit from cheap
392 energy from fossil fuels and from CO₂-improved plant growth without any remorse. And we would have
393 to accept the climate, as it develops on its own, and if necessary, we would have to implement mitigating
394 measures!

395 The results shown here clearly *contradict the mainstream view*. This is a good reason for caution, but the
396 results seem to be backed by physics and by logic, and counterarguments do not seem to be sustainable.
397 Therefore, a careful review is required urgently. This article aims to push the discussion.

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